Acid-Base Status Determines the Renal Expression of Ca\textsuperscript{2+} and Mg\textsuperscript{2+} Transport Proteins

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Chronic metabolic acidosis results in renal Ca\textsuperscript{2+} and Mg\textsuperscript{2+} wasting, whereas chronic metabolic alkalosis is known to exert the reverse effects. It was hypothesized, therefore, that these adaptations are mediated at least in part by the renal Ca\textsuperscript{2+} and Mg\textsuperscript{2+} transport proteins. The aim of this study, therefore, was to determine the effect of systemic acid-base status on renal expression of the epithelial Ca\textsuperscript{2+} channel TRPV5, the Ca\textsuperscript{2+}-binding protein calbindin-D\textsubscript{28K}, and the epithelial Mg\textsuperscript{2+} channel TRPM6 in relation to Ca\textsuperscript{2+} and Mg\textsuperscript{2+} excretion. Chronic metabolic acidosis that was induced by NH\textsubscript{4}Cl loading or administration of the carbonic anhydrase inhibitor acetazolamide for 6 d enhanced calciuresis accompanied by decreased renal TRPV5 and calbindin-D\textsubscript{28K} mRNA and protein abundance in wild-type mice. In contrast, metabolic acidosis did not affect Ca\textsuperscript{2+} excretion in TRPV5 knockout (TRPV5\textsuperscript{−/−}) mice, in which active Ca\textsuperscript{2+} reabsorption is effectively abolished. This demonstrates that downregulation of renal Ca\textsuperscript{2+} transport proteins is responsible for the hypercalciuria. Conversely, chronic metabolic alkalosis that was induced by NaHCO\textsubscript{3} administration for 6 d increased the expression of Ca\textsuperscript{2+} transport proteins accompanied by diminished urine Ca\textsuperscript{2+} excretion in wild-type mice. However, this Ca\textsuperscript{2+}-sparing action persisted in TRPV5\textsuperscript{−/−} mice, suggesting that additional mechanisms apart from upregulation of active Ca\textsuperscript{2+} transport contribute to the hypocalciuria. Furthermore, chronic metabolic acidosis decreased renal TRPM6 expression, increased Mg\textsuperscript{2+} excretion, and decreased serum Mg\textsuperscript{2+} concentration, whereas chronic metabolic alkalosis resulted in the exact opposite effects. In conclusion, these data suggest that regulation of Ca\textsuperscript{2+} and Mg\textsuperscript{2+} transport proteins contributes importantly to the effects of acid-base status on renal divalent handling.

nemias are associated with decreased renal expression of Ca$^{2+}$ and/or Mg$^{2+}$ transporters (15,16). Recently, TRPV5 knockout (TRPV5$^{-/-}$) mice were generated in our laboratory, which display a robust renal Ca$^{2+}$ leak localized to DCT/CNT, illustrating that active Ca$^{2+}$ reabsorption is effectively abolished (17). These mice constitute a unique mouse model to determine the role of TRPV5 and active Ca$^{2+}$ reabsorption in acid-base induced alterations of calcireum.

The aim of this study, therefore, was to determine the effect of chronic metabolic acidosis and alkalosis on the expression of Ca$^{2+}$ and Mg$^{2+}$ transporters in the kidney and to evaluate their contribution to the altered Ca$^{2+}$ and Mg$^{2+}$ excretion. We induced metabolic alkalosis by oral NaHCO$_3$ loading and metabolic acidosis by NH$_4$Cl loading, as well as applied acetazolamide administration in wild-type and TRPV5$^{-/-}$ mice. Acetazolamide specifically inhibits proximal tubular HCO$_3$ reabsorption, resulting in a self-limiting metabolic acidosis with, in contrast to NH$_4$Cl loading, an alkaline urine pH (18–20). This enabled evaluation of the role of luminal pH. Furthermore, whereas acidosis generally increases urine Mg$^{2+}$ excretion, acetazolamide is known for its unexplained Mg$^{2+}$-sparing action (21–23).

