

## PDF hosted at the Radboud Repository of the Radboud University Nijmegen

The following full text is a publisher's version.

For additional information about this publication click this link.

<http://hdl.handle.net/2066/144080>

Please be advised that this information was generated on 2019-02-18 and may be subject to change.

RESEARCH ARTICLE

# To Float or Not to Float: How Interactions between Light and Dissolved Inorganic Carbon Species Determine the Buoyancy of *Stratiotes aloides*

Sarah F. Harpenslager<sup>1\*</sup>, Alfons J. P. Smolders<sup>1,2</sup>, Ariët A. M. Kieskamp<sup>1</sup>, Jan G. M. Roelofs<sup>1</sup>, Leon P. M. Lamers<sup>1</sup>

**1** Department of Aquatic Ecology and Environmental Biology, Institute for Water and Wetland Research, Radboud University, Nijmegen, The Netherlands, **2** B-Ware Research Centre, Nijmegen, The Netherlands

\* [S.Harpenslager@science.ru.nl](mailto:S.Harpenslager@science.ru.nl)



OPEN ACCESS

**Citation:** Harpenslager SF, Smolders AJP, Kieskamp AAM, Roelofs JGM, Lamers LPM (2015) To Float or Not to Float: How Interactions between Light and Dissolved Inorganic Carbon Species Determine the Buoyancy of *Stratiotes aloides*. PLoS ONE 10(4): e0124026. doi:10.1371/journal.pone.0124026

**Academic Editor:** Panagiotis Kalaitzis, Mediterranean Agronomic Institute at Chania, GREECE

**Received:** October 30, 2014

**Accepted:** March 9, 2015

**Published:** April 24, 2015

**Copyright:** © 2015 Harpenslager et al. This is an open access article distributed under the terms of the [Creative Commons Attribution License](https://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

**Data Availability Statement:** All relevant data are within the paper.

**Funding:** Research performed as part of Project: 11264; PeatCap: Natural capping in the polluted Volgermeer Polder (the Netherlands) provides new chances for peat growth counteracting land subsidence. Subsidised by Technology Foundation STW, giving grants to applied research.

**Competing Interests:** The authors have declared that no competing interests exist.

## Abstract

Structural diversity formed by dense, floating *Stratiotes aloides* stands, generates hotspots of biodiversity of flora and fauna in wetlands. However, only part of the populations become emergent and provide this important facilitation. Since it has been hypothesised that its buoyancy depends on the rates of underwater photosynthesis, we investigated the role of dissolved CO<sub>2</sub> availability and PAR on photosynthesis, biomass production and buoyancy in a controlled greenhouse experiment. Photosynthesis and growth were strongly influenced by both PAR and CO<sub>2</sub> availability. At low PAR, plants formed less biomass and produced no emergent leaves, even when CO<sub>2</sub> was abundant. At low CO<sub>2</sub> levels, *S. aloides* switched to HCO<sub>3</sub><sup>-</sup> use, resulting in a lower photosynthetic O<sub>2</sub> production, decreased emergent leaf formation and increased CaCO<sub>3</sub> precipitation on its leaves, all of which impaired buoyancy. At high PAR, low CO<sub>2</sub> availability resulted in slower colonisation of the water layer, whereas CO<sub>2</sub> availability did not influence PAR-limited plants. Our study shows that site conditions, rather than the sole abundance of potentially facilitating species, may strongly determine whether or not they form the structure necessary to act as a facilitator for biodiversity in aquatic environments.

## Introduction

In the natural succession of wetlands, the process of terrestrialisation, the transition from an aquatic to a terrestrial phase, is very important in shaping a structurally diverse habitat and supporting high biodiversity [1, 2]. However, changes in land use and water quality during the 20<sup>th</sup> century have severely influenced the functioning of wetlands in Western Europe [2, 3] and natural succession of open water to species-rich marshes through the formation of floating sediments or vegetation mats has become rare [1]. Floating stands of emergent aquatic macrophytes, such as *Stratiotes aloides*, can provide structure for a wide variety of flora and fauna

species, including the endangered green hawkler (*Aeshna viridis*) and black tern (*Chlidonias niger*) [4–6]. Furthermore, biodiversity in these vegetation types is much higher than in similar systems where the species is absent [7, 8].

*S. aloides* used to be very common in Europe and North-western Asia until the 1960s but has since declined severely by changes in hydrology, by eutrophication and by ammonium ( $\text{NH}_4^+$ ) or sulphide ( $\text{H}_2\text{S}$ ) toxicity [3, 9, 10]. Through their high clonal growth rates, *S. aloides* populations can completely fill up surface waters within a few growing seasons under meso- to eutrophic conditions [11]. Such dense vegetation also decreases turbidity of the water layer by preventing re-suspension of sediment particles [12]. Furthermore, the species produces allelopathic substances that reduce algal growth [13]. As a result of these traits, the species is a strong competitor in aquatic systems and has even been reported as a noxious weed in parts of North America and Australia, where it is non-native [14–16].

One of the most characteristic features of *S. aloides* is its alternating life cycle with an emergent and submerged life stage [11]. During summer, plants may become buoyant and can form dense floating vegetation mats. In autumn, they sink and remain submerged throughout the winter until they re-emerge in spring. The increased buoyancy of *S. aloides* in spring has been speculated to be the result of enhanced underwater photosynthesis [9, 11, 17, 18]. However, field observations and reports from literature clearly show that not all *S. aloides* stands follow this alternating life cycle and that populations may remain submerged during summer [17, 19, 20]. Understanding the factors that determine whether a population remains submerged or becomes emergent is important, not only to understand the ecophysiology of floating macrophytes, but also because submerged populations show a different community composition and much lower biodiversity than emergent populations [21] and are unable to reproduce generatively [11, 22].

So far, the mechanism behind buoyancy of *S. aloides* has not yet been unravelled under controlled experimental conditions, but as it has been hypothesised to be the direct result of photosynthesis, the absence of floating plants will most probably relate to decreased submerged photosynthetic activity. Reduced photosynthesis may not only be the result of limited nutrient (phosphorus, nitrogen) availability, but also of limited light (PAR; photosynthetically active radiation) conditions, or reduced dissolved inorganic carbon (DIC) availability. Increased turbidity and the presence of humic acids or algae can indeed limit PAR for *S. aloides* and thereby its photosynthetic rate [23]. C limitation, on the other hand, may not only occur in softwater lakes, typically showing low DIC concentrations, but also in alkaline waters with increased pH (>8.2) in which DIC is present as bicarbonate ( $\text{HCO}_3^-$ ) instead of carbon dioxide ( $\text{CO}_2$ ) [24]. Aquatic plant species differ considerably in their efficiency to use  $\text{HCO}_3^-$  as an alternative to  $\text{CO}_2$  [25, 26]. Although *S. aloides* is able to use  $\text{HCO}_3^-$  [27], it may only reach approximately 15% of its maximum net photosynthesis when grown solely on  $\text{HCO}_3^-$  [17].

To understand the ecophysiological traits and the potential role of *S. aloides* as a facilitating species, it is essential to understand the optimal abiotic conditions for floating mat formation. In this study, we therefore tested the interacting effects of PAR and  $\text{CO}_2$  availability on the buoyancy of *S. aloides* in a controlled greenhouse experiment. It was hypothesised that plants that experienced either limited PAR or low  $\text{CO}_2$  availability would survive but not become buoyant, since their photosynthetic capacity would be impaired. This process was expected to be even more profound when both  $\text{CO}_2$  and PAR are limiting. Given their facilitating role, research into the regulation of buoyancy of *S. aloides* plants is also highly relevant for restoration projects focussing on the natural succession of wetlands and the conservation of endangered species.

## Materials and Methods

### Experimental set-up

In total, 54 *Stratiotes aloides* plants were collected from a privately owned garden pond in the Netherlands, with permission from the owner (51°44'6"N; 6°51'49"E), in autumn 2011 and kept in artificial ponds inside the greenhouse facility of the Radboud University Nijmegen, where the experiment was carried out between April and July 2012. Nine experimental ponds of 2300 L (ø 180 cm; h 90 cm) were each divided into three equally sized compartments by nets to ensure free water movement. One compartment was covered with a double layer of shadow fabric, one with a single layer and one was kept clear of fabric to ensure full light conditions. As a result, three light intensities were created. Taking full light as 100% PAR intensity, shading led to ±40% and ±10% of full PAR intensity for the shaded and double-shaded treatments respectively. Light was mostly natural, but an artificial light regime with 200 μmol m<sup>-2</sup>s<sup>-1</sup> lamps (Philips, Master SonT, 400W) of 16h light/8h dark was maintained to prevent large fluctuations in the light availability. Average PAR levels at the water surface level reached 130 ± 32, 50 ± 22 and 15 ± 5 μmol m<sup>-2</sup> s<sup>-1</sup> (mean ± SEM) for the 100%, 40% and 10% PAR treatments respectively.

Before the addition of the plants, all experimental units received 125 L of underwater peat sediment (see [Table 1](#) for sediment characteristics) originating from a minerotrophic peatland

**Table 1. Statistical results (*P*- and *F*-values) of CO<sub>2</sub>, PAR and their interactions on plant parameters and sediment nutrient development during the experiment.**

