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Application of Metal-Free Triazole Formation in the Synthesis of Cyclic RGD-DTPA Conjugates

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The tandem 1,3-dipolar cycloaddition-retro-Diels-Alder (tandem crDA) reaction is presented as a versatile method for metal-free chemoselective conjugation of a DTPA radiolabel to N- δ -azido-cyclo(-Arg-Gly-Asp-D-Phe-Orn-) via oxanorbornadiene derivatives. To this end, the behavior of several trifluoromethyl-substituted oxanorbornadiene derivatives in the 1,3-dipolar cycloaddition

was studied and optimized to give a clean and efficient method for bio-orthogonal ligation in an aqueous environment. After radioisotope treatment, the resulting ¹¹¹In-labeled c(RGD)-CF₃-triazole-DTPA conjugate was subjected to preliminary biological evaluation and showed high affinity for $\alpha_{\rm v}\beta_3$ (IC₅₀=192 nm) and favorable pharmacokinetics.

Introduction

Functionalized cyclic RGD-containing peptides are of significant diagnostic and therapeutic interest because of their versatile application in both tumor targeting and tumor imaging. The major underlying mechanism responsible for the interest in such peptides is based on the binding affinity of the Arg-Gly-Asp (RGD) motif to integrin receptors, that is, alpha-beta heterodimeric cell-surface receptors which are overexpressed on developing capillary cells.[1,2] The interaction of integrins with specific matrix ligands is fundamental to invasion and formation of tumor-induced angiogenesis^[3,4] and metastasis.^[5] Based on the fact that the RGD sequence serves as a recurring motif for cell attachment in a large number of adhesive extracellular matrix, blood, and cell-surface proteins, a variety of RGD mimetics have been prepared for binding to $\alpha_v \beta_3$ integrin of the endothelium and tumor cells, leading to inhibition of cell-matrix interaction, [6] interruption of signal transmission, [7] disturbed cell migration, [8,9] and regression [10,11] or apoptosis^[12,13] of tumor cells. Whereas linear and flexible RGD-containing peptides bind to a number of integrin receptor subtypes, including $\alpha IIb\beta_3$, $\alpha_v\beta_3$, and $\alpha_v\beta_5^{[14]}$ the constrained cyclic RGDcontaining peptides selectively bind to the $\alpha_v\beta_3$ subtype. [15] As a result, cyclic RGD derivatives have become an important therapeutic target for the diagnosis of various solid tumors. Various modifications have been introduced to cyclic RGD peptides to improve receptor binding affinity (for example, polyvalent constructs)[16,17,18] and pharmacokinetic modifiers (for example; polymer conjugates^[19,20] and glycosylated derivatives^[21,22]).

Cyclic pentapeptides are frequently used in tumor targeting^[23] and imaging^[24] as well as for stimulation of cell adhesion.^[25] Similarly, the ϵ -amino group of the lysine residue in *cyclo*(-RGDfK-) is frequently used for modification with ligands for radiolabeling (18 F, 125 I, 64 Cu, 99m Tc, 111 In) $^{[26]}$ or with other biologically relevant moieties. $^{[27]}$ However, conjugation strategies for chemoselective functionalization of cyclic RGD peptides

with radiolabels or bioactive compounds are limited. Conjugation can be achieved with typical peptide coupling reagents, [28] or with other methods such as thioester or thioether coupling, [29] bromoacetyl derivatization, [30] or oxime ligation, [31] but most of the typical conjugation procedures suffer from poor selectivity or low yield. The synthesis of triazole-linked cyclic RGD-conjugates by copper-catalyzed, azide-alkyne cycloaddition has also recently been described. [18,22] Nevertheless, there still exists a strong demand for a bioconjugation reaction with increased efficiency and chemoselectivity involving mutually reactive conjugation partners that are synthetically readily accessible.[18] We recently reported on a copper-free tandem 1,3dipolar cycloaddition-retro-Diels-Alder (tandem crDA) ligation, resulting in a stable [1,2,3]-triazole linkage. [32] The methodology was applied to ligate several bio(macro)molecules, under physiological conditions, including a linear RGD peptide. It appeared to us that the absence of copper during conjugation is particularly favorable for the introduction of ligands for radiolabeling such as DOTA or DTPA when metal-free conditions are crucial. For example, the tandem crDA paves the way for coupling of DTPA constructs to azido-functionalized cyclic RGD peptides resulting in conjugates for application in tumor targeting and imaging. In this paper, we wish to report on the efficient crDA-ligation of radiolabeled oxanorbornadiene-DTPA conjugates for functionalization of an N- δ -azido-derivative of

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cyclo(-Arg-Gly-Asp-p-Phe-Orn-). Additionally, the efficiency of the crDA approach was further enhanced and a preliminary biological evaluation of the radiolabeled DTPA-cRGD conjugates is described.

Results and Discussion

Synthesis

Our approach for the synthesis of the oxanorbornadienespacer-DTPA 6 is depicted in Scheme 1. Oxanorbornadiene 2 was prepared according to literature procedures^[32,33] and was coupled to a short amino-modified ethylene glycol spacer. The particular ethylene glycol spacer was chosen, both for distancing the bulky diethylenetriaminepentaacetic acid (DTPA) ligand from the reactive oxanorbornadiene, and to potentially improve the pharmacokinetic characteristics of the conjugate. [19b,34] Secondly, as a result of the amino-modification of the spacer, an amide-linked conjugate 3 was obtained and was projected to display greater stability over the earlier reported esters under physiological conditions. After Boc-deprotection of compound 3, the resulting free amine 4 was coupled to tetra-O-tert-butyl protected DTPA to give compound 5 in a reasonable yield (64%). Finally, the desired oxanorbornadiene-DTPA conjugate (oxanor-DTPA, 6) was obtained in a sluggish reaction (four days) in quantitative yield after removal of the tert-butyl groups with concentrated TFA in CH₂Cl₂. Next, our attention was focused on the synthesis of N-δ-azido-cyclo(-Arg-Gly-Asp-D-Phe-Orn-) 12 (Scheme 2). Haubner et al. [35] demonstrated that substitution of the valine residue in the parent peptide c(-RGDfV-) for a lysine residue did not affect the activity or selectivity, enabling this position for any anchoring without interference with biological activity.[36] Consequently, the valine residue was substituted for an azide-functionalized ornithine derivative. $N-\delta$ -azido-Fmoc-L-ornithine (8) was prepared by diazotransfer reaction on Fmoc-L-Orn-OH and N-α-Fmoc-Dphenylalanine (9) was synthesized by Fmoc-protection of H-D-

Phe-OH in an excellent yield (91%). The linear pentapeptide was synthesized using Fmoc-based solid-phase peptide synthesis (SPSS) on a trityl resin according to a modified literature procedure. (31) Cleavage of the protected peptide from the resin gave *N*-δ-azido-(Arg(PMC)-Gly-Asp(OtBu)-D-Phe-Orn-OH) (10) in an overall yield of 29%. Subsequent cyclization under the influence of diphenylphosphorylazide (DPPA) followed by crystallization resulted in the desired *N*-δ-azido-*cyclo*(-Arg(PMC)-Gly-Asp(OtBu)-D-Phe-Orn-) (11) in an excellent yield (87%). In the final step, *N*-δ-azido-*cyclo*(-Arg-Gly-Asp-D-Phe-Orn-) (12) was obtained by deprotection of PMC and *tert*-butyl protecting groups in concentrated trifluoroacetic acid (TFA). Characterization by ¹H NMR and LCMS analysis showed that the desired product was obtained with a purity of more than 97%.

Conjugation studies

The conjugation behavior of the cyclic RGD peptide with DTPA-functionalized oxanorbornadiene in the tandem crDA was studied by means of 1H NMR spectroscopy, LCMS analysis, and HPLC with radiolabel detection. Monitoring the reaction by NMR spectroscopy was done by dissolving oxanor-DTPA (6) in a mixture of D_2O and CD_3OD which was subsequently added to a slight excess of cyclic RGD peptide 12 (1.5 equiv), leading to a final concentration of 6.6 μ m. Under these conditions, 1,3-dipolar cycloaddition retro-Diels-Alder-reaction of 6 and 12 was expected to lead, via furan elimination, to triazole adduct 13 (Scheme 3). Compound 13, although depicted as a single compound in Scheme 3, was expected to be formed as a mixture of regioisomeric products A_1 and A_2 as a result of the lack of regioselectivity in the first step of the crDA process, that is, the 1,3-dipolar cycloaddition (Scheme 4).

The reaction mixture was placed in a 400 MHz NMR-apparatus and conversion was monitored by integration of the furan protons with respect to the disappearing bridgehead protons of the oxanorbornadiene. First of all, the reaction was found to proceed rather slowly, requiring four days for complete disap-

Scheme 1. a) and b) Literature procedures^[32] (60% over 2 steps); c) 1-*N*-Boc-3,6-dioxa-8-octane-1,8-diamine, EDC, DMAP, DMF (64%); d) TFA, CH₂Cl₂ (99%); e) tetra-*O-tert*-Bu-DTPA, EDC, DMAP, CH₂Cl₂ (81%); f) TFA, CH₂Cl₂ (99%); g) NH₄OAc buffer (pH 5.5), InCl₃.

