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cough". Three weeks later, she progressively developed a symmetrical eruption composed of 1-3 mm erythematous papules on the knees, ankles, elbows and forearms. She had mild fever and was quite miserable. Cervical and axillar lymphadenopathy was present. Liver and spleen were enlarged, being 2 cm and 1 cm respectively below the costal margins. A clinical diagnosis of Gianotti-Crosti syndrome was made. Laboratory studies showed mild microcytic anaemia (Hb 11.8) g/dl, MCV 75 μ^3), lymphocytosis (leucocytes 15300/mm³ with 15% neutrophils, 2.5% eosinophils, 1% basophils, 7.5% monocytes, 72.5 lymphocytes and 1,5% plasmocytes), and elevated liver enzymes (AST 74 U/I, ALT 69 U/I). Seroconversion to EBV was found with anti-viral capsid antigen IgM antibodies positive and anti-viral capsid antigen IgG titres raising from 1/40 to 1/320 in 2 months. Antibodies to human immunodeficiency virus 1–2, cytomegalovirus, hepatitis A-B-C, parvovirus and toxoplasmosis remained negative. The child was unwell for 3 weeks and the eruption then gradually disappeared over a 3week period. No treatment was given.

Children may develop Gianotti-Crosti syndrome associated with EBV as the sole stimulatory factor, and it might well have been the case with this child. However, since Gianotti-Crosti syndrome can be considered the result of an individual immune response to a viral infection and not the direct consequence of the virus itself, and since it has also been described after immunization, it is reasonable to think that in this child, both the immunization and the EBV infection acted as triggers to the eruption. Such a sequence is in accordance with the recent suggestion that Gianotti-Crosti syndrome may need both a previous immune stimulation and a viral infection [1].

References

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Studies with pre-treatment of milk with calcium acetata to reduce the phosphate content

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Sir: In the management of chronic and end stage renal failure the dietary intake of potassium and phosphate has to be reduced. Drugs that bind these compounds are prescribed on a large scale. These drugs have an unpleasant taste and have to be administered in relatively large volumes. Pre-treatment of drinks with calcium polystyrene-sulphonate leads to a 50% reduction in potassium content [1, 2]. The present study aims to reduce the phosphate content of milk, since this is the major compound of infant nutrition. The method must be simple enough to be applied at home by the patient, and must not negatively influence the taste.

Five ml of normal cows milk were treated with calcium acetate (CaAc) in different quantities (15, 30, 45, 60, 75, and 150 mg). Samples were mixed thoroughly, and stored for different periods (30, 60, 90, 120 min and 21 h) at two different temperatures (4°C and room temperature). After filtration at 4°C pH as well as calcium, phosphate, and protein contents were measured.

Using this method no significant decrease in phosphate concentration was observed. Table I shows the results of the addition of different quantities of CaAc on calcium and phosphate concentrations. Calcium concentration increased to 93 mmol/l with the highest CaAc addition.

Table 1 Effect of different amounts of CaAc on phosphate and calcium concentration in 5 ml of cows milk (4°C, 30 min)

	Phosphate (mmol/l)	
Untreated	30	32
15 mg CaAc	31	52
30 mg CaAc	27	61
45 mg CaAc	27	74
60 mg CaAc	28	87
75 mg CaAc	25	78
150 mg CaAc	28	93

Table 2 Influence of duration of incubation on calcium and phosphate concentration after the addition of 30 mg of CaAc, at two different temperatures. The first number gives the concentration at 4°C, the second at room temperature

Incubation	Phosphate (mmol/l)	Calcium (mmol/l)
30 m	30/30	66/64
60 m	27/30	60/63
90 m	35/34	72/56
120 m	29/29	63/63
21 h	31/31	65/64

pH and protein contents decreased markedly (from 6.71 to 6.05 and from 37 to 29 g/l with 45 mg of CaAc). The milk developed an unpleasant sour taste if more than 30 mg of CaAc was added. The influence of duration of incubation and of temperature is given in Table 2, using an amount of 30 mg of CaAc.

From these experiments it is clear that a simple precipitation reaction with CaAc is not a suitable method to reduce the phosphate content of milk. Other methods, such as dialysis may be applied, but these techniques are not suitable for routine domestic use

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