Materials and Methods

**Metabolic Acidosis and Alkalosis in Wild-Type and TRPV5$^{-/-}$ Mice**

**Metabolic Acidosis.** TRPV5$^{-/-}$ mice were recently generated by targeted ablation of the TRPV5 gene and genotyped as described previously (17). Ten-week-old wild-type (TRPV5$^{+/+}$) mice and TRPV5$^{-/-}$ littermates were kept in a light- and temperature-controlled room with ad libitum access to deionized drinking water. Mice were ration-fed standard pelleted chow (0.25% [wt/wt] NaCl, 1.1% [wt/wt] Ca, 0.2% [wt/wt] Mg) during the metabolic balance studies. For evaluation of the effects of metabolic acidosis, mice were randomly assigned to a group that received either 0.28 M (TRPV5$^{+/+}$) or 0.14 M (TRPV5$^{-/-}$) NH$_4$Cl via the drinking water during 6 d or a control group that received normal deionized drinking water. NH$_4$Cl loading is a generally accepted and validated method to induce metabolic acidosis in rodents (11–13,24). Because the oral fluid intake of TRPV5$^{-/-}$ mice is approximately two-fold higher than that in wild-type mice, the lower NH$_4$Cl concentration in their drinking water ensured a similar oral acid load compared with wild-type mice to prevent acid overloading in these TRPV5$^{-/-}$ mice. Alternatively, subcutaneous administration of the carbonic anhydrase inhibitor acetazolamide (20 mg/kg per d) during 6 d by osmotic minipumps was applied. This enabled evaluation of the role of urine pH in metabolic acidosis-induced changes in Ca$^{2+}$ excretion as well as the role of TRPM6 in the unexplained Mg$^{2+}$-sparing action of acetazolamide.

**Metabolic Alkalosis.** Metabolic alkalosis was induced by oral administration of 0.2 and 0.1 M NaHCO$_3$ to TRPV5$^{+/+}$ and TRPV5$^{-/-}$ mice, respectively. NaHCO$_3$ loading was previously shown to induce metabolic alkalosis in rodents (12,13). Mice that received 0.2 or 0.1 M NaCl constituted the control group, thereby correcting for possible effects of the increased Na$^+$ load. The latter is particularly important because passive Ca$^{2+}$ reabsorption is functionally coupled to Na$^+$ reabsorption. Mice were treated for 6 d, after which they were housed in metabolic cages to enable collection of 24-h urine samples under mineral oil, preventing evaporation. At the end of the experiment, the mice were killed, blood samples were taken, and kidneys were sampled. The animal ethics board of the Radboud University Nijmegen approved all animal studies.

**Analytical Procedures**

Serum and urine Ca$^{2+}$ and Mg$^{2+}$ concentrations were determined using colorimetric assays as described previously (16,25). Blood gas measurements were performed using a Hitachi auto-analyzer (Hitachi, Laval, Quebec, Canada). Na$^+$, K$^+$, and Li$^+$ concentrations were measured using flame-spectrophotometrically (Eppendorf FCM 6343, Hamburg, Germany). Urine pH was determined using an electronic ion analyzer (Hanna Instruments, Szeged, Hungary), and osmolality was measured with an Osmette A automatic osmometer (Precision Instruments, Sudbury, MA).

**Real-Time Quantitative PCR**

Total RNA was extracted from kidney using TriZol Total RNA Isolation Reagent (Life Technologies BRL, Breda, The Netherlands). The obtained RNA was subjected to DNase treatment and reverse transcribed using Moloney-Murine Leukemia Virus-Reverse Transcriptase (Life Technologies BRL) as described previously (26,27). Subsequently, the acquired cDNA was used to determine TRPV5, calbindin-D$_{28K}$, and TRPM6 mRNA levels in kidney by real-time quantitative PCR on an ABI Prism 7700 Sequence Detection System (PE Biosystems, Rotkreuz, Switzerland) as described previously (15,26). In addition, mRNA expression of the housekeeping gene hypoxanthine-guanine phosphoribosyl transferase was determined as an endogenous control, which enabled calculation of specific mRNA expression levels as a ratio of hypoxanthine-guanine phosphoribosyl transferase.

**Immunohistochemistry**

Staining of kidney sections for TRPV5, calbindin-D$_{28K}$, and TRPM6 was performed on cryosections of periodate-lysine-paraformaldehyde-fixed kidney samples as described previously (7,28). For semiquantitative determination of protein abundance, images were made using a Zeiss fluorescence microscope equipped with a digital camera (Nikon DXM1200), which were analyzed with the Image Pro Plus 4.1 image analysis software (Media Cybernetics, Silver Spring, MD). The entire cortex in two separate kidney sections of each animal was included in the analysis, resulting in quantification of protein levels as the mean of integrated optical density.

**Immunoblotting**

Calbindin-D$_{28K}$ protein levels were semiquantified by immunoblotting as described previously (27). In short, kidney cortex sections were homogenized and samples were normalized according to protein concentration. Subsequently, protein samples were separated on 16.5% (wt/vol) SDS-PAGE gels and blotted to polyvinylidene difluoride-nitrocellulose membranes (Immunobilon-P; Millipore Corp., Bedford, MA), and protein was detected using a rabbit calbindin-D$_{28K}$ antibody.