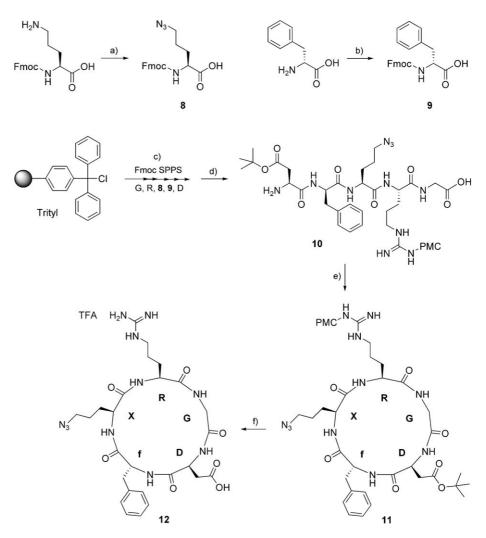
pearance of 6. After this time, LCMS analysis (UV detection of the RGD motif) of the reaction mixture (Figure 1 A) revealed two signals for the desired triazole-containing conjugate 13 as a mixture of two regioisomers $(55.5\%, A_1 \text{ and } A_2 \text{ ratio } 1:1.5),$ along with residual cyclic RGD (initially peptide 1.5 equiv, found; 34.2%, Figure 1B). Surprisingly, a third product was also observed (10.1%) and based on mass spectral analysis the structure was assigned to a 1*H*-1,2,3-triazole product **B** (R^2 = H, Scheme 4) resulting from undesired cycloaddition on the unsubstituted double bond of the oxanorbonadiene moiety, followed by CF₃-substituted furan elimination by retro-Diels-Alder reaction.

In addition to analysis by NMR spectroscopy and LCMS, HPLC analysis with radiolabel detection was performed to confirm complete substrate conversion. Monitoring the reaction by radiolabel detection required a radiolabeled DTPA-complex. ¹¹¹In was selected as a suitable radiolabel because of its long half-life (111 In, $t_{1/2}$ = 67.2 h) and its stability in the DTPA chelate. Thus, the oxanor-

DTPA conjugate (**6**) was labeled with ¹¹¹In by addition of approximately 100 μ Ci ¹¹¹InCl₃ to a solution of **6** in a metal-free NH₄OAc buffer (pH 5.5). The mixture was incubated for one hour at room temperature after which the ¹¹¹In-oxanor-DTPA complex (**7**, Scheme 1) was formed, concluded by a radiochemical purity check using an HPLC fitted with an in-line Nal detector (Figure 1 D). A total volume of 50 μ L of the ¹¹¹In-labeled oxanor-DTPA complex (ca. 10 μ Ci) was used to perform the tandem crDA reaction with *N*- δ -azido-*cyclo*(-Arg-Gly-Asp-D-Phe-Orn-) (**12**).

Figure 1 E clearly shows the conversion of 111 In-oxanor-DTPA (7) into the two isomers of c(RGD)-CF₃-triazole-DTPA conjugate 13 (marked with *). In correspondence with LCMS analysis, approximately 14% of the undesired 111 In-DTPA-labeled CF₃-furan formed concomitantly with product B was also detected.

In order to suppress cycloaddition on the unsubstituted side of the amide-oxanorbornadiene system, optimization of the tandem crDA process was undertaken. To this end, a small series of methyl-substitut-



Scheme 2. a) TfN₃, CuSO₄, H₂O (64%); b) FmocCl, DiPEA, 1,4-dioxane (91%); c) Fmoc-based SPPS; d) TFA, TIPS, H₂O (90:4.75:4.75) (29% over two steps); e) DPPA, NaHCO₃, DMF (87%); f) TFA, CH₂Cl₂ (93%).

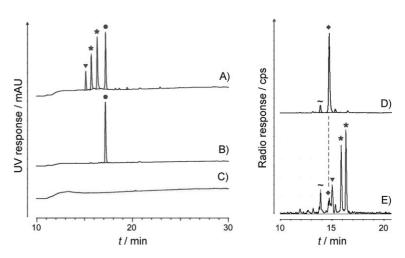


Figure 1. LCMS chromatograms; A = crude crDA reaction of 12 and 6 after four days: (∇) = isomer B (10.1%), (*) = isomer A₁ (21.3%), (*) = isomer A₂ (33.2%), and (\bullet) = residual c(RGD) 12 (34.2%). B = cRGD 12. C = oxanor-DTPA derivative 6. Radiolabel traces; D = (\bullet) ¹¹¹In-oxanor-DTPA complex (7). E = reaction mixture comprised of (~) = byproduct in starting material (\bullet) = 7, (∇) = ¹¹¹In-DTPA-labeled CF₃-furan formed concomitantly with product B, (*) = isomer A₁, (*) = isomer A₂.

ed oxanorbornadienes was synthesized (14–17, Scheme 5) and the reaction characteristics in the tandem crDA reaction evaluated by ¹H NMR spectroscopy using benzyl azide in CD₃OD as a model system (Table 1).

Reaction of ethyl 4,4,4-trifluorobutynoate with 3-methylfuran resulted in the formation of two regioisomers (14a and 14b) in good yield (72%). Saponification of 14a and 14b was followed by coupling of the resulting acids 15a and 15b to glycine methyl ester, leading to compounds 16a and 16b as a model system for the amidelinked glycol spacer. For comparison, the unsubstituted oxanorbornadiene glycine methyl ester derivative (18) was also synthesized. Apart from that, 1,4-dimethyloxanorbornadiene derivative 17 was prepared starting from 2,5-dimethylfuran.

The oxanorbornadiene derivatives (1 and 14-18) were reacted with benzyl azide in the tandem crDA reaction. Product formation in the tandem crDA reaction with benzyl azide (R4-N3) and the specification of R1-R3 is illustrated in Scheme 4. The kinetic data obtained from ¹H NMR spectroscopy are depicted in Table 1. The cycloaddition reaction of the ethyl ester oxanorbornadiene derivative 1 with benzyl azide gave clean conversion to the desired isomers A_1 and A_2 (Table 1, entry 1).[37] Subjecting the methyl-substituted oxanorbornadiene 14a/b to the tandem crDA reaction conditions (entry 2) resulted, apart from the expected cycloaddition product, in the partial formation oxaguadricyclanes (19 a/b, Scheme 6).[38] The unexpected oxaquadricyclane formation can be attributed to an intramolecular migration of the electron-rich methyl-substituted double bond to the accepting electron-deficient ester-substituted double bond.

Scheme 3. Conjugation experiment of N- δ -azido-cyclo-(Arg-Gly-Asp-D-Phe-Orn) 12 with oxanor-DTPA (6) or Meoxanor-DTPA (23) resulting in c(RGD)-CF $_3$ -triazole-DTPA conjugate 13 (major 1,4-isomer shown).

Scheme 4. Reaction pathways for the formation of triazole compounds A_1 , A_2 , and B in the tandem crDA reaction.

Scheme 5. a) Neat, 40 °C (14 a/b 72%); b) THF, 1 M NaOH, (15 a/b 80%); c) H-Gly-OMe, DMAP, EDC, CH₂Cl₂ (74% for 16 a/b; 56% for 18); d) 1,4-dioxane, reflux (91%).

Table 1. Products and kinetic data of reactions between oxanorbornadiene derivatives 1 and 14–18 and benzyl azide.										
	Compound	R^1	R^2	\mathbb{R}^3	Equiv N ₃	A [%]	$A_1:A_2$	B [%]	t _{1/2} [min]	Rate [$\times 10^4 \mathrm{m}^{-1} \mathrm{s}^{-1}$]
1	1	Н	Н	OEt	0.99	97	1:1.5	3	205	8.7 ± 0.09
2 ^[b]	14	Н	Me	OEt	1.20	n.d.	n.d.	n.d.	n.d.	n.d.
3	15	Н	Me	ОН	1.04	>99	1:1.3	trace	360	4.2 ± 0.18
4	17	Me	Н	OEt	1.39	94	1:1.2	6	490	2.6 ± 0.10
5	18	Н	Н	Gly-OMe	1.32	84	1:2.4	16	590	1.9 ± 0.03
6	16	Н	Me	Gly-OMe	1.08	97	1:2.1	3	> 900	1.5 ± 0.01

[a] At 25 °C and 100 mm in CD₃OD. Obtained by monitoring the reactions with ^{1}H NMR spectroscopy (400 MHz). R^{1} , R^{2} , and R^{3} as depicted in Scheme 4 (R^{4} = Bn). [b] Unstable, partial oxaquadricyclane formation. n.d. = not determined.

O CF₃
$$\xrightarrow{\text{BnN}_3}$$
 $\xrightarrow{\text{MeOD}}$ $\xrightarrow{\text{Since Color of C$

Scheme 6. Formation of oxa-quadricyclanes 19a/b from compound 14a/b upon subjection to tandem crDA reaction conditions.