| Characteristic                        | CO <sub>2</sub> |          | PAR          |          | CO <sub>2</sub> * PAR |          |
|---------------------------------------|-----------------|----------|--------------|----------|-----------------------|----------|
|                                       | <i>P</i>        | <i>F</i> | <i>P</i>     | <i>F</i> | <i>P</i>              | <i>F</i> |
| CO <sub>2</sub> treatment             | <b>0.000</b>    | 233.082  |              |          |                       |          |
| NH <sub>4</sub> <sup>+</sup> decrease | <b>0.005</b>    | 5.485    | <b>0.000</b> | 15.028   | <b>0.024</b>          | 2.912    |
| Cover increase                        | <b>0.002</b>    | 6.682    | <b>0.000</b> | 40.760   | <b>0.012</b>          | 3.386    |
| Vegetative reproduction               | <b>0.013</b>    | 4.617    | <b>0.000</b> | 79.765   | <b>0.016</b>          | 3.265    |
| Rosette depth                         | <b>0.000</b>    | 23.377   | <b>0.000</b> | 37.767   | <b>0.005</b>          | 3.883    |
| CaCO <sub>3</sub> accumulation        | <b>0.001</b>    | 8.057    | <b>0.000</b> | 17.285   | 0.581                 | 0.721    |
| Final biomass                         | 0.386           | 1.005    | <b>0.000</b> | 53.006   | 0.506                 | 0.861    |
| No of roots                           | <b>0.018</b>    | 5.336    | <b>0.000</b> | 17.160   | 0.566                 | 0.763    |
| Final length of roots                 | 0.712           | 0.348    | <b>0.001</b> | 11.617   | 0.068                 | 2.745    |
| DW/ FW ratio                          | 0.552           | 0.615    | <b>0.005</b> | 7.341    | 0.468                 | 0.931    |
| Leaf thickness                        | 0.061           | 3.401    | <b>0.006</b> | 7.432    | 0.631                 | 0.658    |
| Photosynthesis                        | <b>0.003</b>    | 8.067    | <b>0.000</b> | 12.911   | 0.335                 | 1.225    |
| PAM                                   | 0.668           | 0.415    | <b>0.002</b> | 9.847    | 0.152                 | 1.966    |
| Chlorophyll                           | 0.499           | 0.722    | <b>0.012</b> | 5.647    | 0.288                 | 1.356    |
| C: N ratio                            | 0.055           | 3.447    | <b>0.000</b> | 15.307   | 0.509                 | 0.856    |
| C: P ratio                            | 0.723           | 0.330    | 0.163        | 2.022    | 0.349                 | 1.194    |
| N: P ratio                            | 0.180           | 1.898    | 0.242        | 1.546    | 0.749                 | 0.481    |
| C                                     | <b>0.006</b>    | 6.984    | <b>0.004</b> | 7.937    | 0.202                 | 1.677    |
| N                                     | <b>0.004</b>    | 7.882    | <b>0.000</b> | 22.340   | 0.093                 | 2.373    |
| P                                     | 0.764           | 0.274    | <b>0.008</b> | 6.411    | 0.740                 | 0.495    |
| K                                     | 0.893           | 0.113    | 0.507        | 0.706    | 0.376                 | 1.128    |
| Fe                                    | <b>0.012</b>    | 5.815    | 0.747        | 0.297    | 0.389                 | 1.098    |

Significant *P*-values (*P*<0.05) are indicated in bold.

doi:10.1371/journal.pone.0124026.t001

area in the Netherlands (52°18'32"N; 4°45'42"E) and a water layer of 1500 L (60 cm) Nijmegen tap water. Since the experimental basins were dug into the greenhouse floor, water temperature remained quite constant (Median: 23.3°C; Min: 21.1°C; Max: 24.8°C). As a result of the alkaline, bicarbonate ( $\text{HCO}_3^-$ ) rich sediments, surface water pH was  $7.8 \pm 0.1$  (mean  $\pm$  SEM) and  $\text{CO}_2$  concentration was  $86 \pm 53 \mu\text{mol L}^{-1}$  (mean  $\pm$  SEM). Two higher  $\text{CO}_2$  treatments were created by gently bubbling the water layer with pure  $\text{CO}_2$  4–8 times a day using mass flow controllers (EL-FLOW select F201CV, Bronkhorst, Veenendaal, the Netherlands). These two treatments had concentrations of  $228 \pm 106 \mu\text{mol L}^{-1}$  (pH  $7.4 \pm 0.1$ ; mean  $\pm$  SEM) and  $933 \pm 436 \mu\text{mol L}^{-1}$  (pH  $7.0 \pm 0.1$ ; mean  $\pm$  SEM)  $\text{CO}_2$ . In the Results and Discussion sections, treatments are called 90, 230 and 930  $\mu\text{mol L}^{-1}$  respectively. During the experiment, the  $\text{CO}_2$  concentrations of the water layer were measured three times a week using an ABB Advance Optima Infrared Gas Analyser (ABB Analytical, Frankfurt, Germany) and treatments remained significantly different throughout the experiment ( $P = 0.018$ ; [Table 1](#)).

## Chemical analyses

Water layer and pore water samples were collected every two weeks. Since the water layers of the three compartments of each mesocosm were connected and pH and  $\text{CO}_2$  concentrations showed no differences between the compartments, analyses were performed on one pooled sample from the water layer of each mesocosm. Furthermore, pore water samples of all individual compartments were taken using vacuum syringes attached to ceramic cups (Eijkelkamp Agrisearch Equipment, Giesbeek, The Netherlands) that were fixed at a depth of 10 cm in the sediment. pH was measured with a standard combined glass Ag/AgCl pH electrode (Orion Research, Beverly, CA, USA) connected to a pH meter (Tim800; Radiometer analytical, Lyon, France) and alkalinity by titrating down to pH 4.2 with  $0.1 \text{ mmol L}^{-1}$  HCl using an auto burette (ABU901, Radiometer, Lyon, France). Concentrations of  $\text{NO}_3^-$  and  $\text{NH}_4^+$  were measured colourimetrically on an auto analyser 3 system (Bran&Lubbe, Norderstedt, Germany) using hydrazine sulphate [28] and salicylate [29] respectively. Concentrations of Ca, Fe, K, Mg, total-P and S were analysed by inductively coupled plasma spectrometry (ICP-OES icap 6000; Thermo Fischer scientific, Waltham, MA, USA).

## Plant parameters

While being kept in the greenhouse facility during winter 2011–2012, all plants shed their roots and started forming new ones in April 2012, just before the start of the experiment. Fresh weights, numbers of leaves and offsets, and lengths of the three largest leaves were recorded for each plant before placing two *S. aloides* plants ( $24.6 \pm 1.6 \text{ g DW}$ ; diameter  $48 \pm 0.5 \text{ cm}$ ; mean  $\pm$  SEM) in each compartment. The plants were allowed to grow for 4 months, during which the position of the rosette relative to the water surface (measure of buoyancy), production of emergent leaves, number of offsets and the number of roots penetrating the sediment were recorded regularly. Furthermore, plant mortality during the experiment was recorded and remaining biomass of these dead plants was harvested prematurely. Since it was not possible to record biomass during the experiment without damaging the plants, plant coverage was estimated digitally (Photoshop CS6 for Mac, Measurements Tool; Adobe Systems Incorporated, Mountain View, CA, USA) from pictures that were taken every two weeks, to determine increase in plant cover. After harvest, root lengths were measured and the fresh weights of shoots, roots and offsets were determined separately. Plant material was dried at 70°C for 48h to establish final plant biomass dry weights. These values were corrected for calcium carbonate ( $\text{CaCO}_3$ ) precipitation.

At the start and at the end of the experiment, the maximum quantum yields of photochemistry ( $F_v/F_M$ ) were measured for all plants using a Pulse Amplified Modulation fluorometer (JUNIOR-PAM, Walz, Effeltrich, Germany). This method provides a relative measure of the reaction centres of photosystem II that are actively involved in photosynthesis. Also, at the start and at the end of the experiment, medium aged leaves were collected for analysis of chlorophyll content, which was extracted and measured on a spectrophotometer (Lambda 25, UV/VIS Spectrometer, PerkinElmer Instruments), according to Lichtenthaler and Wellburn [30].

Photosynthetic rates of medium aged, submerged leaves were measured after 5 weeks of experimental treatments by incubating freshly cut leaves in airtight, water filled flasks.  $O_2$  production inside the flasks was determined after 0, 2 and 4 hours of incubation using an oxygen electrode (HQ40d multi, HACH, Loveland, Colorado, U.S.A.). We chose to measure after 5 weeks, since this was within the crucial period where all plants were still submerged but were expected to become emergent soon. Measurements were carried out at  $195 \mu\text{mol m}^{-2} \text{s}^{-1}$  PAR intensity, created by artificial light (Master SonT, 400W, Philips, The Netherlands). To maintain experimental light conditions, flasks containing leaves from 40% and 10% light treatments were covered with single and double layers of shadow fabric respectively and flasks were filled with water from the corresponding  $CO_2$  treatments. Measurements were carried out at room temperature ( $22.5^\circ\text{C}$ ).

$CaCO_3$  precipitation on leaves was quantified after 25, 50 and 65 days by incubating leaves in airtight bottles completely filled with  $0.1 \text{ mmol L}^{-1}$  HCl for 24 hours, after which the  $CO_2$  concentration of the solution was measured using an ABB Advance Optima Infrared Gas Analyser (ABB Analytical, Frankfurt, Germany) and the amount of  $CaCO_3$  was calculated. Homogenised dried plant material was digested with  $HNO_3$  (65%) and  $H_2O_2$  (30%) using a microwave oven (mls 1200 Mega, Milestone Inc., Sorisole, Italy). Digestates were diluted and analysed for Fe, K and P by inductively coupled plasma spectrometry (ICP-OES icap 6000; Thermo Fischer Scientific, Waltham, MA, U.S.A.). In addition, C and N contents (%) of dried plant material were determined using an elemental analyser (Carlo Erba NA1500, Thermo Fisher Scientific, Waltham, MA, U.S.A.).

## Statistical analyses

Our experimental set-up ensured a full-factorial experiment containing all possible combinations of  $CO_2$  ( $90 \mu\text{mol L}^{-1}$ ,  $230 \mu\text{mol L}^{-1}$  and  $930 \mu\text{mol L}^{-1}$ ) and PAR (100%, 40% and 10%) treatments, all of which were replicated 3 times. All replicates consisted of 2 plants in the same compartment.

Normality of residuals and homogeneity of variance were checked using the Shapiro-Wilk Test for Normality and Levene's Test of Equality of Error Variances, respectively. Non-normal and heteroscedastic data were log transformed or square-rooted before analyses in order to meet the assumptions of parametric tests. Data on final biomass, water chemistry, chemical composition of plant material and photosynthetic parameters were analysed by two-way ANOVAs at the 0.05 confidence limit followed by a Tukey post hoc test. Experimental  $CO_2$  concentrations, pore water  $NH_4^+$  concentrations, plant cover, offset production, buoyancy depth and  $CaCO_3$  accumulation were analysed over time using linear mixed models. Pairwise comparisons were performed using a Bonferroni adjustment for multiple comparisons to determine significant differences between all possible combinations of  $CO_2$  or PAR treatments. For all analyses,  $P$  and  $F$  values and interaction effects are presented in [Table 1](#). All statistical tests were carried out using SPSS v21 (IBM Statistics, 2012).

## Results

### Biogeochemistry

Due to the  $\text{HCO}_3^-$ -rich sediments (chemical properties shown in Table 2), characteristic for minerotrophic mires, alkalinity of the water layer in the ponds was high and ranged from 2.4 meq  $\text{L}^{-1}$  to 3.8 meq  $\text{L}^{-1}$  (mean: 2.8 meq  $\text{L}^{-1}$ ). In general, nutrient concentrations in the water layer were low (Table 2). Pore water  $\text{NH}_4^+$  concentrations decreased by 17–53% for 10% PAR plants and by 78–99% for plants with 40% or 100% PAR during the experiment ( $P < 0.001$ ; data not shown).  $\text{CO}_2$  concentrations of 230 or 930  $\mu\text{mol L}^{-1}$  also resulted in a stronger decrease than when plants were grown with 90  $\mu\text{mol L}^{-1}$   $\text{CO}_2$  ( $P = 0.004$ ). An interaction effect between PAR and  $\text{CO}_2$  ( $P = 0.024$ ) indicated that the average decrease of  $98 \pm 1\%$  observed in plants grown at  $\geq 230 \mu\text{mol L}^{-1}$   $\text{CO}_2$  and  $\geq 40\%$  PAR (Table 1) was significantly stronger than the  $56 \pm 13\%$  for those grown at  $< 230 \mu\text{mol L}^{-1}$  and  $< 40\%$  PAR.