**Statistical Analyses**

Data are expressed as means ± SEM. Statistical comparisons were analyzed by one-way ANOVA and Fisher multiple comparison. P < 0.05 was considered statistically significant. All analyses were performed using the StatView Statistical Package software (Power PC version 4.51, Berkely, CA) on an Apple iMac computer.

**Results**

**Metabolic Acidosis and Alkalosis in Wild-Type and TRPV5$^{-/-}$ Mice**

Oral NH$_4$Cl loading induced a similar metabolic acidosis in wild-type and TRPV5$^{-/-}$ mice, as demonstrated by the signif-
after 3 d compared with day 6 of treatment (data not shown). Indeed, urine pH was more alkaline in all animals

Accordingly, NH₄Cl reduced urine pH in wild-type and TRPV5 mice. Likewise, chronic acetazolamide treatment significantly increased urine pH and decreased blood pH and HCO₃⁻ concentration in wild-type but not in TRPV5/mice. Of note, acetazolamide-induced acidosis and urinary alkalization is often self-limiting. Indeed, urine pH was more alkaline in all animals after 3 d compared with day 6 of treatment (data not shown). Furthermore, blood gas determination substantiated the effectiveness of the oral NH₄HCO₃ loading protocol in wild-type and TRPV5/mice. Serum pH was similar in both genotypes during NH₄HCO₃ loading, whereas serum HCO₃⁻ levels in TRPV5/mice were significantly higher. This reflects higher pCO₂ in the latter situation, most probably as a result of differences in time or depth of anesthesia. The absence of this difference in pCO₂ between wild-type and TRPV5/mice in the other treatment groups suggests that there is no intrinsic increased susceptibility to retain CO₂ in TRPV5/mice. It is interesting that urine pH was consistently 0.5 to 1 pH unit lower in TRPV5/mice as compared with the corresponding wild-type mice, but, importantly, TRPV5/mice did not display metabolic acidosis at baseline. Diuresis and Na⁺ excretion were not affected by NH₄Cl or NH₄HCO₃ loading, whereas

### Table 1. Acid-base status during different treatment protocols in TRPV5⁺/⁺ and TRPV5⁻/⁻ mice

<table>
<thead>
<tr>
<th>Treatment</th>
<th>pH</th>
<th>[HCO₃⁻] (mM)</th>
</tr>
</thead>
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<td>Controls</td>
<td>7.27 ± 0.01</td>
<td>22.3 ± 0.7</td>
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<td>NH₄Cl</td>
<td>7.11 ± 0.05ᵇ</td>
<td>17.2 ± 1.3ᵇ</td>
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<td>ACTZ</td>
<td>7.22 ± 0.02ᵇ</td>
<td>19.0 ± 0.9ᵇ</td>
</tr>
<tr>
<td>NaCl</td>
<td>7.27 ± 0.02</td>
<td>22.6 ± 0.8</td>
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<tr>
<td>NaHCO₃</td>
<td>7.35 ± 0.01ᵇ,ᶜ</td>
<td>25.8 ± 1.0ᵇ,ᶜ</td>
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<table>
<thead>
<tr>
<th>Treatment</th>
<th>pH</th>
<th>[HCO₃⁻] (mM)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Controls</td>
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<td>21.0 ± 0.5</td>
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<td>NH₄Cl</td>
<td>7.16 ± 0.01ᵇ</td>
<td>18.3 ± 0.7ᵇ</td>
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<tr>
<td>ACTZ</td>
<td>7.22 ± 0.01</td>
<td>23.1 ± 0.9</td>
</tr>
<tr>
<td>NaCl</td>
<td>7.25 ± 0.02</td>
<td>21.7 ± 0.5</td>
</tr>
<tr>
<td>NaHCO₃</td>
<td>7.35 ± 0.01ᵇ,ᶜ</td>
<td>36.3 ± 3.8ᵇ,ᶜ</td>
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*Controls, animals that received deionized drinking water only; NH₄Cl, animals that received 0.28 (TRPV5⁺/⁺) or 0.14 M (TRPV5⁻/⁻) NH₄Cl via the drinking water; ACTZ, animals that received acetazolamide (20 mg/kg per d) subcutaneously by osmotic minipump; NaCl, animals that received 0.2 (TRPV5⁺/⁺) or 0.1 M (TRPV5⁻/⁻) NaCl via the drinking water; NaHCO₃, animals that received 0.2 (TRPV5⁺/⁺) or 0.1 M (TRPV5⁻/⁻) NaHCO₃ via the drinking water. Data are presented as means ± SEM.

ᵇP < 0.05 versus respective TRPV5⁺/⁺ controls.
ᶜP < 0.05 versus respective TRPV5⁺/⁺ or TRPV5⁻/⁻ NaCl-treated mice.