Much to our satisfaction, upon performing the crDA reaction with the less electron-withdrawing carboxylic acid derivative **15 a/b** (entry 3), neither quadricyclane nor undesired isomer **B** formation was observed. The reaction rate, however, slightly

decreased compared to the ethyl ester oxanorbornadiene derivative 1. The oxanorbornadiene with methyl substituents on both bridgehead positions (i.e., compound 17) gave comparable amounts of isomer B with respect to compound 1 (entries 4 and 1, respectively). However, the rate of the reaction dropped by a factor of four, which suggests that substitution at the bridgehead position merely affects the rate of the tandem crDA reaction. Repeating the experiments with the desired, more stable, amide bond at the 3-position of the oxanorbornadiene (that is, compound 18) considerably increased the amount of undesired isomer B compared to compound 1 (16% and 3%, respectively). Apart from that, the rate of the cycloaddition reactions diminished considerably as a fivefold decrease in reaction rate was observed. A similar decrease in reaction rate was found for the amide containing Me-oxanorbornadiene (16, entry 6), though with the same order of magnitude as the unsubstituted oxanorbornadiene glycine conjugate 18 (1.5 and 1.9, respectively). Gratifyingly, almost full suppression of isomer

B formation was achieved with the glycine-substituted Me-oxanorbornadiene (16 a/b). Taking everything into consideration, the monomethyl-substituted oxanorbornadiene was considered the most suitable reaction partner for conjugation to *N*-δ-azido-*cyclo*(-Arg-Gly-Asp-p-Phe-Orn-) 12 as the reaction rate in the tandem crDA reaction is not negatively influenced, whereas formation of the undesired isomer B is effectively suppressed. To this end, a mixture of methyl-substituted oxanor-bornadiene-DTPA complex (Me-oxanor-DTPA, 23 a and 23 b)

was synthesized from compound **15a/b** in a four-step reaction sequence (Scheme 7). Radiolabeling of compound **23 a/b** gave the desired ¹¹¹In-Me-oxanor-DTPA complex (**24a** and **24b**) in the final step.

OCF₃ a)-c)
$$RO_2C$$
 N RO_2C RO_2R RO_2C $RO_$

Scheme 7. a) 1-N-Boc-3,6-dioxa-octane-1,8-diamine, EDC, DMAP, CH_2CI_2 (47% for **20 a/b**); b) TFA, CH_2CI_2 (99% for **21 a/b**); c) tetra-O-tert-Bu-DTPA, EDC, DMAP, CH_2CI_2 (44% for **22 a/b**); d) TFA, CH_2CI_2 (99%); e) NH_4OAc buffer (pH 5.5), $InCI_3$.

The unlabeled Me-oxanor-DTPA complex 23 a/b and N- δ -azido-cyclo(-Arg-Gly-Asp-D-Phe-Orn-) (12) were used in the tandem crDA reaction and monitored by 1 H NMR spectroscopy. Similar reaction conditions as mentioned before were applied to give nearly full conversion to conjugate 13 after five days at $37\,^{\circ}$ C. After lyophilization of the reaction mixture, the sample was subjected to LCMS analysis (Figure 2). Much to our satisfaction, only three peaks were observed, comprising the two isomers of c(RGD)-CF $_3$ -triazole-DTPA conjugate 13 (51.4%, A_1 and A_2 1:1.5, Scheme 4) along with residual cRGD peptide

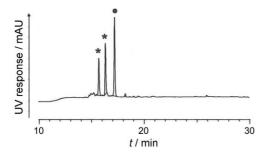


Figure 2. LCMS chromatogram of the crude crDA reaction of 23 and 12 after five days (*) = isomer A_1 (22.4%), (*) = isomer A_2 (29.0%), and (\bullet) = c(RGD) (12, 48.2%).

(48.2%, initially 1.5 equiv used) and no undesired 1-c(RGD)-1H-1,2,3-triazole (isomer **B**) could be detected.

Solid-phase $\alpha_{\nu}\beta_{3}$ binding assay

The affinities of the c(RGD)- CF_3 -triazole-DTPA conjugate **13** and DOTA-E-[c(RGDfK)]₂ (reference compound) for the $\alpha_v\beta_3$ integrin receptor were determined in a solid-phase competitive binding assay by using ¹¹¹In-DOTA-E-[c(RGDfK)]₂ as a tracer. ^[39,40]

Both ligands showed concentration-dependent inhibition of $^{111}\text{In-DOTA-E-}[c(\text{RGDfK})]_2$ binding to $\alpha_{\nu}\beta_3$, resulting in sigmoid curves. The IC $_{50}$ values found for DOTA-E-[c(RGDfK)] $_2$ and c-(RGD)-triazole-DTPA conjugate **13** were 125 nm and 192 nm, respectively. Comparing the IC $_{50}$ value of conjugate **13** with other triazole-containing monomeric or dimeric c(RGD)-conjugates, $^{[18,22]}$ similar results were obtained, indicating a limited effect on $\alpha_{\nu}\beta_3$ receptor binding for CF $_3$ -containing triazole-conjugates (Figure 3).

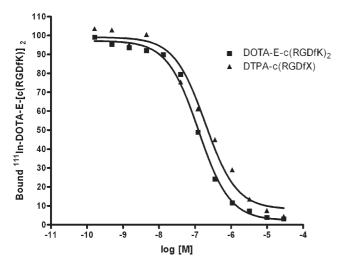


Figure 3. Competition of specific binding of ¹¹¹In-DOTA-E- $[c(RGDfK)]_2$ to $\alpha_v\beta_3$ with DOTA-E- $[c(RGDfK)]_2$, (\blacksquare) and DTPA-triazole-c(RGD) (13) (\blacktriangle).

Lipophilicity studies

The DTPA-triazole-c(RGD)-conjugate showed a log P value of -2.89 ± 0.38 obtained from n-octanol/water partition coeffi-

cient measurements, a number comparable to previously reported $\log P$ values for poly(ethylene glycol)^[19] or glucose^[21]_functionalised cyclic RGD derivatives.

Conclusions

Tandem cycloaddition-retro-Diels-Alder reaction of substituted oxanorbornadienes and functionalized azides is a powerful tool for constructing complex bioconjugates by forming stable triazole adducts. The 1,3-dipolar cycloaddition of DTPA-functionalized oxanorbornadiene and N-δ-azido-cyclo(-Arg-Gly-Asp-D-Phe-Orn-) resulted in the formation of the desired c(RGD)-CF₃-triazole-DTPA conjugate and a small amount of an unexpected 1-c(RGD)-1H-1,2,3-triazole. Introduction of methyl substituents on the oxanorbornadiene suppressed cycloaddition on the unfunctionalized side of the oxanorbornadiene systems, thereby effectively eliminating the formation of undesired 1-c-(RGD)-1H-1,2,3-triazole. Preliminary biological evaluation of the c(RGD)-CF₃-triazole-DTPA conjugate showed a good IC₅₀ value and favorable hydrophilicity that could result in positive pharmacokinetic behavior in vivo. A detailed biological evaluation of the c(RGD)-CF3-triazole-DTPA conjugate and further optimization of the new ligation method is currently under investigation in our laboratory.

Experimental Section

Instruments and methods: Unless otherwise stated, all chemicals were obtained from commercial sources and used without further purification. 111 InCl₃ was obtained from Tyco Mallinckrodt, Petten, The Netherlands. Analytical thin layer chromatography (TLC) was performed on Merck precoated silica gel 60 F-254 plates (layer thickness 0.25 mm) with visualization by ultraviolet (UV) irradiation at $\lambda = 254$ nm and/or $\lambda = 366$ nm and/or staining with KMnO₄. Preparative thin layer chromatography (Prep-TLC) was performed on Merck precoated silica gel 60 F-254 plates (layer thickness 1.00 mm) with concentration zone and visualization by UV irradiation at $\lambda = 254$ nm and/or $\lambda = 366$ nm. Purifications by silica gel chromatography were performed using Acros (0.035-0.070 mm, pore diameter ca. 6 nm) silica gel. Unless otherwise stated, all experiments were performed under ambient atmosphere and temperature. The water used in the biological procedures was deionised using a Labconco Water Pro PS purification system. THF was distilled under nitrogen from sodium/benzophenone. CH₂Cl₂ was distilled under nitrogen from CaH₂. FTIR spectra were recorded on an ATI Matson Genesis Series FTIR spectrometer fitted with an ATR cell. The vibrations (ν) are given in cm⁻¹. NMR spectra were recorded on Bruker DPX200 (200 MHz and 50 MHz for ¹H and ¹³C, respectively), Bruker DMX300 (300 MHz and 75 MHz for ^{1}H and ^{13}C , respectively) and Varian Inova 400 spectrometers. ¹H NMR chemical shifts are reported in parts per million (ppm) relative to a residual proton peak of the solvent, $\delta = 3.31$ for CD₃OD, $\delta = 7.26$ for CDCl₃, and $\delta = 4.79$ for D₂O. Broad peaks are indicated by the addition of br. Coupling constants are reported as a J value in Hertz (Hz). The number of protons (n) for a given resonance is indicated as nH, and is based on spectral integration values. ¹³C NMR chemical shifts are reported in ppm relative to CD₃OD (δ = 49.0) or CDCl₃ (δ = 77.0). Electrospray LC/MS analysis was performed on a Shimadzu LC/MS 2010A system, equipped with a Zorbax Extend C18 column, 3.5 um, 4.6×150 mm, Agilent Technologies, Palo Alto, CA, USA, eluting with a mobile phase gradient-profile: 0–5 min 10% acetonitrile/90% water (0.1% TFA), 5–30 min gradient to 95% acetonitrile/5% water (0.1% TFA), 30–40 min 95% acetonitrile/5% water (0.1% TFA). Matrix-assisted laser desorption/ionisation time-of-flight (MALDI-ToF) spectra were measured on a Bruker Biflex III spectrometer and samples were prepared from MeOH solutions using indoleacrylic acid (IAA) (20 mg mL⁻¹) as a matrix. LCQ/MS analysis was performed using Thermo scientific Advantage LCQ linear ion-trap electrospray (ESI-MS). Electrospray ionisation time-of-flight (ESI-ToF) spectra were measured with a JEOL AccuToF.