### Plant growth and photosynthesis

PAR was the most important factor determining final biomass of *Stratiotes aloides* (Table 3). While the biomass of PAR-limited plants decreased during the experiment due to the production of thinner leaves ( $P = 0.006$ ; Table 3) and shedding of large leaves, plants grown under 100% PAR had produced approximately three times their initial weight after four months of experimental treatments. Furthermore, these plants covered the water layer faster and more completely ( $P < 0.001$ ; Fig 1), and had a higher vegetative reproduction than PAR-limited plants ( $P < 0.001$ ; Table 1; Fig 2). Plants grown under 100% PAR produced  $1.7 \pm 0.2$  offsets per plant, whereas plants grown under 40% or 10% PAR conditions produced only  $0.4 \pm 0.2$  and  $0.05 \pm 0.05$  offsets respectively.

All plants started producing roots approximately 1 week before the start of the experiment and the plants were all firmly rooted in the sediment after 3 weeks. Plants grown at 100% PAR produced more roots ( $P < 0.001$ ; Table 1; Table 3) than those from the other light treatments, while plants from the lowest PAR treatment produced the shortest roots ( $P = 0.001$ ; Table 1; Table 3). From this 10% PAR treatment, however, a substantial part of the plants died ( $\pm 40\%$ ,

**Table 2. Chemical composition and characteristics (mean  $\pm$  SEM) of the water layer, sediment pore water and sediment moisture and organic matter content.**

|  | Water layer      | Sediment           |
|--|------------------|--------------------|
| Alkalinity (meq $\text{L}^{-1}$ )                    | 2.85 $\pm$ 0.11  | 9.82 $\pm$ 0.25    |
| Total Inorganic Carbon (TIC) (mmol $\text{L}^{-1}$ ) | 3.57 $\pm$ 0.37  | 11.86 $\pm$ 0.42   |
| $\text{Ca}^{2+}$ (mmol $\text{L}^{-1}$ )             | 3.14 $\pm$ 0.14  | 11.96 $\pm$ 0.31   |
| $\text{Fe}^{2+}$ ( $\mu\text{mol L}^{-1}$ )          | 0.50 $\pm$ 0.04  | 186.83 $\pm$ 11.50 |
| $\text{K}^+$ (mmol $\text{L}^{-1}$ )                 | 0.22 $\pm$ 0.01  | 0.50 $\pm$ 0.02    |
| $\text{Mg}^{2+}$ (mmol $\text{L}^{-1}$ )             | 0.63 $\pm$ 0.02  | 2.16 $\pm$ 0.06    |
| $\text{SO}_4^{2-}$ (mmol $\text{L}^{-1}$ )           | 2.48 $\pm$ 0.14  | 9.98 $\pm$ 0.28    |
| $\text{NH}_4^+$ ( $\mu\text{mol L}^{-1}$ )*          | 0.73 $\pm$ 0.01  | 79.52 $\pm$ 12.88  |
| $\text{NO}_3^-$ ( $\mu\text{mol L}^{-1}$ )           | 74.41 $\pm$ 9.52 | 2.06 $\pm$ 0.80    |
| Total P ( $\mu\text{mol L}^{-1}$ )                   | 0.68 $\pm$ 0.17  | 21.95 $\pm$ 1.14   |
| Moisture content (%)                                 |                  | 50 $\pm$ 7         |
| Organic matter content (%)                           |                  | 20 $\pm$ 6         |

\* For results of statistical analyses on the decline of  $\text{NH}_4^+$  concentrations in the pore water during the experiment, see Table 1. Please note the different units.

doi:10.1371/journal.pone.0124026.t002

**Table 3. Plant growth characteristics (mean ± SEM) of *Stratiotes aloides* after 4 months of different PAR and CO<sub>2</sub> treatments.**

| Characteristic             | Unit                           | 100% PAR                                |  |  | 40% PAR                                 |  |  | 10% PAR                                 |  |  |
|----------------------------|--------------------------------|---|--|--|---|--|--|---|--|--|
|                            |                                | 90 μmol L <sup>-1</sup> CO <sub>2</sub> | 230 μmol L <sup>-1</sup> CO <sub>2</sub> | 930 μmol L <sup>-1</sup> CO <sub>2</sub> | 90 μmol L <sup>-1</sup> CO <sub>2</sub> | 230 μmol L <sup>-1</sup> CO <sub>2</sub> | 930 μmol L <sup>-1</sup> CO <sub>2</sub> | 90 μmol L <sup>-1</sup> CO <sub>2</sub> | 230 μmol L <sup>-1</sup> CO <sub>2</sub> | 930 μmol L <sup>-1</sup> CO <sub>2</sub> |
| Final Plant Biomass        | g DW                           | 129.5 ±30.7 <sup>C</sup>                | 168.4±13.6 <sup>C</sup>                  | 128.7±19.1 <sup>C</sup>                  | 53±11.3 <sup>B</sup>                    | 63.7±12.3 <sup>B</sup>                   | 49.7±10.5 <sup>B</sup>                   | 19.9±2.8 <sup>A</sup>                   | 16.5±9.4 <sup>A</sup>                    | 27.3±2.9 <sup>A</sup>                    |
| Chlorophyll a+b            | mg Chl g DW <sup>-1</sup>      | 0.79±0.04 <sup>A</sup>                  | 0.68±0.18 <sup>A</sup>                   | 0.65±0.09 <sup>A</sup>                   | 1.24±0.26 <sup>B</sup>                  | 0.95±0.15 <sup>B</sup>                   | 0.82±0.08 <sup>B</sup>                   | 0.93±0.12 <sup>B</sup>                  | 1.14±0.14 <sup>B</sup>                   | 1.09±0.02 <sup>B</sup>                   |
| Photosynthetic Yield (PAM) | F <sub>v</sub> /F <sub>M</sub> | 0.70±0.01 <sup>A</sup>                  | 0.64±0.04 <sup>A</sup>                   | 0.72±0.02 <sup>A</sup>                   | 0.72±0.01 <sup>A</sup>                  | 0.76±0.01 <sup>A,B</sup>                 | 0.73±0.03 <sup>A,B</sup>                 | 0.75±0.01 <sup>B</sup>                  | 0.77±0.01 <sup>B</sup>                   | 0.78±0.01 <sup>B</sup>                   |
| Final number of roots      |                                | 3.3±1.6 <sup>B</sup>                    | 5.8±2.6 <sup>B</sup>                     | 8.7±2.0 <sup>B</sup>                     | 2.0±1.2 <sup>A</sup>                    | 3.3±0.9 <sup>A,B</sup>                   | 5.3±0.8 <sup>A,B</sup>                   | 1.7±1.4 <sup>A</sup>                    | 0.5±0.3 <sup>A</sup>                     | 1.3±0.7 <sup>A</sup>                     |
| Final length of roots      | cm                             | 88.2±4.0 <sup>B</sup>                   | 84.3±9.2 <sup>B</sup>                    | 94.3±4.4 <sup>B</sup>                    | 95.0±1.4 <sup>B</sup>                   | 81.3±12.1 <sup>B</sup>                   | 74.3±1.3 <sup>B</sup>                    | 36.5±0.4 <sup>A</sup>                   | 69.3±2.2 <sup>A</sup>                    | 65.7±14.8 <sup>A</sup>                   |
| DW/FW ratio leaves         |                                | 0.25±0.06 <sup>B</sup>                  | 0.24±0.03 <sup>B</sup>                   | 0.20±0.02 <sup>B</sup>                   | 0.17±0.02 <sup>A</sup>                  | 0.13±0.03 <sup>A</sup>                   | 0.16±0.003 <sup>A</sup>                  | 0.14±0.01 <sup>A</sup>                  | 0.13±0.01 <sup>A</sup>                   | 0.14±0.01 <sup>A</sup>                   |
| Leaf thickness             | mm                             | 0.53±0.12 <sup>B</sup>                  | 0.61±0.01 <sup>B</sup>                   | 0.56±0.06 <sup>B</sup>                   | 0.44±0.07 <sup>A,B</sup>                | 0.53±0.06 <sup>A,B</sup>                 | 0.40±0.02 <sup>A,B</sup>                 | 0.23±0.01 <sup>A</sup>                  | 0.47±0.01 <sup>A</sup>                   | 0.39±0.03 <sup>A</sup>                   |
| Emergent leaf formation    | weeks                          | 15                                      | 9  | 6  | -                                       | -  | -  | -                                       | -  | -  |

Significant differences among PAR treatments are indicated by different capital letters (A, B, C). See Table 1 for P and F values.