### Table 2. Urine composition and fluid intake during chronic metabolic acidosis and alkalosis in TRPV5⁺/⁺ and TRPV5⁻/⁻ mice

<table>
<thead>
<tr>
<th>Treatment</th>
<th>pH</th>
<th>Urine Volume (mL/24 h)</th>
<th>Na⁺ Excretion (mmol/24 h)</th>
<th>K⁺ Excretion (mmol/24 h)</th>
<th>Li⁺ Clearance (µL/min)</th>
<th>Urine Osmolarity (mOsmol/kg)</th>
<th>Fluid Intake (mL/24 h)</th>
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</thead>
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<tr>
<td>TRPV5⁺/⁺</td>
<td>controls</td>
<td>7.0 ± 0.2</td>
<td>5.3 ± 1.4</td>
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<td>1.4 ± 0.1</td>
<td>13 ± 1</td>
<td>2434 ± 175</td>
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<td>NH₄Cl</td>
<td>6.0 ± 0.1ᵇ</td>
<td>5.1 ± 0.7</td>
<td>0.5 ± 0.1</td>
<td>1.2 ± 0.1</td>
<td>11 ± 1</td>
<td>2710 ± 246</td>
<td>12.1 ± 0.3</td>
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<td>ACTZ</td>
<td>8.3 ± 0.1ᵇ</td>
<td>9.7 ± 0.9ᵇ</td>
<td>0.7 ± 0.2ᵇ</td>
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<td>19 ± 3ᵇ</td>
<td>1720 ± 64ᵇ</td>
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<td>NaCl</td>
<td>7.5 ± 0.2</td>
<td>17.1 ± 1.7ᵇ</td>
<td>3.6 ± 0.3ᵇ</td>
<td>1.4 ± 0.1</td>
<td>30 ± 2ᵇ</td>
<td>1484 ± 52ᵇ</td>
<td>26.7 ± 4.8</td>
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<td>NaHCO₃</td>
<td>8.8 ± 0.1ᵇ,ᶜ</td>
<td>10.8 ± 3.3ᵇ</td>
<td>1.6 ± 0.2ᵇ,ᶜ</td>
<td>1.0 ± 0.1ᶜ</td>
<td>17 ± 3ᵇ,ᶜ</td>
<td>1540 ± 146ᵇ</td>
<td>27.4 ± 2.4</td>
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<td>TRPV5⁻/⁻</td>
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<td>6.0 ± 0.1</td>
<td>18.7 ± 2.4</td>
<td>0.5 ± 0.1</td>
<td>1.5 ± 0.1</td>
<td>16 ± 2</td>
<td>1113 ± 101</td>
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<td>NH₄Cl</td>
<td>5.4 ± 0.1ᵇ</td>
<td>15.8 ± 1.7</td>
<td>0.5 ± 0.1</td>
<td>1.3 ± 0.2</td>
<td>13 ± 2</td>
<td>1488 ± 99</td>
<td>26.4 ± 4.1</td>
</tr>
<tr>
<td>ACTZ</td>
<td>7.4 ± 0.2ᵇ</td>
<td>33.1 ± 4.0ᵇ</td>
<td>0.9 ± 0.1ᵇ</td>
<td>2.3 ± 0.1ᵇ</td>
<td>19 ± 1ᵇ</td>
<td>709 ± 28ᵇ</td>
<td>44.6 ± 4.8</td>
</tr>
<tr>
<td>NaCl</td>
<td>6.1 ± 0.1</td>
<td>23.6 ± 2.2ᵇ</td>
<td>3.1 ± 0.5ᵇ</td>
<td>1.4 ± 0.1</td>
<td>25 ± 2ᵇ</td>
<td>1000 ± 30</td>
<td>29.4 ± 7.5</td>
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<tr>
<td>NaHCO₃</td>
<td>8.3 ± 0.1ᵇ,ᶜ</td>
<td>18.6 ± 5.7</td>
<td>1.3 ± 0.1ᵇ,ᶜ</td>
<td>0.6 ± 0.1ᵇ,ᶜ</td>
<td>11 ± 2ᵇ,ᶜ</td>
<td>1036 ± 90</td>
<td>37.1 ± 1.1</td>
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*Controls, animals that received deionized drinking water only; NH₄Cl, animals that received 0.28 (TRPV5⁺/⁺) or 0.14 M (TRPV5⁻/⁻) NH₄Cl via the drinking water; ACTZ, animals that received acetazolamide (20 mg/kg per d) subcutaneously by osmotic minipump; NaCl, animals that received 0.2 (TRPV5⁺/⁺) or 0.1 M (TRPV5⁻/⁻) NaCl via the drinking water; NaHCO₃, animals that received 0.2 (TRPV5⁺/⁺) or 0.1 M (TRPV5⁻/⁻) NaHCO₃ via the drinking water; n = 3 animals per cage. Data are presented as means ± SEM.