Radiolabeling experiments: The DTPA-linked oxanorbornadiene systems (6 and 23 a/b) were labeled with 111 ln by dissolving oxanor-DTPA 6 or Me-oxanor-DTPA 23a/b (5 μg, 7.0 nmol) in 5 μL H₂O. Subsequently, metal-free NH₄OAc buffer (90 μL, 0.25 м, pH 5.5) and 5 μL (~100 μCi) 111 lnCl₃ were added to each of the reaction mixtures. The mixtures were allowed to incubate for 1 h at RT after which the radiochemical purity was checked by HPLC (HP1100 series, LC system, Agilent Technologies, Palo Alto, CA, USA) using a RP-C18 column (5 μm, 4.6 mm×250 mm, Alltech, Deerfield, IL, USA) eluated with a gradient mobile phase (0–100% B over 20 min, solvent A=0.1% TFA in water, solvent B=0.1% TFA in acetonitrile) at 1 mL min $^{-1}$. The radioactivity of the eluates was monitored with an in-line Nal radiodetector (Raytest GmbH, Straubenhardt, Germany.)

Conjugation studies

General procedure for reactions between oxanorbornadiene derivatives and benzyl azide compounds monitored by ¹H NMR spectroscopy: A solution of an oxanorbornadiene derivative (0.05 mmol) in a deuterated solvent (0.5 mL) was added to a test tube containing benzyl azide (various equivalents). The mixture was briefly mixed with a vortex and then added to an NMR tube. Directly after the addition, the tube was placed in a Varion Inova 400 NMR apparatus at 25 °C and reaction was monitored following a preset measurement schedule.

General procedure for the synthesis of c(RGD)-CF₃-triazole-DTPA conjugate 13 via cycloaddition reaction between oxanorbornadiene derivatives and N-δ-azido-cyclo(-Arg-Gly-Asp-D-Phe-Orn-) (12) monitored by 1H NMR spectroscopy: A solution of oxanor-DTPA (6) or Me-oxanor-DTPA (23 a/b; 2.82 and 2.88 mg, respectively, 3.97 μmol) in CD₃OD/D₂O (0.5 mL 0.1 mL $^{-1}$) was added to a test tube containing cyclic RGD peptide 12 (4.3 mg, 5.96 μmol). The mixture was briefly mixed using a vortex and added to an NMR tube. The tube was then directly placed in a Varion Inova 400 NMR apparatus at 37 °C and the reaction was monitored following a preset measurement schedule. After completion of the reaction the resulted mixtures were lyophilized and analyzed by HRMS and electrospray LC/MS analysis performed on a Shimadzu LC/MS 2010A system; HRMS(ESI+): m/z calcd for $C_{50}H_{74}F_3N_{16}O_{19}$: 1257.5112 [M+H] $^+$, found 1257.5235.

General procedure for cycloaddition reactions between ¹¹¹In-labeled oxanorbornadiene derivatives and N-δ-azido-cyclo(-Arg-Gly-Asp-D-Phe-Orn-) (12) monitored by HPLC with an in-line Nal radiodetector: All tandem crDA reactions were performed at $37\,^{\circ}$ C in a total volume of 50 μL with 0.5 μg (0.70×10^{-3} μmol) ¹¹¹In-labeled DTPA (10 μCi) unless described otherwise. An incubator was used to warm the reactions to $37\,^{\circ}$ C for four days. The conversion was checked by HPLC using a RP-C18 column eluted with a gradient mobile phase (0–100% B over 20 min, solvent A=0.1% TFA in water, solvent B=0.1% TFA in acetonitrile) at 1 mL min⁻¹. The radioactivity of the eluates was monitored by using an in-line Nal radiodetector.

Octanol/water partition coefficient: For the lipophilicity determination, approximately 70 000 cpm ¹¹¹In-DTPA-cRGD was diluted to a volume of 3 mL with phosphate-buffered saline (PBS) and an equal volume of *n*-octanol was added to obtain a binary phase system. After mixing the two layers vigorously for ten seconds and gently for another 2 min, the two layers were separated by centrifugation (500 G, 5 min). Three 250 µL samples were taken from each layer and their activity was measured in a 3^{II} well type Nal gamma counter (Wallac 1480-Wizard 3). The log *P* value was determined in two independent experiments.

Solid-phase $\alpha_{\nu}\beta_{3}$ binding assay: Affinity of the DTPA-cRGD conjugate (13) and the conventional DOTA-E-[cRGDfK)], for the $\alpha_v \beta_3$ integrin was determined using a solid-phase competitive binding assay using 111In-DOTA-E-[cRGDfK)]₂ as a tracer. Labeling was performed following the procedure described by Boerman et al.[41] 111 InCl₃ (90 μ Ci μ g⁻¹) was added to 5 μ g DOTA-E-[cRGDfK)]₂ dissolved in 100 μL metal free HEPES buffer (1 M, pH 6.0). The mixture was heated to 100 °C during 15 min after which the chemical purity was determined with HPLC. Microtiter 96-well vinyl assay plates (Corning B.V., Schiphol-Rijk, The Netherlands) were coated with a solution of purified human integrin $\alpha_v \beta_3$ (150 ng mL⁻¹) in Triton X-100 Formulation (Chemical International, Temecula, CA, USA) in coating buffer (25 mm Tris·HCl (pH 7.4), 150 mm NaCl, $1 \text{ mM } \text{CaCl}_2$, $0.5 \text{ mM } \text{MgCl}_2$ and $1 \text{ mM } \text{MnCl}_2$) for 18 h at $4 ^{\circ}\text{C}$ (100 μL per well). After washing the plate twice with binding buffer (coating buffer supplemented with 0.1% BSA (bovine serum albumin)), all wells were blocked with 200 µL blocking buffer (coating buffer supplemented with 1% BSA) during 3.5 h at RT. The wells were washed twice with binding buffer and subsequently incubated with 100 µL binding buffer containing 200 000 cpm of ¹¹¹In-DOTA-E-[cRGDfK)]₂ and appropriate dilutions of nonlabeled DOTA-E-[cRGDfK)]₂ or DTPA-cRGD conjugate dissolved in 20 μL binding buffer for 1 h at 37 °C. The competitive displacement study of the dilutions was performed in triplicate. Finally, the plate was washed three times, the wells were cut, and the radioactivity in each well was counted using a γ-counter (Wallac 1480-Wizard® 3, Perkin-Elmer, Boston, MA, USA). The IC₅₀ values of both the DOTA-E-[cRGDfK)]₂ and DTPA-cRGD conjugate were determined by nonlinear regression (GraphPad Pris 4.0 Software Package, San Diego, CA, USA).

Statistical analysis: The mean values for the solid phase binding assay and the lipophilicity studies are given \pm standard deviation (S.D.). Statistical analysis was performed by using a Welch's corrected unpaired t-test or one-way analysis of variance with GraphPad InStat software (version 4.00, GraphPad Software). The level of significance was set at p < 0.05.

Synthesis

3-Trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxylic acid (2): According to the literature procedure: [32] Ethyl 2-fluorobut-2-ynoate (1.00 g, 0.86 mL, 6.02 mmol) yielded 726 mg of 2 (60% over two steps) as a white solid.