doi:10.1371/journal.pone.0124026.t003

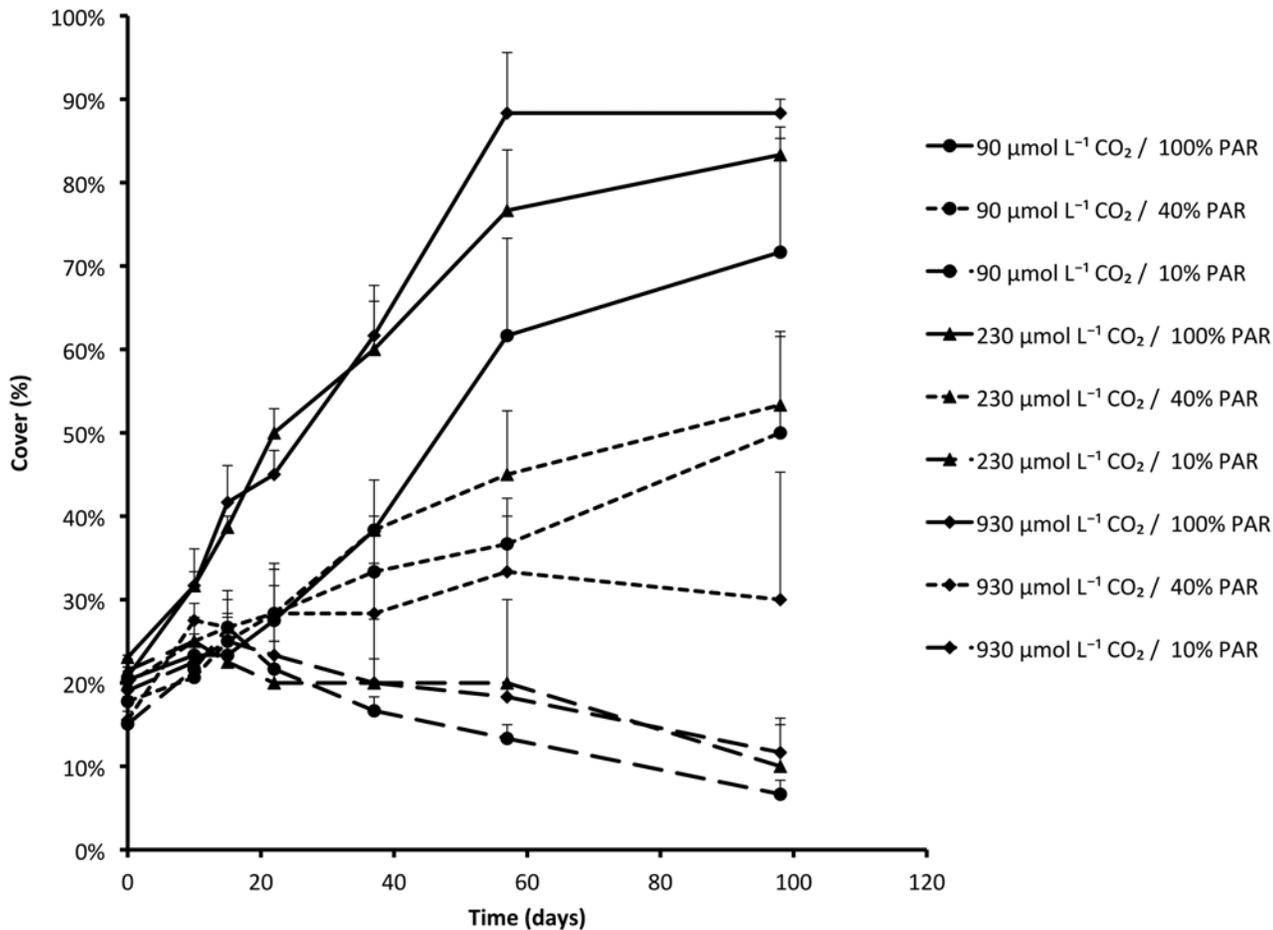
compared to ±20% and ±5% of the plants in the 40% and 100% PAR treatments respectively), which started with the dying off of the roots. A sufficient number of plants survived, however, for statistical analyses of plant parameters. The total length of the roots from all treatments, apart from the 10% PAR and 90 μmol L<sup>-1</sup> CO<sub>2</sub> treatment, measured more than 60 cm and thus easily reached the sediment of the experimental basins (Table 3).

CO<sub>2</sub>-limited plants initially showed much lower growth rates than plants with ample CO<sub>2</sub>. This was illustrated by both lower increase in plant cover (P = 0.001; Table 1; Fig 1) and slower production of offsets (P = 0.021; Table 1; Fig 2). Plants from 90 μmol L<sup>-1</sup> CO<sub>2</sub> treatments grown with full PAR eventually reached similar coverage and number of offsets, but took much longer to reach these values than plants from 930 μmol L<sup>-1</sup> CO<sub>2</sub> availability.

Increase in plant cover (P = 0.012) and vegetative production (P = 0.016) were both also significantly influenced by interaction effects of CO<sub>2</sub> and PAR, since one factor enhanced the effects of the other. When plants received both high CO<sub>2</sub> concentrations and high PAR, the effects were stronger than the separate effects. Similarly, plants grown at the lowest CO<sub>2</sub> concentration and the lowest PAR did significantly worse than plants limited by only one of these factors.

The photosynthetic rates of the submerged plants were strongly influenced by both PAR (P<0.001; Table 1; Fig 3) and CO<sub>2</sub> availability (P = 0.003; Table 1; Fig 3). Reduced PAR led to a reduction of approximately 50% of O<sub>2</sub> production in plants from the 10% PAR treatment compared to those grown at full PAR. Furthermore, compared to plants grown at 930 μmol L<sup>-1</sup>, the photosynthetic rates of plants from the 90 μmol L<sup>-1</sup> and 230 μmol L<sup>-1</sup> CO<sub>2</sub> treatments were reduced by 55% and 40% respectively.

Maximum quantum yield values showed that all of the remaining plants had an active photosystem at the end of the experiment. These values were significantly higher for plants grown at 40% and 10% PAR than for those grown in the 100% PAR treatment (P = 0.002; Table 1; Table 3). Furthermore, plants that were grown under darker conditions contained significantly more chlorophyll a and b than those grown in full light (P = 0.009; Table 1; Table 3).



**Fig 1. Increase in cover of the water layer by *Stratiotes aloides* (mean + SEM) subjected to different PAR levels and CO<sub>2</sub> availability.** Low PAR significantly reduced final cover ( $P < 0.001$ ), whereas CO<sub>2</sub> limitation resulted in slower colonisation rates ( $P = 0.001$ ). Results of statistical tests are presented in [Table 1](#).

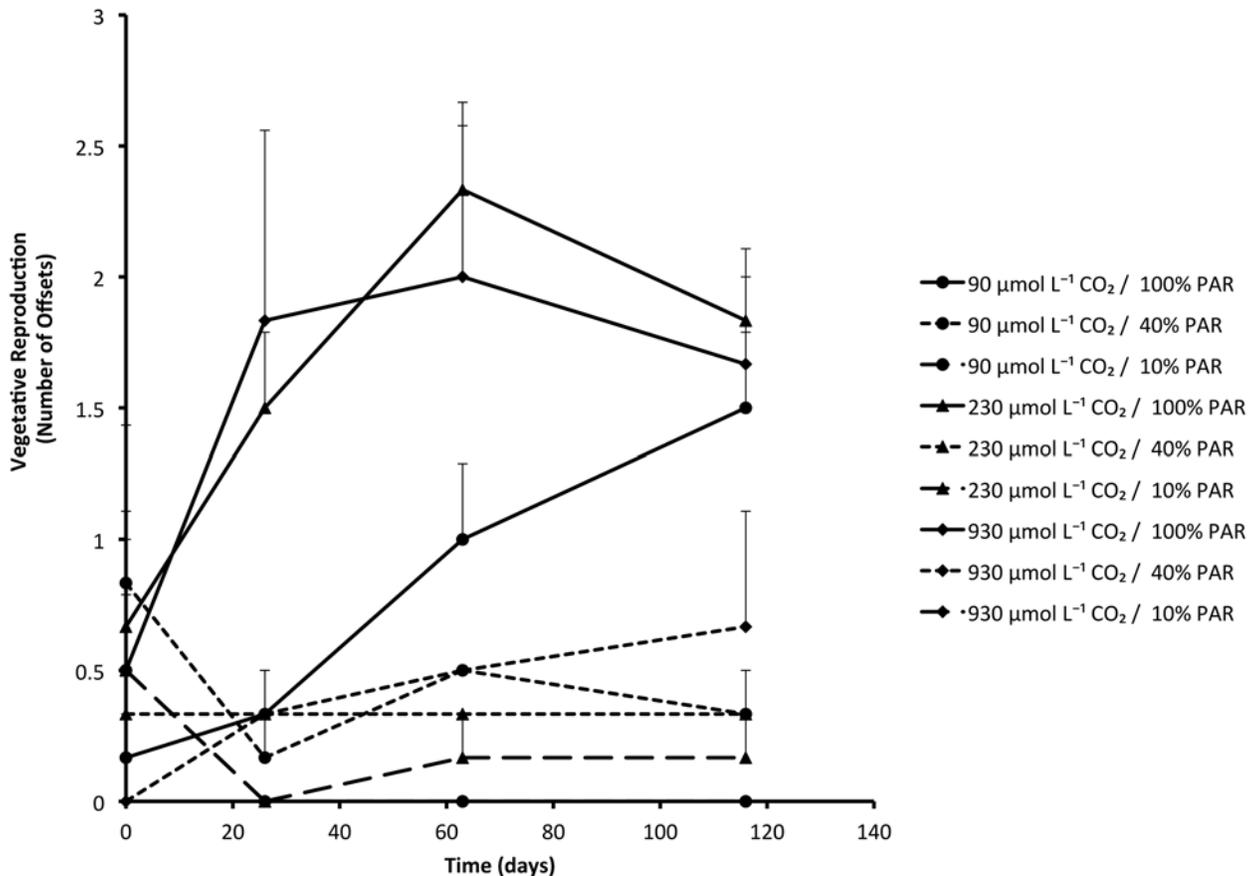
doi:10.1371/journal.pone.0124026.g001

After harvest, the C: N ratios of the plants grown at 40% and 10% PAR were lower than those of the plants grown in full PAR ( $P = 0.010$ ; [Table 1](#); [Table 4](#)). Furthermore, plants grown at 930  $\mu\text{mol L}^{-1}$  CO<sub>2</sub> had significantly higher N contents than those of lower CO<sub>2</sub> availability, which also led to lower C: N ratios. C: P and N: P ratios did not differ among treatments ([Table 4](#)). Plants also showed lower Fe contents ( $P = 0.012$ ; [Table 1](#); [Table 4](#)) when CO<sub>2</sub> concentrations were 230  $\mu\text{mol L}^{-1}$  or lower, while K contents did not differ among treatments, with all plants containing around  $630 \pm 50 \text{ mmol K kg DW}^{-1}$  ([Table 4](#); mean  $\pm$  SEM).

### Buoyancy and calcification

Under full PAR, formation of emergent leaves started in the 930  $\mu\text{mol L}^{-1}$  CO<sub>2</sub> treatments after approximately 6 weeks ([Table 3](#)). Under limited CO<sub>2</sub> availability, formation of emergent leaves was delayed by another 3 and 9 weeks for the 230  $\mu\text{mol L}^{-1}$  and 90  $\mu\text{mol L}^{-1}$  CO<sub>2</sub> treatments respectively. When PAR was limited, plants did not form emergent leaves. Submerged leaves were significantly thinner than emergent leaves ( $P = 0.002$ ; [Table 1](#); [Fig 4](#)).