ᵇP < 0.05 versus respective TRPV5⁺/⁺ or TRPV5⁻/⁻ controls.
ᶜP < 0.05 versus respective TRPV5⁺/⁺ or TRPV5⁻/⁻ NaCl-treated mice.
acetazolamide significantly increased urine volume and natriuresis in wild-type and TRPV5−/− mice (Table 2).

Ca2+ Homeostasis during Chronic Metabolic Acidosis and Alkalosis

Genetic ablation of TRPV5 resulted in a strikingly increased calciuresis compared with wild-type littermates (Figure 1A). NH4Cl loading significantly enhanced urine Ca2+ excretion in wild-type mice, whereas Ca2+ excretion was not affected in TRPV5−/− mice. Likewise, acetazolamide treatment significantly enhanced calciuresis in wild-type mice, whereas this effect was not present in TRPV5−/− mice. Serum Ca2+ levels remained unaltered during NH4Cl loading and acetazolamide treatment (Figure 1C). In contrast to metabolic acidosis, NaHCO3 administration significantly reduced urine Ca2+ excretion in wild-type and TRPV5−/− mice. Likewise, acetazolamide treatment significantly reduced urine Ca2+ excretion in wild-type as well as in TRPV5−/− mice (Figure 1B). Serum Ca2+ levels and urine volume did not differ between the alkalosis and control groups (Figure 1D, Table 1). Because Li+ and Na+ are transported in parallel by the proximal tubule, endogenous Li+ clearance was used as an inverse measure of proximal tubular Na+ reabsorption, to which in turn passive Ca2+ reabsorption is functionally coupled (16). Li+ clearance was significantly increased by acetazolamide treatment and NaCl loading in wild-type and TRPV5−/− mice, suggesting decreased proximal tubular Na+ reabsorption (Table 1). Conversely, NaHCO3 loading decreased Li+ clearance compared with NaCl-treated controls.

Mg2+ Homeostasis during Chronic Metabolic Acidosis and Alkalosis

In addition, we evaluated the effect of the different treatment protocols on renal Mg2+ homeostasis in wild-type mice. Oral NH4Cl loading significantly enhanced Mg2+ excretion (Figure 2A), which was accompanied by decreased serum Mg2+ levels (Figure 2C). In contrast, acetazolamide displayed a Mg2+-sparing effect accompanied by a significantly increased serum Mg2+ concentration. Likewise, metabolic alkalosis that was induced by NaHCO3 treatment significantly reduced urine Mg2+ excretion (Figure 2B) and increased the serum Mg2+ level (Figure 2D).

Renal mRNA and Protein Expression of Ca2+ Transporters

For studying the effect of systemic acid-base status on renal Ca2+ transporter expression, TRPV5 and calbindin-D28K mRNA levels were determined by real-time quantitative PCR analysis, and protein abundance was analyzed by immunohistochemistry and immunoblotting. NH4Cl loading significantly reduced both TRPV5 and calbindin-D28K mRNA levels in kidney cortex of wild-type mice (Figure 3, A and C). Calbindin-D28K mRNA levels were significantly decreased in TRPV5−/− mice (Figure 3B) compared with TRPV5+/− mice. In addition, NH4Cl treatment further reduced calbindin-D28K mRNA levels compared with control TRPV5−/− mice. Figure 4A shows representative immunohistochemical images of kidney cortex probed with anti-TRPV5 and anti–calbindin-D28K antibodies. In addition, calbindin-D28K protein abundance was determined by immunoblotting (Figure 4B). In line with the mRNA levels, semiquantification of the immunohistochemical analysis showed that NH4Cl decreases TRPV5 and calbindin-D28K protein abundance (Figure 5, A and C). These results were confirmed by immunoblotting (100 ± 4 versus 63 ± 9% in wild-type and 100 ± 5 versus 75 ± 8% in TRPV5−/− mice, respectively). Acetazolamide treatment decreased TRPV5 mRNA and protein expression in wild-type mice (Figures 3A and 5A). Furthermore, calbindin-D28K protein abundance was reduced in acetazolamide-treated mice (Figure 5C), as confirmed by immunoblotting (100 ± 1 versus 63 ± 6 and 40 ± 8% in wild-type and TRPV5−/− mice, respectively). In contrast to metabolic acidosis, chronic metabolic alkalosis that was induced by NaHCO3 loading increased TRPV5 mRNA and protein expression in wild-type mice (Figures 3B and 5B). Likewise, calbindin-D28K expression was increased as determined by real-time PCR and immunohistochemistry (Figures 3D and 5D), as well as by
immunoblotting (100 ± 41 versus 293 ± 40%). Conversely, calbindin-D28K mRNA and protein levels were not significantly increased in NaHCO3-treated TRPV5/mice.