Fmoc-Orn(N₃)-OH (8): Modified literature procedure: [42] CH₂Cl₂ (3 mL) and triflic anhydride (1.3 mL, 7.73 mmol) at 0 °C were added to a solution of sodium azide (1.00 g, 15.4 mmol) in water (3 mL), while the mixture was vigorously stirred. After being stirred for 2 h at 0 °C the phases were separated, and the aqueous phase was extracted with CH₂Cl₂ (5 mL). The combined organic phases were washed with saturated aqueous NaHCO₃-solution (5 mL). The resulting triflic azide solution was added dropwise to a solution of Fmoc-Orn-OH·HCI (1.00 g, 2.57 mmol), CuSO₄ (16 mg, 64 μmol) and NaHCO₃ (436 mg, 5.19 mmol) in water (9 mL) followed by the addi-

tion of methanol until the mixture became a suspension. The mixture was stirred for 4 h at RT. The organic solvents were removed in vacuo and the resulting suspension was acidified with 1 m aqueous HCl to pH 3 after which the precipitate was isolated by filtration and washed with water. The residue was dissolved in ether, dried over Na₂SO₄, and concentrated in vacuo. Purification was performed by flash chromatography (CH₂Cl₂/methanol/AcOH 98:2:0.3, 98:5:0.5) and the product was coevaporated with toluene (4×) yielding a white solid (621 mg, 64%). ¹H NMR (CD₃OD, 200 MHz) δ =7.83 (m, 2H), 7.72 (m, 2H), 7.24 (m, 4H), 4.41 (d, J=6.3 Hz, 2H), 4.29 (m, 1 H), 4.19 (m, 1 H), 4.29 (m, 1 H), 4.19 (m, 1 H), 3.39 (m, 1 H), 1.90 (m, 2 H), 1.75 (m, 2 H); LRMS (ESI-) m/z calcd for C₂₀H₁₉O₄N₄: 379.1 [M-H]⁻ found 378.9. FTIR ν_{max} film: 2109, 1718, 1630, 1238 cm⁻¹.

Fmoc-D-Phe-OH (9): According to the literature procedure: [43] H-D-Phe-OH (530 mg, 1.37 mmol) yielded 1.13 g (91%) Fmoc-D-Phe-OH as a white solid after purification by crystallization from EtOAc.

H-Asp(OtBu)-D-Phe-Orn(N₃)-Arg(PMC)-Gly-OH (10): Compound 9 was synthesized by standard solid-phase peptide synthesis (SPPS) using a modified literature procedure: [31] A solution of Fmoc-Gly-OH (0.174 g, 0.58 mmol) and DIPEA (235 μ L, 1.35 mmol) in dry CH₂Cl₂ (2.5 mL) was added to a suspension of trityl chloride resin (300 mg of 1.5 mmol g⁻¹ loaded resin) in dry CH₂Cl₂ (2.5 mL) and the mixture was shaken at RT for 2.5 h. MeOH (470 μ L) and DIPEA (94 μ L) were added and the mixture was shaken for another 15 min. The resin was washed with NMP (3×2.5 mL), CH₂Cl₂ (5×2.5 mL), and MeOH (3×2.5 mL) and dried in vacuo. The Fmoc-protected resin was suspended in a solution of 20% piperidine in NMP (2×20 mL) and shaken for 5 and 15 min. The resin was washed with NMP (6 \times 20 mL). A positive Kaiser test^[44] indicated completion of the peptide coupling reaction. After filtering and washing with DMF ($3 \times$ 20 mL), the next amino acid was coupled by adding a mixture of Fmoc-Arg(PMC)-OH (960 mg, 1.45 mmol, 2.5 equiv), TBTU (465 mg, 1.45 mmol, 2.5 equiv), HOBt (196 mg, 1.45 mmol, 2.5 equiv), and DIPEA (0.67 mL, 4.1 mmol, 7 equiv) dissolved in NMP (100 mL) to the resin. The reaction mixture was shaken at RT for 90 min and washed with NMP (6×20 mL). A negative Kaiser test indicated completion of the peptide coupling reaction. The deprotectioncoupling sequence was repeated with the following amino acids: Fmoc-Orn(N₃)-OH (551 mg, 1.45 mmol, 2.5 equiv), Fmoc-D-Phe-OH (561 mg, 1.45 mmol, 2.5 equiv), and Fmoc-Asp(OtBu)-OH (596 mg, 1.45 mmol, 2.5 equiv). For the coupling of Fmoc-D-Phe-OH 2,4,6collidine (1.9 mL, 14.5 mmol, 25 equiv) was used instead of DIPEA. The resin was washed with DCM (4×20 mL) and treated with a solution of TFA, H₂O, and triisopropylsilane (TIPS; 90:4.75:4.75; 20 mL) for 3×10 min. After removal of the resin by filtration, the filtrates were combined and stirred for another 2.5 h. The collected filtrates were evaporated, coevaporated with toluene, and lyophilized from tert-BuOH/H2O/dioxane yielding a white foam (126 mg, 29%). ¹H NMR (400 MHz, CD₃OD) δ = 7.33–7.23 (m, 5 H), 4.59 (t, J = 8.0 Hz, 1H), 4.39 (dd, J=9.5, 4.4 Hz, 1H), 4.19 (t, J=6.5 Hz, 1H), 4.00 (dd, J = 10.5, 3.9 Hz, 1 H), 3.85 + 3.62 (AB-system J = 17.0 Hz, 2H), 3.20-3.14 (m, 4H), 3.02 (d, J=8.0 Hz, 2H), 2.83-2.77 (m, 2H), 2.67 (t, J=6.7 Hz, 2 H), 2.56 (s, 3 H), 2.55 (s, 3 H), 2.10 (s, 3 H), 1.95-1.80 (m, 4H), 1.80-1.73 (m, 1H), 1.69-1.60 (m, 1H), 1.58-1.49 (m, 2H), 1.47 (s, 9H), 1.31 (s, 6H), 1.22-1.08 (m, 2H); LCMS (EI+) purity +97%, m/z calcd for C₄₄H₆₆N₁₁O₁₁S: 955.5 [M+H]⁺, found 956.6.

Cyclo-[Asp(OtBu)-D-Phe-Orn(N_3)-Arg(PMC)-Gly] (11): Diphenylphosphorylazide (68 μ L, 0.31 mmol) and NaHCO $_3$ (43.9 mg, 0.52 mmol) were added to a solution of the linear peptide 10 (100 mg, 0.10 mmol) in dry DMF (21 mL) and the mixture was stirred for 44 h at RT. The mixture was filtered, diluted with EtOAc (100 mL),

and washed with saturated aqueous NH₄Cl-solution (2×100 mL) and brine (2×100 mL). The organic phase was dried over Na₂SO₄ and concentrated. The material was stirred with Et₂O for 10 min and the white precipitate was collected by filtration, washed with Et₂O (3×), and dried, yielding a white solid (85 mg, 87%). ¹H NMR (400 MHz, CDCl₃/CD₃OD/D₂O) δ =7.28–7.18 (m, 5H), 4.70 (dd, J=8.2 Hz, 1 H), 4.55 (d, J=8.0 Hz, 1 H), 4.21 (brs, 1 H), 4.21 + 3.34 (AB-system, J=14.9 Hz, 2 H), 3.94 (dd, J=10.4 Hz, 1 H), 3.14–3.11 (m, 4H), 2.96 (d, J=8.0 Hz, 2 H), 2.74 (dd, J=16.1, 8.2 Hz, 1 H), 2.63 (t, J=6.9 Hz, 2 H), 2.52 (s, 3 H), 2.51 (dd, J=16.1, 8.2 Hz, 1 H), 2.50 (s, 3 H), 2.06 (s, 3 H), 1.85–1.77 (m, 3 H), 1.76–1.67 (m, 1 H), 1.64–1.55 (m, 1 H), 1.52–1.41 (m, 3 H), 1.39 (s, 9 H), 1.28 (s, 6 H) 1.21–1.12 (m, 2 H); LCMS (El+) purity +97%, m/z calcd for $C_{44}H_{64}N_{11}O_{10}S$: 937.5 $[M+H]^+$, found 938.7.

Cyclo-[Asp-D-Phe-Orn(N₃)-Arg-Gly] (12): The cyclic peptide 11 (84 mg, 90 μmol) was dissolved in a mixture of TFA/H₂O 95:5 (5 mL) and the mixture was stirred for 3 h. The solvent was removed and the residue was co-evaporated with toluene. The material was dissolved in H_2O (15 mL), extracted with EtOAc (2×10 mL), and the aqueous phase was evaporated to dryness. The product was lyophilized out of H₂O/dioxane, yielding a white solid (61 mg, 93%). ¹H NMR (400 MHz, D₂O): δ = 7.42–7.21 (m, 5H), 4.64 (dd, J = 10.0, 6.0 Hz, 1 H), 4.37 (dd, J=9.1, 5.6 Hz, 1 H), 4.21+3.52 (ABsystem, J=15.0 Hz, 2H), 3.92 (dd, J=10.4, 4.8 Hz, 1H), 3.25-3.15 (m, 4H), 3.09 (dd, J=13.1, 6.2 Hz, 1H), 3.00-2.90 (m, 2H), 2.75 (dd, J=16.8, 6.5 Hz, 1 H), 1.92–1.62 (m, 3 H), 1.59–1.47 (m, 3 H), 1.26– 1.16 (m, 2 H); ¹³C NMR (50 MHz, D₂O (dioxane residue used as reference) δ = 174.9, 174.8, 173.7, 173.3, 172.0, 171.9, 157.3, 136.6, 129.8 (2C), 129.5 (2C), 127.9, 67.19 (dioxane), 55.8, 55.6, 53.1, 50.7, 50.2, 44.1, 41.1, 37.5, 34.9, 28.1, 27.9, 25.2, 25.1; HRMS (ESI+): *m/z* calcd for $C_{26}H_{38}N_{11}O_7$ [M+H] $^+$ 616.2956, found 616.2907. FTIR ν_{max} film: 3274, 2098, 1640, 1126 cm⁻¹.