Still, even in the absence of emergent leaves, PAR-limited plants were often observed to float just below the water surface and buoyancy depth of the rosette did not differ between light regimes for plants grown at 230  $\mu\text{mol L}^{-1}$  and 930  $\mu\text{mol L}^{-1}$  CO<sub>2</sub>. However, at 100%

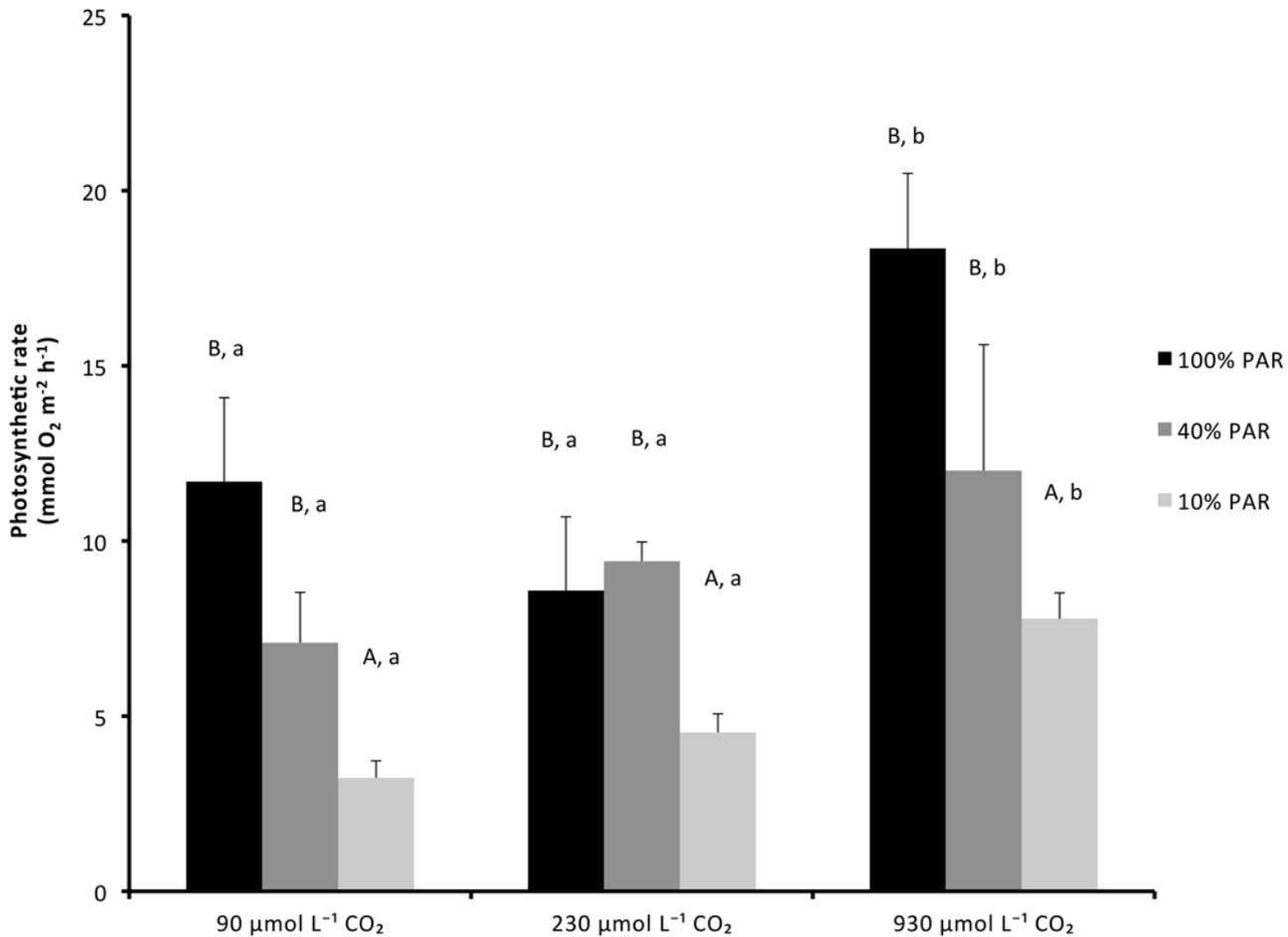


**Fig 2. Production of offsets through vegetative reproduction in *Stratiotes aloides* (mean + SEM) subjected to different PAR levels and CO<sub>2</sub> availabilities.** Low PAR significantly reduced the ability to produce offsets ( $P < 0.001$ ), whereas CO<sub>2</sub> limitation resulted in slower production rates ( $P = 0.021$ ). Results of statistical tests are presented in Table 1.

doi:10.1371/journal.pone.0124026.g002

light, plants that received the lowest amount of CO<sub>2</sub> were lying lower in the water layer than plants with 230 or 930 μmol L<sup>-1</sup> during the entire experimental period (CO<sub>2</sub>\*PAR effect;  $P = 0.005$ ; Table 1; Fig 5). Furthermore, plants from 230 μmol L<sup>-1</sup> CO<sub>2</sub> treatments initially also stayed lower in the water layer ( $P = 0.017$ ; Table 1; Fig 5) than plants treated with 930 μmol L<sup>-1</sup> CO<sub>2</sub>.

Observations of submerged leaves showed significant accumulation of, what proved to be, CaCO<sub>3</sub> on the upper side of the leaves. While this accumulation was most apparent on plants from the 90 μmol L<sup>-1</sup> CO<sub>2</sub> and 100% PAR treatment, eventually all plants showed some CaCO<sub>3</sub> precipitation on their leaves. This calcification of *S. aloides* leaves first became visible after approximately 3–4 weeks. Both high PAR ( $P = 0.002$ ; Table 1) and low CO<sub>2</sub> availability ( $P = 0.016$ ; Table 1) significantly increased the amount of CaCO<sub>3</sub> that accumulated on the leaves (Fig 6). The increase in CaCO<sub>3</sub> was highest during the first month of the experiment and stabilised after that to values of around  $40 \pm 7$  g CaCO<sub>3</sub> m<sup>-2</sup> (data not shown). Accumulation of CaCO<sub>3</sub> during the experiment led to an increase in leaf density of  $5.7 \pm 0.9\%$  in full PAR, whereas those in 40% and 10% PAR increased  $3.2 \pm 1.0\%$  and  $1.8 \pm 0.4\%$  respectively (data not shown).



**Fig 3. Photosynthetic oxygen production (mean + SEM) in *Stratiotes aloides* under different PAR and CO<sub>2</sub> availabilities.** Significant differences between different PAR and CO<sub>2</sub> treatments are indicated with capital and lower cased letters, respectively (P<0.01). Additional results of statistical tests are presented in [Table 1](#).

doi:10.1371/journal.pone.0124026.g003

## Discussion

### Buoyancy and photosynthesis

We show here that both PAR and the availability of dissolved CO<sub>2</sub> in the water layer strongly influence photosynthetic rates, buoyancy and formation of emergent leaves for *Stratiotes aloides*. A synthesis of the most important effects of limited CO<sub>2</sub> or PAR on these plants is presented in [Fig 7](#).

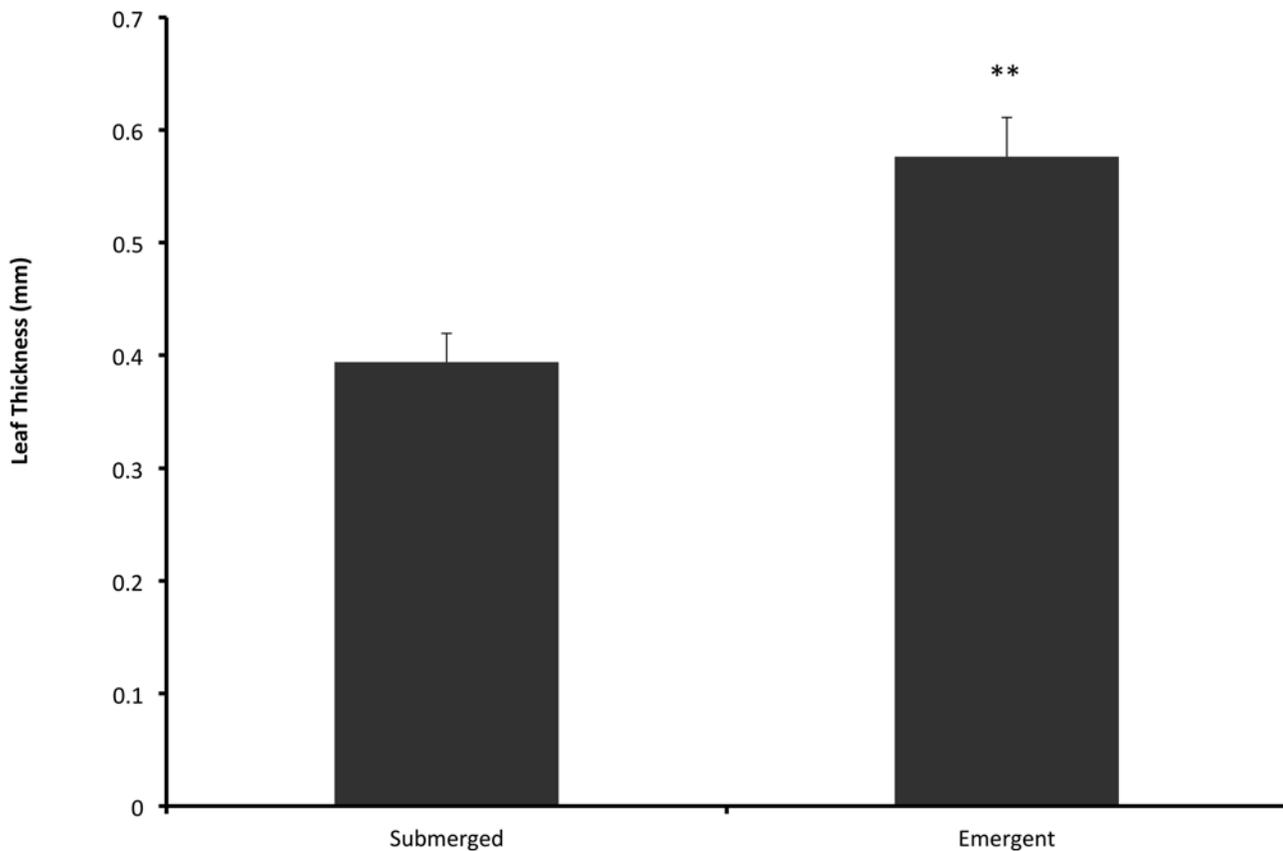
Emergent leaves did not appear at all when PAR was limiting, whereas their formation was delayed by 3–9 weeks under CO<sub>2</sub> limited conditions. All plants were firmly rooted in the sediment and produced roots that were long enough for the plants to become buoyant. Root growth, therefore, was not a constraint on reaching the water surface. PAR-limited plants, however, never became emergent but they floated just below the water surface ([Fig 7](#)). We observed that these plants show several adaptations to low light availability, including thinner leaves, a higher chlorophyll content and a higher efficiency of photosystem II. Still, the photosynthetic rate is too low to produce emergent leaves and form a dense floating mat under PAR-limiting conditions and several plants died or shed their roots during the experiment, resulting, on average, in shorter roots. We observed that PAR limited plants, which happened to have shed all of their