**Figure 2.** Urinary Mg2+ excretion and serum Mg2+ concentration during metabolic acidosis and alkalosis in wild-type mice. The effects of chronic metabolic acidosis and acetazolamide (A) as well as during metabolic alkalosis (B) on renal Mg2+ excretion and serum Mg2+ concentration (C and D) were determined in metabolic cage experiments (n = 9 animals; n = 3 animals per cage). Controls, animals that received deionized drinking water only; NH4Cl, animals that received 0.28 M NH4Cl via the drinking water; Acetazolamide, animals that received acetazolamide (20 mg/kg per d) subcutaneously by osmotic minipump; NaCl, animals that received 0.2 M NaCl via the drinking water; NaHCO3 animals that received 0.2 M NaHCO3 via the drinking water. Data are presented as means ± SEM. *P < 0.05 versus respective control group (controls or NaCl-treated animals).

**Figure 3.** Effect of chronic metabolic acidosis and alkalosis on renal mRNA expression of Ca2+ transport proteins in TRPV5+/+ and TRPV5−/− mice. Renal mRNA expression levels of the epithelial Ca2+ channel TRPV5 and the cytosolic Ca2+-binding protein calbindin-D28K (CaBP28K) were determined during chronic metabolic acidosis and acetazolamide treatment (A and C, respectively) and chronic metabolic alkalosis (B and D, respectively) by real-time quantitative PCR analysis as the ratio of hypoxanthine-guanine phosphoribosyl transferase (HPRT) and depicted as percentage of respective controls. Controls, animals that received deionized drinking water only; NH4Cl, animals that received 0.28 (TRPV5+/+) or 0.14 M (TRPV5−/−) NH4Cl via the drinking water; Acetazolamide, animals that received acetazolamide (20 mg/kg per d) subcutaneously by osmotic minipump; NaCl, animals that received 0.2 (TRPV5+/+) or 0.1 M (TRPV5−/−) NaCl via the drinking water; NaHCO3 animals that received 0.2 (TRPV5+/+) or 0.1 M (TRPV5−/−) NaHCO3 via the drinking water; n = 9 animals per group. Data are presented as means ± SEM. *P < 0.05 versus respective TRPV5+/+ or TRPV5−/− control group (controls or NaCl-treated animals); #P < 0.05 versus TRPV5+/+ controls.

**Effect of Acid-Base Status on Renal TRPM6 Expression**

Renal TRPM6 mRNA and protein expression levels were determined by real-time quantitative PCR analysis (Figure 6) and semi-quantitative immunohistochemistry (Figure 7A). Both NH4Cl loading and acetazolamide treatment significantly reduced renal TRPM6 mRNA as well as protein abundance in wild-type mice (Figures 6A and 7B), whereas NaHCO3-treated mice displayed increased TRPM6 expression (Figures 6B and 7C).

**Discussion**

This study demonstrated that systemic acid-base status regulates the expression of proteins that are involved in active Ca2+ and Mg2+ reabsorption. Our data showed that downregulation of renal Ca2+ transport proteins is responsible for the hypercalciuria during chronic metabolic acidosis. In contrast, the Ca2+-sparring effect of chronic metabolic alkalosis was associated with enhanced Ca2+ transporter abundance. However, the Ca2+-sparring action persisted in TRPV5−/− mice, suggesting that additional mechanisms apart from upregulation of active Ca2+ transport contribute to the hypocalciuria. Further-
more, metabolic acidosis decreased renal TRPM6 abundance as well as Mg\(^{2+}\) reabsorption, whereas metabolic alkalosis had the opposite effect. These data indicate that regulation of TRPM6 explains the effects of acid-base status on renal Mg\(^{2+}\) handling.

Chronic metabolic acidosis that was induced by NH\(_4\)Cl loading enhanced Ca\(^{2+}\) excretion and decreased the expression of the epithelial Ca\(^{2+}\) channel TRPV5 and the cytosolic Ca\(^{2+}\)-binding and buffering protein calbindin-D\(_{28k}\) in wild-type mice. Both proteins play a central role in active Ca\(^{2+}\) reabsorption in DCT/CNT (1). We showed that 0.14 M NH\(_4\)Cl loading induced a similar metabolic acidosis in polydipsic TRPV5\(^{-/-}\) mice compared with wild-type mice. Importantly, Ca\(^{2+}\) excretion was not altered during chronic metabolic acidosis in TRPV5\(^{-/-}\) mice, in which active Ca\(^{2+}\) reabsorption is effectively abolished (17). These results indicated that downregulation of Ca\(^{2+}\) transport proteins that are present in DCT/CNT underlies the increased Ca\(^{2+}\) excretion during NH\(_4\)Cl loading. Alternatively, increased Ca\(^{2+}\) mobilization from bone has been shown in chronic metabolic acidosis and was suggested to explain the Ca\(^{2+}\) wasting (4,29). Our study provides evidence for a primary renal Ca\(^{2+}\) leak. In line with our data, previous micropuncture experiments suggested that tubular Ca\(^{2+}\) reabsorption in DCT/CNT is specifically diminished during chronic metabolic acidosis in dogs (2). In contrast, Rizzo et al. (30) previously reported that NH\(_4\)Cl-induced acidosis in rats was accompanied by a moderate although significant increase of calbindin-D\(_{28k}\). The reason for the discrepancy with our study, which shows a consistent decrease of TRPV5 and calbindin-D\(_{28k}\) mRNA as well as protein expression, is not known.