N-{Boc}-N'-{3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2carboxyl}-3,6-dioxaoctane-1,8-diamine (3): 4-(dimethylamino)-pyridine (DMAP, 41.1 mg, 0.34 mmol) was added to a solution of oxanorbornadiene 2 (34.7 mg, 0.17 mmol) and 1-N-Boc-3,6-dioxa-8octane-1,8-diamine (41.8 mg, 0.17 mmol) in CH₂Cl₂ (1.5 mL). The mixture was cooled to 0 °C and 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDC·HCl, 21 mg, 0.11 mmol) was added slowly. The mixture was stirred for 30 min at 0 °C and 16 h at RT. The reaction mixture was acidified with HCl (2 M) to a pH of 1-2 and extracted with CH₂Cl₂ (2×5 mL). The combined organic layers were dried over MgSO₄, concentrated in vacuo, and purified by preparative TLC (CH₂Cl₂/MeOH 9:1) resulting in compound 3 as a slightly yellow oil (47.8 mg, 64%). $R_F = 0.50$ (CH₂Cl₂/MeOH 9:1). 1 H NMR (400 MHz, CD $_{3}$ OD) δ = 7.30 (dd, J = 5.3, 1.9 Hz, 1 H), 7.21 (dd, J=5.3, 1.9 Hz, 1H), 5.65 (t, J=1.6, 1.6 Hz, 1H), 5.55 (m, 1H),3.59 (s, 4H) 3.56 (dd, J=10.5, 5.0 Hz, 2H), 3.49 (t, J=5.7, 5.7 Hz, 2H), 3.45 (dd, J = 11.3, 5.5 Hz, 2H), 3.20 (t, J = 5.7 Hz, 2H), 1.41 (s, 9H); 13 C NMR (50 MHz, CD₃OD): $\delta = 165.2$, 158.5, 144.7, 143.6, [126.6, 121.2] CF₃, 87.1, 84.4, 80.1, 71.3, 70.3, 41.2, 40.4, 28.8; LRMS (ESI+) m/z calcd for $C_{19}H_{28}F_3N_2O_6$: 437.1 $[M+H]^+$, found 437.1.

N,N'-{3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carbox-yl}-3,6-dioxaoctane-1,8-diamine-TFA (4): Trifluoroacetic acid (TFA, 0.5 mL, 2.85 mmol) was added dropwise to a cooled solution (0 °C) of **3** (47.8 mg, 0.11 mmol) in dry CH₂Cl₂ (2 mL). The reaction was stirred at 0 °C for 1 h after which the reaction was complete. The solvent was evaporated and the crude mixture was dissolved in H₂O (5 mL) and dioxane (5 mL) and freeze-dried to afford compound **4** as a light yellow solid (49.0 mg, +99%). $R_{\rm F}$ =0.09 (CH₂Cl₂/MeOH 9:1). ¹H NMR (400 MHz, CD₃OD): δ =7.30 (ddd, J=5.3, 1.9 Hz, 0.7 Hz, 1 H), 7.22 (ddd, J=5.3, 1.9, 0.7 Hz, 1 H), 5.66 (s, 1 H),

5.55 (s, 1 H), 3.76 (t, J = 4.9 Hz, 2 H) 3.64, (s, 4 H), 3.57 (t, J = 5.9 Hz, 2 H), 3.38–3.55 (m, 4 H), 3.09 (t, J = 5.7 Hz, 2 H); 13 C NMR (50 MHz, CD₃OD): δ = 165.3, 156.2, 144.6, 143.7, 127.8, [126.5, 121.2] CF₃, 87.1, 84.5, 71.3, 70.3, 67.9, 40.7, 40.3; HRMS (ESI +) m/z calcd for $C_{14}H_{20}F_3$ N_2O_4 : 337.1375 $[M+H]^+$, found 337.1376.

N-{2-O-tert-butyl-DTPA-acetamide},N'-{3-trifluoromethyl-7-oxa-bicyclo-[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine Compound 4 (49 mg, 0.11 mmol) was dissolved in dry CH_2CI_2 (2 mL), and 4-(dimethylamino)pyridine (DMAP, 26.6 mg, 0.22 mmol) and diethylenetriamine-N,N,N",N"-tetra-tert-butyl acetate-N'-acetic acid ((DTPA-tert-butyl ester) 67.8 mg, 0.11 mmol) were added. The mixture was cooled to 0°C followed by addition of 1-ethyl-3-(3dimethylaminopropyl) carbodiimide hydrochloride (EDC·HCl, 23.1 mg, 0.12 mmol). After being stirred for 1 h at 0 °C, the mixture was allowed to warm to RT and was stirred for an additional 4 h. The reaction mixture was quenched with 1 m HCl (2 mL) and the water layer was extracted with CH₂Cl₂ (2×2 mL). The combined organic layers was dried over Na₂SO₄ and subsequently evaporated. The crude product was purified by preparative TLC (MeOH/CH₂Cl₂ 1:9) to obtain pure product as a colorless oil (38 mg, 81%). ¹H NMR (400 MHz, CDCl₃): $\delta = 8.23$ (brs, 1 H, NH), 7.32 (dd, J = 5.3, 2.1 Hz, 1H), 7.13 (dd, J=5.3, 2.1 Hz, 1H) 6.81 (brs, 1H, NH), 5.62 (dd, J=1.4, 0.5 Hz, 2H), 3.62-3.50 (m, 8H), 3.49-3.42 (m, 4H), 3.39 (s, 8H), 3.12 (s, 2H), 2.77 (t, J=6.6 Hz, 4H), 2.61 (t, J=6.5 Hz, 4H), 1.43 (s, 36H); ¹³C NMR (50 MHz, CDCl₃): δ = 172.2, 170.5 (4C), 162.2, 143.7, 141.9, 86.0, 83.5, 81.0 (4C), 70.5, 70.0, 69.7, 69.4, 58.5, 55.8 (4C), 53.7, 53.4, 52.1 (2C), 39.4, 38.6, 28.1; LRMS (ESI+) m/z calcd for $C_{44}H_{74}F_3N_5O1_3$: 936.52 [*M*+H]⁺, found 936.40.

N-{2-DTPA-acetamide},N'-{3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine (**6**): TFA (200 μL, 1.14 mmol) was added to a solution of **5** (49 mg, 0.11 mmol) in DCM (2 mL). This mixture was stirred for five days (until MS analysis did not show starting material). The product was obtained after lyophilized from H₂O/dioxane as a white solid (28 mg, +99%). ¹H NMR (400 MHz, CD₃OD): δ =7.32 (dd, J=5.3, 1.9 Hz, 1 H), 7.23 (dd, J=5.3, 1.9 Hz, 1 H), 5.67 (s, 1 H), 5.58 (s, 1 H), 4.24 (br s, 1 H, NH), 3.66 (s, 2 H) 3.62–3.57 (m, 16 H) ppm 3.50–3.37 (m, 8 H), 3.20 (br s, 4 H); ¹³C NMR (50 MHz, CD₃OD): δ =174.31, 174.28, 174.15,174.14, 167.2, 164.9, 155.9, 144.3, 143.4, 127.6 (q, CF₃), 86.9, 84.2, 71.1, 71.0, 70.04, 70.00, 67.9, 55.8 (4C), 53.8 (2C), 50.9, 50.8, 40.2, 40.1; LRMS (ESI–) m/z calcd for C₂₈H₃₉F₃N₅O₁₃: 710.25 [M-H]⁻, found 710.30.