**Table 4. Plant chemical composition (mean ± SEM) of *Stratiotes aloides* after 4 months of different PAR and CO<sub>2</sub> treatments.**

| Characteristic | Unit                     | 100% PAR                                |  |  | 40% PAR                                 |  |  | 10% PAR                                 |  |  |
|----------------|--------------------------|---|--|--|---|--|--|---|--|--|
|                |                          | 90 μmol L <sup>-1</sup> CO <sub>2</sub> | 230 μmol L <sup>-1</sup> CO <sub>2</sub> | 930 μmol L <sup>-1</sup> CO <sub>2</sub> | 90 μmol L <sup>-1</sup> CO <sub>2</sub> | 230 μmol L <sup>-1</sup> CO <sub>2</sub> | 930 μmol L <sup>-1</sup> CO <sub>2</sub> | 90 μmol L <sup>-1</sup> CO <sub>2</sub> | 230 μmol L <sup>-1</sup> CO <sub>2</sub> | 930 μmol L <sup>-1</sup> CO <sub>2</sub> |
| C              | mol kg DW <sup>-1</sup>  | <sup>a,b</sup> 21.77±1.76 <sup>A</sup>  | <sup>a</sup> 22.53±1.92 <sup>A</sup>     | <sup>a,b</sup> 24.06±2.51 <sup>A</sup>   | <sup>a,b</sup> 23.10±1.47 <sup>A</sup>  | <sup>a</sup> 21.63±1.24 <sup>A</sup>     | <sup>b</sup> 24.47±1.12 <sup>A</sup>     | <sup>a,b</sup> 25.85±0.24 <sup>B</sup>  | <sup>a</sup> 24.24±2.07 <sup>B</sup>     | <sup>b</sup> 25.44±2.44 <sup>B</sup>     |
| N              | mmol kg DW <sup>-1</sup> | <sup>a</sup> 788±269 <sup>A</sup>       | <sup>a</sup> 947±299 <sup>A</sup>        | <sup>b</sup> 1066±390 <sup>A</sup>       | <sup>a</sup> 909±366 <sup>B</sup>       | <sup>a</sup> 708±107 <sup>B</sup>        | <sup>b</sup> 914±39 <sup>B</sup>         | <sup>a</sup> 1204±394 <sup>C</sup>      | <sup>a</sup> 958±295 <sup>C</sup>        | <sup>b</sup> 1338±311 <sup>C</sup>       |
| P              | mmol kg DW <sup>-1</sup> | 36.09±2.82 <sup>A</sup>                 | 25.54±7.13 <sup>A</sup>                  | 36.42±4.71 <sup>A</sup>                  | 38.65±2.98 <sup>A,B</sup>               | 38.28±3.76 <sup>A,B</sup>                | 41.06±9.64 <sup>A,B</sup>                | 47.66±3.80 <sup>B</sup>                 | 57.39±14.75 <sup>B</sup>                 | 55.34±11.13 <sup>B</sup>                 |
| K              | mmol kg DW <sup>-1</sup> | 675±54                                  | 412±113                                  | 567±123                                  | 819±186                                 | 618±145                                  | 566±166                                  | 520±99                                  | 814±167                                  | 743±250                                  |
| Fe             | mmol kg DW <sup>-1</sup> | <sup>a</sup> 3.50±1.39                  | <sup>a</sup> 4.34±0.11                   | <sup>b</sup> 9.27±3.05                   | <sup>a</sup> 2.34±0.37                  | <sup>a</sup> 5.41±0.85                   | <sup>b</sup> 11.29±1.91                  | <sup>a</sup> 7.01±2.62                  | <sup>a</sup> 6.11±2.44                   | <sup>b</sup> 7.84±2.86                   |
| C: N           | g g <sup>-1</sup>        | <sup>a,b</sup> 33.70±4.90 <sup>B</sup>  | <sup>b</sup> 42.79±0.65 <sup>B</sup>     | <sup>a</sup> 28.52±3.29 <sup>B</sup>     | <sup>a,b</sup> 25.62±5.45 <sup>A</sup>  | <sup>b</sup> 27.19±3.90 <sup>A</sup>     | <sup>a</sup> 16.90±1.93 <sup>A</sup>     | <sup>a,b</sup> 19.85±7.58 <sup>A</sup>  | <sup>b</sup> 14.80±1.17 <sup>A</sup>     | <sup>a</sup> 13.80±0.25 <sup>A</sup>     |
| C: P           | g g <sup>-1</sup>        | 929±96                                  | 1453±252                                 | 1052±126                                 | 898±55                                  | 850±81                                   | 1152±126                                 | 862±58                                  | 669±45                                   | 874±135                                  |
| N: P           | g g <sup>-1</sup>        | 30.06±5.55                              | 33.92±11.59                              | 38.78±6.46                               | 37.75±2.31                              | 33.47±1.83                               | 72.58±4.53                               | 54.50±6.89                              | 44.78±2.90                               | 64.02±7.38                               |

Significant differences among PAR treatments are indicated on the right by different capital letters (A, B, C) while significant differences among CO<sub>2</sub> treatments are indicated on the left by different lower case letters (a, b, c). See Table 1 for P and F values. Please note a different unit for C-content (mol kg DW<sup>-1</sup>).

doi:10.1371/journal.pone.0124026.t004

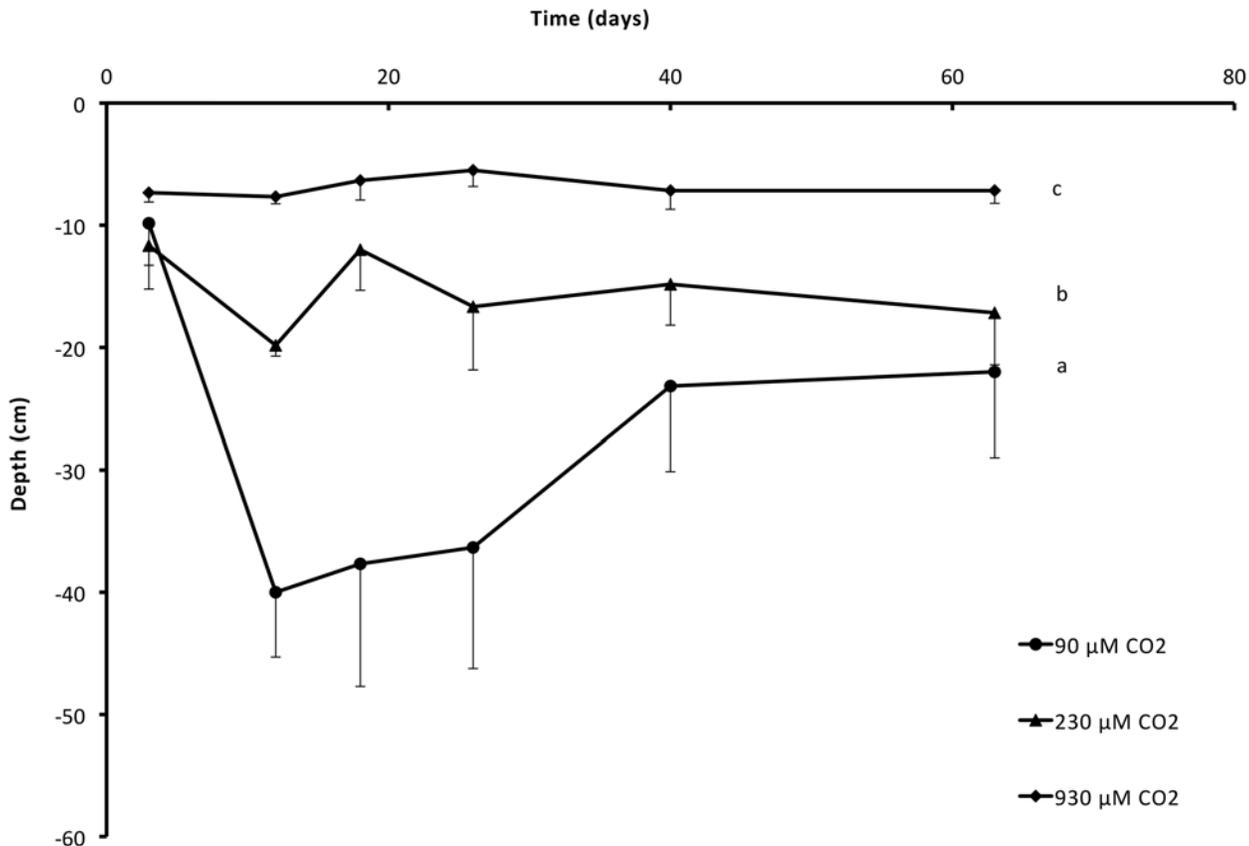


**Fig 4. Thickness of submerged and emergent leaves (mean + SEM) formed by *Stratiotes aloides* at the end of the four-month experimental treatments.** Submerged leaves were mainly produced by PAR-limited plants and plants from  $90 \mu\text{mol L}^{-1}$  treatments, whereas emergent leaves were only produced by plants grown at 100% PAR. Emergent leaves were significantly thicker than submerged leaves ( $P = 0.002$ ). Average values of leaf thickness per treatment are presented in Table 3, with statistical details in Table 1.

doi:10.1371/journal.pone.0124026.g004

roots still did not become buoyant. This further indicates that plants were not physically retained under water due to a limited root length. In the field, low PAR may well be caused by blooms of algae or cyanobacteria, high concentrations of humic acids or turbidity by particulate organic matter, all of which occur frequently in peatlands where *S. aloides* occurs [2].

With full PAR, however,  $\text{CO}_2$  limitation can still cause reduced buoyancy (Fig 7). This can be explained by the lower photosynthetic rates of these plants and their subsequently delayed formation of emergent leaves. Even though both light and DIC availability were high enough for photosynthesis in this treatment, the lower photosynthetic rate indicates C-limitation for these plants. Emergent leaves are thicker and have a larger volume of gas-filled intercellular spaces (aerenchyma) than submerged leaves, and are able to provide buoyancy to the plants [11, 18, 31]. The construction costs, in terms of energy and carbon, involved in the formation of the structurally more complex, thicker emergent leaves are much higher [32] and this C allocation is only possible if photosynthesis is sufficiently high. Even though *S. aloides* is capable of using  $\text{HCO}_3^-$  as an alternative C-source [17, 27], photosynthetic rates are lower due to the higher costs associated with  $\text{HCO}_3^-$  use [33]. This would explain why, under natural conditions, this species strongly prefers systems with an organic sediment, where high microbial respiration rates ensure high  $\text{CO}_2$  concentrations in the water layer [9].