Mice that receive NH\(_4\)Cl develop chronic metabolic acidosis that is characterized by a slight decrease in blood pH, a reduced serum HCO\(_3^-\) concentration, and, importantly, a low urine pH as substantiated in this study (24). Vennekens et al. (31) recently demonstrated that extracellular protons inhibit TRPV5 in vitro by titrating glutamate 522 in the extracellular loop between the fifth putative transmembrane domain and the pore region as shown by Yeh et al. (32). Therefore, it was suggested that acidification of the DCT/CNT luminal fluid during chronic metabolic acidosis explains the decreased Ca\(^{2+}\) reabsorption in vivo (31,32). These data raised the question of whether the regulation of Ca\(^{2+}\) transport proteins is secondary to the acidosis per se or due to the low urine pH. This urine acidification has been attributed mainly to parallel enhancement of apical Na\(^+\)/H\(^+\) exchanger (NHE3) and basolateral Na\(^+\)-HCO\(_3^-\) co-transporter activity in the proximal tubule (33–36). In contrast, acetazolamide treatment is known to induce metabolic acidosis by diminishing this proximal tubular HCO\(_3^-\) reabsorptive capacity (18–20). This would result in an increased luminal pH at more distal nephron segments, including TRPV5 and calbindin-D\(_{28k}\)-expressing DCT/CNT. In our study, acetazolamide-treated mice indeed displayed urine alkalinization. Importantly, acetazolamide downregulated the expression of Ca\(^{2+}\) transport proteins. Thus, luminal pH in DCT/CNT does not seem to be crucial in the long-term in vivo hypercalciuric effect of chronic metabolic acidosis. Therefore, our data provide a molecular explanation for the increased Ca\(^{2+}\) excretion in clinically relevant situations, including chronic renal failure, chronic diarrhea, and renal tubular acidosis. In particular, acetazolamide treatment is a common cause of proximal renal tubular acidosis and is
Conversely, chronic metabolic alkalosis is known to decrease Ca\(^{2+}\) reabsorption and, thereby, induces hypercalciuria. However, TRPV5 ablation resulting in the functional lack of active Ca\(^{2+}\) transport did not preclude this effect, suggesting that upregulation of Ca\(^{2+}\) transport proteins in DCT/CNT is not crucial for the induction of hypocalciuria. It is interesting that alkalosis did not enhance calbindin-D\(_{28K}\) abundance in TRPV5\(^{-/-}\) mice. This is in line with previous studies from our laboratory, which demonstrated that blockade of the TRPV5-mediated Ca\(^{2+}\) influx in rabbit CNT/CCD cells downregulates calbindin-D\(_{28K}\) expression (39). This indicated that regulation of the latter protein is highly dependent on the presence of TRPV5. The bulk of filtered Ca\(^{2+}\) is reabsorbed by a passive paracellular mechanism that is localized primarily in the proximal tubule and to a lesser extent in TAL (1). In these nephron segments, Ca\(^{2+}\) reabsorption is secondary to Na\(^{+}\) reabsorption and the resulting water reabsorption, which creates a favorable electrochemical gradient driving passive Ca\(^{2+}\) transport. Determination of Li\(^{+}\) clearance indeed suggested that NaHCO\(_3\)-treated mice show increased proximal tubular Na\(^{+}\) reabsorption and, therefore, possibly display enhanced reabsorption.
passive $\text{Ca}^{2+}$ reabsorption compared with NaCl-treated controls. Thus, the hypocalciuria could alternatively be explained by increased passive $\text{Ca}^{2+}$ reabsorption. Taken together, the present data offer insight into the previously unexplained mechanism by which administration of $\text{HCO}_3^-$ prevents nephrolithiasis in patients with recurrent kidney stones (40,41). Up-regulation of $\text{Ca}^{2+}$ transport proteins in DCT/CNT occurs but is not crucial for the $\text{Ca}^{2+}$-sparing effect.