5-Methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxylic acid ethyl ester (14 a) and 6-methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxylic acid ethyl ester (14 b): A mixture of ethyl 4,4,4-trifluorobutynoate (0.61 g, 3.65 mmol) and 3-methylfuran (0.30 g, 3.65 mmol) was stirred under an Ar atmosphere for four days. The crude mixture was washed out with ether, concentrated, and purified by column chromatography (EtOAc/nheptane 1:5), resulting in a mixture of two regioisomers (ratio 1:1.4 for 14 b and 14 a respectively) as a slightly yellow oil (0.65 g, 72 %). ¹H NMR (300 MHz, CDCl₃) peaks assigned to compound 14 a: δ = 6.65 (m, 1 H), 5.38 (s, 1 H), 5.30 (d, J = 1.5 Hz, 1 H), 4.28 (m, 2 H), 1.99 (d, J = 2.0 Hz, 3 H), 1.31 (t, J = 6.9 Hz, 3 H); Peaks assigned to com-

pound **14b**: δ = 6.58 (m, 1 H), 5.61 (s, 1 H), 5.56 (d, J = 1.5 Hz, 1 H), 4.27 (m, 2 H), 2.05 (d, J = 2.0 Hz, 3 H), 1.32 (t, J = 6.9 Hz, 3 H); 13 C NMR (75 MHz, CDCl₃) δ = 162.1, 162.3, 156.2, 154.8, 152.0 (q), 151.2 (q), 150.8, 150.3, 133.7,134.5, 122.2 (q, CF₃), 121.5 (q, CF₃), 88.6, 87.4 (d), 85.9, 84.8, 61.7 (2C), 14.3, 14.2, 14.0 (2C). LRMS (ESI +) m/z calcd for $C_{11}H_{12}F_{3}O_{3}$ [M+H]⁺ 249.1, found 249.0.

5-Methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxylic acid (15 a) and 6-methyl-3-trifluoromethyl-7-oxa-bicyclo-[2.2.1]hepta-2,5-diene-2-carboxylic acid (15b): A mixture of 14a and 14b (0.30 g, 1.22 mmol) was dissolved in THF (16 mL) and cooled to 0°C. 1 M NaOH (aq) (1.22 mL) was added dropwise and the mixture was stirred overnight at RT. TLC analysis showed some starting material remaining after reacting overnight so another 1.22 mL NaOH (aq) (1 M) was added. After 30 min the reaction was completed and the volume of THF was reduced to 50% of the original volume by evaporation using a nitrogen flow. The mixture was diluted with HCl (aq) (5 mL, 1 M) and extracted with EtOAc (2× 75 mL). The organic layer was dried over Na₂SO₄ and evaporated under reduced pressure to obtain a yellow solid (0.26 g, 95%). A mixture of two regioisomers was obtained in a ratio of 1:1.4 for 15b and 15a respectively. ¹H NMR (300 MHz, CDCl₃) peaks assigned to compound **15 a**: $\delta = 6.67$ (t, J = 2.0 Hz, 1H), 5.56 (s, 1H), 5.34 (d, J = 1.5 Hz, 1 H), 2.01 (d, J = 2.0, 3 H); Peaks assigned to compound **15 b**: δ = 6.59 (t, J = 2.0 Hz, 1 H), 5.60 (s, 1 H), 5.42 (d, J = 1.5 Hz, 1 H), 2.05 (d, J = 2.0 Hz, 3 H); 13 C NMR (75 MHz, CDCl $_3$) δ = $166.7,\ 166.4,\ 150.9\ (q),\ 149.9\ (q),\ 134.4,\ 133.2,\ 121.2\ (q,\ CF_3),\ 121.0$ (q, CF₃), 88.3, 87.5, 84.9, 84.8, 14.1, 14.0; LRMS (ESI-) m/z calcd for $C_9H_6F_3O_3$: 219.1 [*M*-H]⁻, found 219.0.

5-Methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl glycine methyl ester (16a) and 6-methyl-3-trifluoromethyl-7oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl glycine methyl ester (16b): Oxanorbornadiene carboxylic acids 15a and 15b (87.5 mg, 0.39 mmol), glycine methyl ester·HCl (158 mg, 1.26 mmol), and 4-(dimethylamino)pyridine (DMAP, 183 mg, 1.50 mmol) were dissolved in CH₂Cl₂ (16 mL) and cooled to 0 °C. 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC·HCl, 80 mg, 0.42 mmol) was added and the reaction mixture was stirred at 0 °C for 30 min. The mixture was allowed to warm to RT and was stirred for an additional 16 h. The reaction was guenched with a 10% agueous solution of citric acid (20 mL) and extracted with EtOAc (2×20 mL). The combined organic layers were washed with brine (20 mL), dried over Na₂SO₄ and concentrated in vacuo. The crude mixture was purified by preparative TLC (CH₂Cl₂/MeOH 9:1) resulting in a mixture of compounds 16a and 16b (ratio of 1.4:1) as a colorless oil (85 mg, 74%). ¹H NMR (300 MHz, CDCl₃) peaks assigned to compound **16a**: δ = 6.69 (m, 1H), 5.56 (brs, 1H), 5.31 (d, J=1.5 Hz, 1 H), 4.13 (m, 2 H), 3.78 (s, 3 H), 1.98 (d, J=1.5 Hz, 3 H); Peaks assigned to compound **16 b**: δ = 6.56 (m, 1 H), 5.56 (m, 1 H), 5.37 (brs, 1 H), 4.13 (m, 2 H), 3.78 (s, 3 H), 2.09 (d, J = 1.5 Hz, 3 H); 13 C NMR (75 Mhz, CDCl₃): δ = 169.4, 162.2, 153.9, 134.3, 133.2, 89.4, 87.1, 86.7, 84.3, 52.6, 41.4, 14.0; HRMS (ESI+) m/z calcd for $C_{12}H_{12}O_4NF_3Na [M+Na]^+ 314.0616$, found 314.0597.

1,4-Dimethyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxylic acid ethyl ester (17): 2,5-dimethylfuran (279 μL, 2.62 mmol) was added to a solution of ethyl 4,4,4-trifluoro-2-buty-noate (436 mg, 2.62 mmol) in dioxane (0.5 mL). The mixture was heated to 103 °C under a nitrogen atmosphere and stirred overnight. The reaction mixture was cooled and diluted with CH_2CI_2 after which all the solvents were evaporated. The crude product was purified by column chromatography (MeOH/ CH_2CI_2 1:9) to obtain the pure product as a white solid (238 mg, 91%). ¹H NMR (300 MHz, $CDCI_3$): δ =7.01 (d, J=5.1 Hz, 1H), 6.91 (d, J=5.1 Hz,

1 H), 4.30 (m, 2 H), 1.81 (s, 3 H), 1.74 (s, 3 H), 1.31 (t, J=7.1 Hz, 2 H); 13 C NMR (75 MHz, CDCl₃): δ =163.7, 154.6, 149.0 (q, J=35 Hz), 147.5, 146.7, 121.9 (q, CF₃, J=270 Hz), 92.6, 91.4, 61.6, 15.1, 14.8, 13.9. Both HRMS and LRMS techniques were employed to acquire the mass of the described compound. Unfortunately, none of the techniques used gave a comprehensible mass spectrum.

3-Trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl-qlycine methyl ester (18): Oxanorbornadiene carboxylic acid 2 (20.6 mg, 0.1 mmol), H-Gly-OMe·HCl (13.8 mg, 0.11 mmol), and 4-(dimethylamino)-pyridine (DMAP, 24.2 mg, 0.2 mmol) were dissolved in 2 mL CH₂Cl₂ and cooled to 0 °C. 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC·HCl, 21 mg, 0.11 mmol) was added and the reaction mixture was stirred at 0 °C for 30 min. The mixture was allowed to warm to RT and was stirred for an additional 16 h. The reaction was quenched with 2 mL HCl (aq) (2 M) and extracted with EtOAc (2×5 mL). The combined organic layers were washed with brine (5 mL), dried over Na₂SO₄, and concentrated in vacuo. The crude mixture was purified by preparative TLC (CH₂Cl₂/MeOH 9:1) resulting in compound 18 as a slightly yellow solid (15.5 mg, 56%). $R_F = 0.55$ (CH₂Cl₂/MeOH 9:1); ¹H NMR (400 MHz, CDCl₃): $\delta = 7.33$ (dd, J = 5.3, 2.0 Hz, 1 H), 7.16 (dd, J = 5.3, 2.0 Hz, 1 H), 6.41 (brs, 1 H, NH), 5.68 (m, 2 H), 4.15 (dq, J = 18.5, 18.5, 18.5, 5.2 Hz, 2H) 3.80 (s, 3H); 13 C NMR (50 MHz, CDCl₃) $\delta = 166.6$, 154.9, 154.2, 150.7, 144.0, 142.7, [124.8, 118.6] CF₃, 85.1, 84.3; HRMS (ESI+) m/z calcd for $C_{11}H_{10}F_3NaNO_4$: 300.0460 $[M+Na]^+$, found 300.0459; FTIR ν_{max} (film): 2924, 1744, 1636, 1169, 1117, 886 cm $^{-1}$.