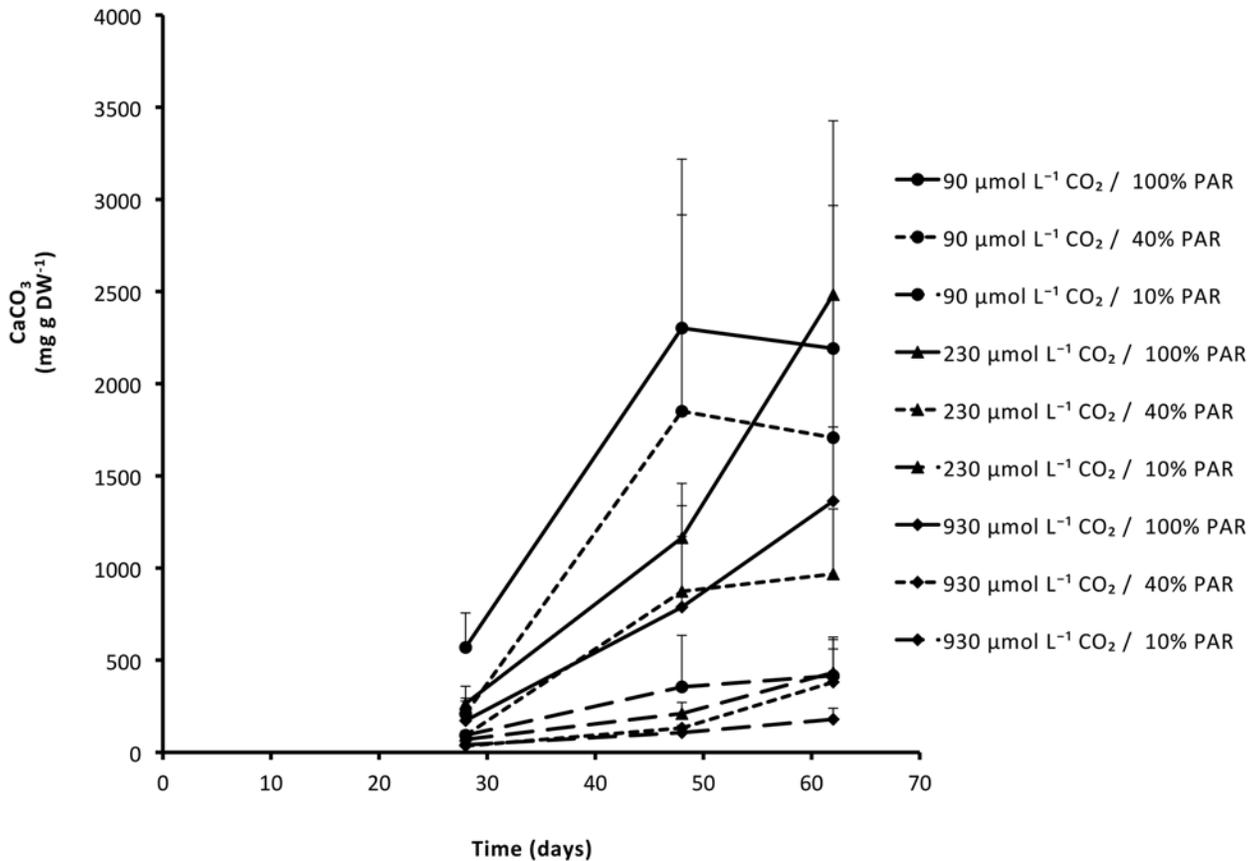
**Fig 5. Depths of the rosettes (mean and SEM) of *Stratiotes aloides*, grown at different CO<sub>2</sub> availabilities and 100% PAR.** Plants from 90 μmol L<sup>-1</sup> CO<sub>2</sub> treatments sank within two weeks, after which plants remained significantly lower in the water layer than those grown at limited 230 and 930 μmol L<sup>-1</sup> ( $P < 0.001$ ), as indicated by different letters. Additional results of statistical tests are presented in Table 1.

doi:10.1371/journal.pone.0124026.g005

Furthermore, as we have shown, HCO<sub>3</sub><sup>-</sup>-based photosynthesis induces accumulation of CaCO<sub>3</sub> on plant leaves. Calcification of aquatic plants is a common phenomenon occurring in species that use HCO<sub>3</sub><sup>-</sup> [34]. When aquatic plants take up HCO<sub>3</sub><sup>-</sup>, it is converted to CO<sub>2</sub> and OH<sup>-</sup>. While CO<sub>2</sub> is used for photosynthesis, OH<sup>-</sup> is excreted on the upper side of the leaf, thereby locally increasing the pH of the boundary layer and causing precipitation of CaCO<sub>3</sub> [34, 35]. Accumulation of CaCO<sub>3</sub> on *S. aloides* leaves has been discovered in the last century, when Montesantos [36] argued that CaCO<sub>3</sub> accumulation could explain the natural sinking of the species in autumn. Even though we could not find a direct link between CaCO<sub>3</sub> accumulation and limited buoyancy, we established that HCO<sub>3</sub><sup>-</sup> use increases the density of plant tissue in *S. aloides*.

### Biomass production and plant nutrition

Buoyancy is the key factor determining the function of *S. aloides* as a facilitator for wetland biodiversity. However, to create dense vegetation mats that can facilitate other species, the growth rate and production of sufficient new biomass are equally important (Fig 7). Plants grown at low PAR produced significantly lower amounts of new biomass than plants with high PAR. Plants were generally smaller and produced fewer offsets than those grown at full PAR availability (Fig 7). Since both PAR and CO<sub>2</sub> influence photosynthesis, it is not surprising that the effects of CO<sub>2</sub> or PAR limitation interact with each other. While plants grown at full PAR



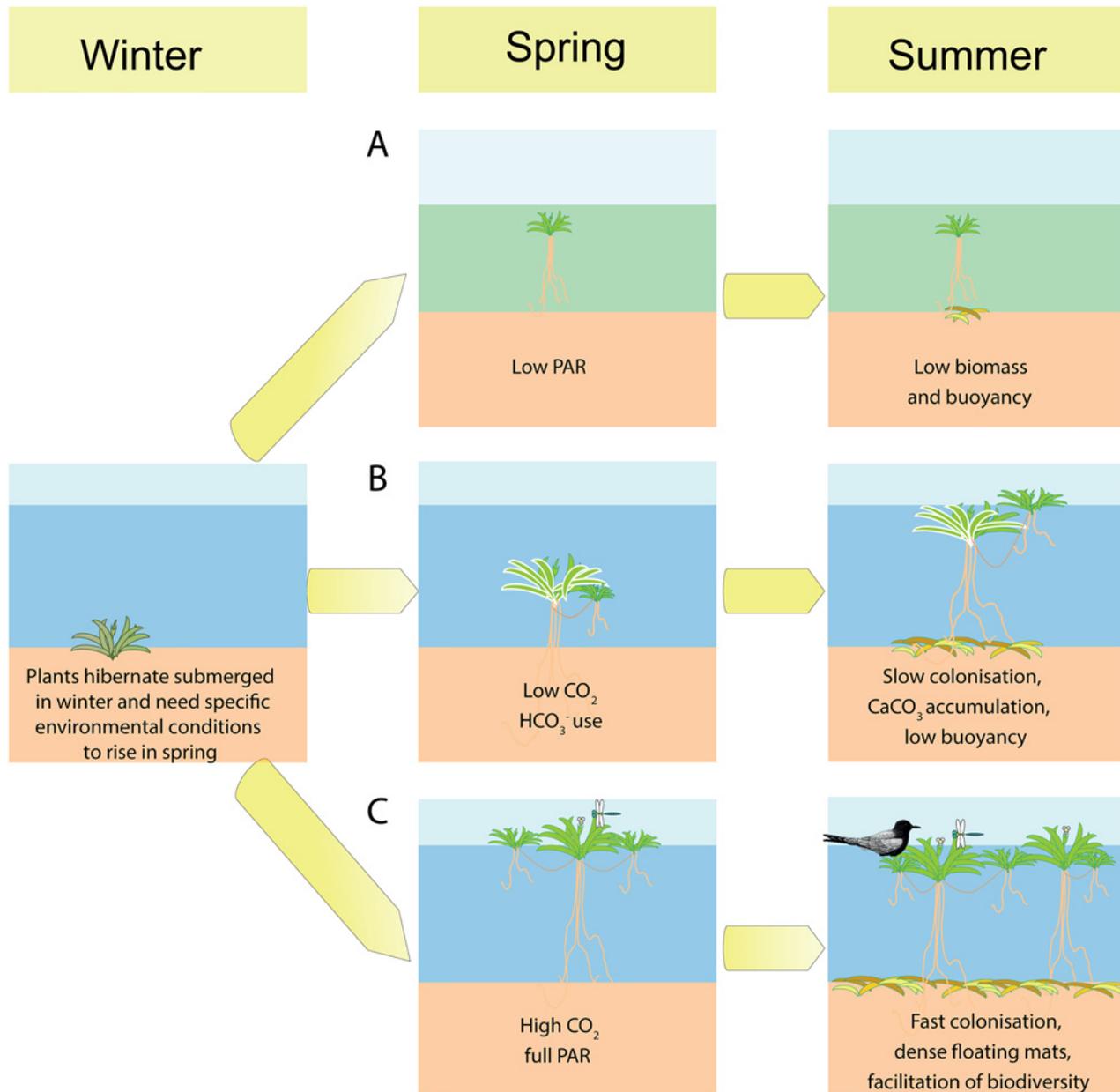
**Fig 6. Accumulation of CaCO<sub>3</sub> (mean + SEM) on leaves of *Stratiotes aloides* under different PAR and CO<sub>2</sub> availabilities.** CaCO<sub>3</sub> was measured from the moment the precipitation became visible (3–4 weeks after start of treatments). Low PAR significantly lowered CaCO<sub>3</sub> accumulation ( $P = 0.002$ ), while limited availability of CO<sub>2</sub> significantly increased the amount of CaCO<sub>3</sub> on the leaves ( $P = 0.016$ ). Additional results of statistical tests are presented in [Table 1](#).

doi:10.1371/journal.pone.0124026.g006

performed best when grown at high CO<sub>2</sub> concentrations, PAR-limited plant propagation did not respond to variations in CO<sub>2</sub> availability. Plants that were limited in CO<sub>2</sub>, but received sufficient PAR for photosynthesis, used HCO<sub>3</sub><sup>-</sup> as an alternative C-source. This not only reduced buoyancy, but also resulted in slower growth and reproductive rates (Fig 7). In wetlands, slower growth rates will easily result in plants being outcompeted by more efficient HCO<sub>3</sub><sup>-</sup> using species, including both plants and algae.