The epithelial $\text{Mg}^{2+}$ channel TRPM6 is the first identified protein involved in active $\text{Mg}^{2+}$ reabsorption (7–9). TRPM6 was localized along the apical membrane of DCT, and mutations in the gene encoding TRPM6 were shown to cause autosomal recessive hypomagnesemia, characterized by inappropriately high $\text{Mg}^{2+}$ excretion and disturbed intestinal $\text{Mg}^{2+}$ absorption. In our study, NH4Cl-induced chronic metabolic acidosis decreased renal TRPM6 abundance accompanied by increased $\text{Mg}^{2+}$ excretion and hypomagnesemia. Conversely, chronic metabolic alkalosis increased TRPM6 expression as well as renal $\text{Mg}^{2+}$ reabsorption, resulting in hypermagnesemia. There is insufficient functional information available regarding the tubular segments that are involved in the altered $\text{Mg}^{2+}$ reabsorption, but Wong et al. (10,42) previously demonstrated altered $\text{Mg}^{2+}$ reabsorption in the distal tubule during metabolic acidosis and alkalosis in the dog. Furthermore, a high extracellular pH was shown to enhance $\text{Mg}^{2+}$ uptake in isolated mouse DCT cells, and conversely a low pH diminished this uptake (43). Thus, these data suggest that alterations of acid-base status regulate TRPM6 expression, thereby affecting renal active $\text{Mg}^{2+}$ reabsorption in DCT and leading to significant changes in serum $\text{Mg}^{2+}$. We previously demonstrated that thiazide administration as well as treatment with the hypomagnesemic immunosuppressant tacrolimus (FK506) reduces renal TRPM6 abundance accompanied by increased urine $\text{Mg}^{2+}$ loss (15,16). Therefore, TRPM6 downregulation seems to be a general mechanism explaining the renal $\text{Mg}^{2+}$ leak and resulting hypomagnesemia in these important clinical situations.

It is interesting that we showed that acetazolamide displays a $\text{Mg}^{2+}$-sparing effect in mice. This carbonic anhydrase inhibitor is also known to result in hypomagnesuria in human via an unknown mechanism (21–23). Importantly, TRPM6 expression was significantly diminished during chronic acetazolamide treatment. This suggests that chronic metabolic acidosis, irrespective of cause or associated urine pH, downregulates TRPM6 expression. At the same time, these data are against enhanced active $\text{Mg}^{2+}$ reabsorption explaining the decreased $\text{Mg}^{2+}$ excretion. In contrast to $\text{Ca}^{2+}$, the bulk of filtered $\text{Mg}^{2+}$ is reabsorbed in TAL, where the $\text{Na}^+\text{-K}^+\text{Cl}^-$ cotransporter is responsible for maintaining the required electrochemical gradient (44). Acetazolamide treatment was associated with increased urine volume and Na$^+$ excretion, which has been shown to result in extracellular volume contraction and enhanced NKCC2 activity (45,46). Furthermore, metabolic acidosis, which developed in acetazolamide-treated wild-type mice, was previously shown to increase NKCC2 mRNA and protein abundance (14,47). Thus, we postulate that these additive stimulatory effects enhance passive $\text{Mg}^{2+}$ reabsorption in TAL. Alternatively, regulation of the tight junction protein paracellin-1, which is supposed to facilitate paracellular reabsorption of $\text{Mg}^{2+}$ in TAL, might be involved (48,49). Together, these mechanisms could counteract the metabolic ac-
dosis-induced TRPM6 downregulation and result in a net Mg\(^{2+}\)-sparing effect.

The mechanism translating the acid-base status to regulation of gene expression remains largely unknown. Our results show that transcriptional regulation occurs irrespective of urine pH. Because intracellular pH will ultimately reflect pH of the urine, apical or intracellular acid sensing does not seem to be involved. Therefore, direct sensing of acid-base status by pH-sensitive proteins at the basolateral membrane is likely. In DCT/CNT, the extracellular Ca\(^{2+}\)/Mg\(^{2+}\)-sensing receptor is expressed at the basolateral membrane (50,51). It was shown recently that extracellular pH directly influences the sensitivity of this receptor to Ca\(^{2+}\) and Mg\(^{2+}\) (52,53). Therefore, altered Ca\(^{2+}\)/Mg\(^{2+}\) sensing might influence transepithelial divalent transport in these nphron segments. Furthermore, other basolateral proton-sensing receptors or channels might act as an acid sensor regulating Ca\(^{2+}\) and Mg\(^{2+}\) transport protein expression (54,55). Thus, whereas the presented data elucidate the mechanism by which acid-base status affects expression of these transporters.

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