N-{Boc}-N'-{5-methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine (20 a) and N-{Boc}-N'-{6-methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine (20b): 4-(dimethylamino)-pyridine (DMAP, (50.6 mg, 0.41 mmol) was added to a solution of a mixture of compounds 15a and 15b (49 mg, 0.21 mmol) and 1-N-Boc-3,6-dioxa-8-octane-1,8-diamine (50.9 mg, 0.21 mmol) in dry CH₂Cl₂ (1.5 mL). The mixture was cooled to 0 °C, and 1-ethyl-3-(3dimethylaminopropyl) carbodiimide hydrochloride (EDC·HCl, 44.1 mg, 0.23 mmol) was added slowly. The mixture was stirred for 30 min at 0 °C and 16 h at RT. The reaction mixture was acidified with HCl (2 M) to a pH of 1–2 and extracted with CH_2Cl_2 (2×5 mL). The combined organic layers were dried over MgSO₄, concentrated in vacuo, and purified by preparative TLC (CH₂Cl₂/MeOH 9:1) resulting in compounds 20a and 20b as a slightly yellow oil (44.4 mg, 47%) A mixture of two regioisomers was obtained in a ratio of 1:1.4 for 20 b and 20 a respectively. ¹H NMR (300 MHz, CDCl₃) peaks assigned to compound **20a**: $\delta = 6.70$ (d, J = 6.7 Hz, 1 H), 6.44 (brs, 1H, NH), 5.54 (s, 1H), 5.28 (s, 1H), 4.96 (brs, 1H, NH), 3.60-3.52 (m, 10 H), 3.30 (q, J = 5.2 Hz, 2 H), 1.97 (d, J = 1.6 Hz, 3 H), 1.44 (s, 9 H); Peaks assigned to compound **20 b**: $\delta = 6.55$ (d, J = 6.6 Hz, 1 H), 6.44 (brs, 1H, NH), 5.54 (s, 1H), 5.34 (s, 1H), 4.96 (brs, 1H, NH), 3.30 (q, J=5.2 Hz, 2H), 3.60–3.52 (m, 10H), 3.30 (q, J=5.2 Hz, 2H), 2.10 (d, J = 1.9 Hz, 3 H), 1.44 (s, 9 H); ¹³C NMR (75 MHz, CDCl₃): $\delta = 162.4$, 155.9, 154.5, 153.8, 134.8, 133.1, 122.8 (q, CF₃), 89.4, 86.7, 84.1, 70.4, 70.2, 70.1, 69.4, 40.3, 39.4, 28.4, 14.0; HRMS (ESI+): m/z calcd for $C_{20}H_{30}F_3N_2O_6$ 451.2056 [*M*+H]⁺, found 451.2078.

N,N'-{5-methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine (**21 a**) and N,N'-{6-methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-diox-aoctane-1,8-diamine (**21 b**): Trifluoroacetic acid (TFA, 0.5 mL, 2.85 mmol) was added dropwise to a cooled solution (0 °C) of a mixture of **20 a** and **20 b** (42 mg, 0.093 mmol) in dry CH_2Cl_2 (2 mL). The reaction was stirred at 0 °C for 1 h after which the reaction was complete. The solvent was evaporated and the crude mixture was dissolved in H_2O (5 mL) and dioxane (5 mL) and freeze-dried

to afford compound **21 a** and **21 b** as a light yellow solid (43.1 mg, +99%). A mixture of two regioisomers was obtained in a ratio of 1:1.4 for **21 b** and **21 a** respectively. 1H NMR (400 MHz, CDCl₃) peaks assigned to compound **21 a** $\delta = 7.84$ (br s, 3 H, NH₃), 7.92 (br s, 1 H, NH), 6.66 (s, 1 H), 5.52 (s, 1 H), 5.27 (s, 1 H), 3.70–3.46 (m, 10 H), 3.15 (m, 2 H), 1.96 (s. 3 H); Peaks assigned to compound **21 b** $\delta = 7.84$ (br s, 3 H, NH₃), 7.92 (br s, 1 H, NH), 6.55 (s, 1 H), 5.52 (s, 1 H), 5.31 (s, 1 H), 3.70–3.46 (m, 10 H), 3.15 (s, 2 H), 2.06 (s, 3 H); 13 C NMR (50 MHz, CDCl₃) $\delta = 159.1$, 154.0, 151.1, 134.8, 133.1, 105.0, 103.0, 101.0, 100.0, 86.8, 86.5, 40.3, 14.0; HRMS (ESI+): m/z calcd for $C_{15}H_{22}F_3N_2O_4$ $[M+H]^+$ 351.1532, found 351.1541

N-{2-O-tert-butyl-DTPA-acetamide},N'-{5-methyl-3-trifluoromethyl-7oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine-TFA (22a) and N-{2-O-tert-butyl-DTPA-acetamide},N'-{6-methyl-3-trifluoromethyl-7-oxa-bicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6dioxaoctane-1,8-diamine-TFA (22b): Compounds 21a and 21b (39 mg, 0.084 mmol) were dissolved in dry CH₂Cl₂ (2 mL), and 4-(dimethylamino)-pyridine (DMAP, 20.4 mg, 0.17 mmol) and diethylenetriamine-N,N,N",N"-tetra-tert-butyl acetate-N'-acetic acid ((DTPAtert-butyl ester) 53 mg, 0.084 mmol) were added. The mixture was cooled to 0 °C followed by addition of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDC·HCl, 17.5 mg, 0.09 mmol). After 1 h stirring at 0°C the mixture was allowed to warm to RT and was stirred for an additional 4 h. The reaction mixture was quenched with 1 m HCl (2 mL) and the water layer was extracted with CH₂Cl₂ (2×2 mL). The combined organic layers was dried over Na₂SO₄ and subsequently evaporated. The crude product was purified by preparative TLC ($CH_2CI_2/MeOH\ 9:1$) to obtain pure product as a light brown oil (35 mg, 44%). A mixture of two regioisomers was obtained in a ratio of 1:1.4 for 22 b and 22 a, respectively. ¹H NMR (400 MHz, CDCl₃) peaks assigned to compound **22a** δ = 8.22 (brs, 1 H, NH), 6.75 (brs, 1 H, NH), 6.69 (t, J=1.9 Hz, 1 H), 5.53 (s, 1 H), 5.27 (s, 1 H), 3.60–3.39 (m, 18 H), 3.11 (s, 2 H), 2.77 (t, J=6.5 Hz, 2H), 2.61 (t, J=6.5 Hz, 2H), 1.96 (d, J=1.3 Hz, 3H), 1.43 (s, 36H); Peaks assigned to compound **22b** δ = 8.22 (brs, 1H, NH), 6.75 (brs, 1 H, NH), 6.54 (t, J = 1.9 Hz, 1 H), 5.51 (s, 1 H), 5.33 (s, 1 H), 3.60–3.39 (m, 18 H), 3.11 (s, 2 H), 2.77 (t, J=6.5 Hz, 2 H), 2.61 (t, J= 6.5 Hz, 2 H), 2.09 (d, J = 1.7 Hz, 3 H), 1.43 (s, 36 H); ¹³C NMR (75 MHz, CDCl₃) δ = 172.2, 170.5, 162.5, 155.2 (q), 153.7, 142.5, 134.4, 122.8 (q, CF₃), 89.3, 86.7, 84.0, 81.0, 70.5, 70.0, 69.8, 69.4, 58.6, 55.8, 53.8, 52.1, 39.4, 38.6, 28.1, 14.0; HRMS (ESI+): m/z calcd for $C_{45}H_{75}F_3N_5O_{13}$ [M+H]⁺ 950.5313, found 950.5361.

N-{2-DTPA-acetamide},N'-{5-methyl-3-trifluoromethyl-7-oxa-bicyclo-[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine (**23 a**) N-{2-DTPA-acetamide},N'-{6-methyl-3-trifluoromethyl-7-oxabicyclo[2.2.1]hepta-2,5-diene-2-carboxyl}-3,6-dioxaoctane-1,8-diamine (23 b): TFAA (200 μ L) was added to a solution of compounds 22 a and 22b (25 mg, 0.026 mmol) in CH₂Cl₂ (2 mL). This mixture was stirred for six days (until MS analysis did not show starting material). The product was obtained as a white solid in quantitative yield ($+99\,\%$) after lyophilization from $H_2O/dioxane$. A mixture of two regioisomers was obtained in a ratio of 1:1.4 for 23 b and 23 a respectively. ¹H NMR (400 MHz, CD₃OD) peaks assigned to compound **23 a** δ = 6.96 (t, J = 1.7 Hz, 1 H), 5.50 (s, 1 H), 5.36 (s, 1 H), 4.31 (br s, 2H), 3.66-3.58 (m, 18H), 3.49-3.42 (m, 2H), 3.30-3.15 (m, 8H), 1.99 (d, J=1.5 Hz, 3 H); Peaks assigned to compound **23 b** $\delta=6.63$ (m, 1H), 5.56 (s, 1H), 5.30 (s, 1H), 4.31 (brs, 2H), 3.66-3.58 (m, 18H), 3.48-3.42 (m, $2H_{\rm H}$), 3.28-3.14 (m, 8H), 2.07 (d, J=1.7 Hz, 3H); ¹³C NMR (75 MHz, CD₃OD) δ = 174.6 (4C), 167.2, 165.6, 155.8, 135.4, 90.5, 87.7, 84.1, 71.4, 71.3, 70.3, 68.2, 55.9, 55.6, 54.3, 50.9, 40.5, 40.4, 14.0; HRMS (ESI+): m/z calcd for $C_{29}H_{43}F_3N_5O_{13}$ [M+H]⁺ 726.2809, found 726.2837.

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