While growth rates were lower under CO<sub>2</sub> limited conditions, there were no significant differences between the final biomasses of the lowest and highest CO<sub>2</sub> treatments. This can most likely be explained by space restrictions in plants grown at high CO<sub>2</sub>, since these plants already covered over 90% of the basin after 60 days. At this point, space would limit recruitment of new offsets due to self-shading and other intra-specific competition factors.

Even though NH<sub>4</sub><sup>+</sup> availability dropped in sediments of plants with high CO<sub>2</sub> and high PAR, N limitation can be ruled out, since N: P ratio's of all plants were higher than 30 g g<sup>-1</sup> which suggests that P, rather than N, was limiting biomass production in all treatments [37]. Furthermore, even though tissue of CO<sub>2</sub> limited plants had lower Fe contents than those grown with ample CO<sub>2</sub>, all concentrations were within the range of healthy *S. aloides* vegetation, as was K, and Fe or K limitation can thus be ruled out [38, 39].



**Fig 7. Schematic overview of the effects of limited PAR or CO<sub>2</sub> availability in spring on the growth rate and buoyancy of *Stratiotes aloides*.** When PAR is limited, plants produce less biomass and do not form emergent leaves. Under limited CO<sub>2</sub> conditions, plants use HCO<sub>3</sub><sup>-</sup> for photosynthesis (when available), which is less efficient and causes lower growth rates, slower formation of emergent leaves and reduced buoyancy. Therefore, only plants receiving ample light and CO<sub>2</sub> can become buoyant and reach the required growth rate necessary to fulfil their role as a facilitator for biodiversity in wetlands.

doi:10.1371/journal.pone.0124026.g007

### Habitat quality and facilitator function

We have shown that one of the most important characteristics making *Stratiotes aloides* a facilitator species for biodiversity, buoyancy, can only be achieved when the plants are able to reach sufficiently high underwater photosynthetic rates in spring. High photosynthesis enables high colonisation rates and the formation of thicker leaves with intercellular spaces that can become filled with gas, most likely oxygen produced by photosynthesis. The resulting lower specific

weight of the plants makes them emerge to the surface. This process is strongly linked to PAR availability and to the species of C, rather than the total DIC concentration, in the aquatic environment. Therefore, contrary to many other facilitators, the potential of this widespread aquatic macrophyte to facilitate other species appears to depend on the prevalent environmental conditions, rather than the mere abundance of the species.

## Acknowledgments

The authors would like to thank Jelle Eygensteyn for his help with the chemical analyses, Peter Crujisen for technical assistance during the set-up of the experiment and Tim ten Broeke and Germa Verheggen for practical assistance. Furthermore, we thank Gerard van der Weerden for the use of the greenhouse facilities. This experiment was part of a project financed by the Dutch Technology Foundation STW (PeatCap; project number 11264) and the municipality of Amsterdam.

## Author Contributions

Conceived and designed the experiments: SH AS AK JR LL. Performed the experiments: SH AK. Analyzed the data: SH AK. Wrote the paper: SH AS AK JR LL.

## References

1. Verhoeven JTA, Bobbink R (2001) Plant diversity of fen landscapes in The Netherlands. In: Gopal B., editor editors. Biodiversity in wetlands: assessment, function and conservation. Leiden: Blackhuys Publishers. pp. 65–87.
2. Lamers LPM, Smolders AJP, Roelofs JGM (2002) The restoration of fens in The Netherlands. *Hydrobiologia* 478: 107–130.
3. Roelofs JGM (1991) Inlet of alkaline river water into peaty lowlands: effects on water quality and *Stratiotes aloides* L. stands. *Aquatic Botany* 39: 267–293.
4. Suhonen J, Suutari E, Kaunisto KM, Krams I (2013) Patch area of macrophyte *Stratiotes aloides* as a critical resource for declining dragonfly *Aeshna viridis*. *Journal of Insect Conservation* 17: 393–398.
5. Rantala MJ, Ilmonen J, Koskimaki J, Suhonen J, Tynkkynen K (2004) The macrophyte, *Stratiotes aloides*, protects larvae of dragonfly *Aeshna viridis* against fish predation. *Aquatic Ecology* 38: 77–82.
6. Van der Winden J, Beintema AJ, Heemskerck L (2004) Habitat-related Black Tern (*Chlidonias niger*) breeding success in The Netherlands. *Ardea* 92: 53–61.
7. Strzałek M, Koperski P (2009) The *Stratiotes aloides* L. stand as a habitat in oxbow lake Bużysko. *Aquatic Botany* 90: 1–6.
8. Sugier P, Lorens B, Chmiel S, Turczyński M (2010) The influence of *Ceratophyllum demersum* L. and *Stratiotes aloides* L. on richness and diversity of aquatic vegetation in the lakes of mid-eastern Poland. *Hydrobiologia* 656: 43–53.
9. Smolders AJP, Lamers LPM, den Hartog C, Roelofs JGM (2003) Mechanisms involved in the decline of *Stratiotes aloides* L. in The Netherlands: sulphate as a key variable. *Hydrobiologia* 506: 603–610.
10. Smolders AJP, Roelofs JGM (1996) The roles of internal iron hydroxide precipitation, sulphide toxicity and oxidizing ability in the survival of *Stratiotes aloides* roots at different iron concentrations in sediment pore water. *New Phytologist* 133: 253–260.
11. Cook CDK, Urmi-König K (1983) A revision of the genus *Stratiotes* (Hydrocharitaceae). *Aquatic Botany* 16: 213–249.
12. Madsen JD, Chambers PA, James WF, Koch EW, Westlake DF (2001) The interaction between water movement, sediment dynamics and submersed macrophytes. *Hydrobiologia* 444: 71–84.
13. Mulderij G, Smolders AJP, Donk E (2006) Allelopathic effect of the aquatic macrophyte, *Stratiotes aloides*, on natural phytoplankton. *Freshwater Biology* 51: 554–561.
14. NRM-SA (2011) State Alert Weed: *Stratiotes aloides*. In: N. R. M. G. o. S. Australia, editor editors.
15. MNR (2013) Ontario's Invasive Aquatic Plant List. In: O. M. o. N. Resources, editor editors. *ontarioca/invasivespecies*.
16. NRCS (2014) State noxious weed lists for 46 states. In: U. S. D. o. A. N. R. C. Service, editor editors. <https://plants.usda.gov/java/invasiveOne>.

17. Nielsen LT, Borum J (2008) Why the free floating macrophyte *Stratiotes aloides* mainly grows in highly CO<sub>2</sub>-supersaturated waters. *Aquatic Botany* 89: 379–384.
18. Sculthorpe CD (1967) The biology of aquatic vascular plants. London, UK: Edward Arnold. 610 p.
19. Erixon G (1979) Population ecology of a *Stratiotes aloides* L. stand in a riverside lagoon in N Sweden. *Hydrobiologia* 67: 215–221.
20. Renman G (1989) Life histories of 2 clonal populations of *Stratiotes aloides* L. *Hydrobiologia* 185: 211–222.
21. Higler LWG (1977) Macrofauna-cenoses on *Stratiotes* plants in Dutch broads. Rijksinstituut voor Natuurbeheer.
22. Toma C (2006) Distribution and comparison of two morphological forms of water soldier (*Stratiotes aloides* L.): a case study on Lake Stosineckie Wielkie (Northwest Poland). *Biodiversity: Research and Conservation* 3–4: 251–257.
23. Bloemendaal FHJL, Roelofs JGM (1988) *Waterplanten en Waterkwaliteit*. Utrecht: Stichting Uitgeverij van de Koninklijke Natuurhistorische Vereniging.
24. Stumm W, Morgan JJ (1996) *Aquatic Chemistry*. In: Schnoor J. L. and Zehnder A., editors. *Current Contents/Agriculture Biology & Environmental Sciences*. New York, U.S.A.: John Wiley & Sons. pp. 157–163. doi: [10.7717/peerj.328](https://doi.org/10.7717/peerj.328) PMID: [24749007](https://pubmed.ncbi.nlm.nih.gov/24749007/)
25. Allen ED, Spence DHN (1981) The differential ability of aquatic plants to utilize the inorganic carbon supply in fresh waters. *New Phytologist* 87: 269–283.
26. Madsen TV, Maberly SC (1991) Diurnal-variation in light and carbon limitation of photosynthesis by 2 species of submerged freshwater macrophyte with a differential ability to use bicarbonate. *Freshwater Biology* 26: 175–187.
27. Prins HBA, De Guia MB (1986) Carbon source of the water soldier, *Stratiotes aloides* L. *Aquatic Botany* 26: 225–234.
28. Kamphake LJ, Hannah SA, Cohen JM (1967) Automated analysis for nitrate by hydrazine reduction. *Water Research* 1: 205–8.
29. Grasshof K, Johannse H (1972) New sensitive and direct method for automatic determination of ammonia in seawater. *Journal Du Conseil* 34: 516–521.
30. Lichtenthaler HK, Wellburn AR (1983) Determinations of total carotenoids and chlorophylls a and b of leaf extracts in different solvents. *Biochemical Society Transactions* 11: 519–520.
31. Efremov AN, Sviridenko BF (2012) Seasonal and spatial dynamics of *Stratiotes aloides* (Hydrocharitaceae) plants. *Botanica Serbica* 36: 59–62.
32. Milla R, Reich PB (2007) The scaling of leaf area and mass: the cost of light interception increases with leaf size. *Proceedings of the Royal Society B-Biological Sciences* 274: 2109–2114. PMID: [17591590](https://pubmed.ncbi.nlm.nih.gov/17591590/)
33. Madsen TV, Sand-Jensen K (1991) Photosynthetic carbon assimilation in aquatic macrophytes. *Aquatic Botany* 41: 5–40.
34. Borowitzka MA (1984) Calcification in aquatic plants. *Plant Cell and Environment* 7: 457–466.
35. Prins HBA, Snel JFH, Zanstra PE, Helder RJ (1982) The mechanism of bicarbonate assimilation by the polar leaves of *Potamogeton* and *Elodea*—CO<sub>2</sub> concentrations at the leaf surface. *Plant Cell and Environment* 5: 207–214.
36. Montesantos N (1913) Morphologische und biologische Untersuchungen über einige Hydrocharideen. *Flora* 105: 1–32.
37. Güsewell S (2004) N: P ratios in terrestrial plants: variation and functional significance. *New Phytologist* 164: 243–266.
38. Smolders A, Roelofs JGM (1993) Sulfate mediated iron limitation and eutrophication in aquatic ecosystems. *Aquatic Botany* 46: 247–253.
39. Marschner H (1995) *Mineral Nutrition in Higher Plants*. London: Academic Press Limited. 889 p.