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Transition Probabilities in the CuCl\textsuperscript{2−}-Complex

By

A. VAN DER AVOIRD* and P. ROS**

Transition probabilities in the CuCl\textsuperscript{2−}-complex have been calculated using the L.C.A.O.-
functions obtained in the preceding paper by P. Ros and G. C. A. SCHUIT. For all transitions
the calculated absolute oscillator strengths turn out to be a factor 2 too large with respect to
the experimental values. The calculated transition probabilities are mainly determined by the
amount of mixing of the ligand functions in the molecular orbitals. If this mixing is assumed
somewhat less the agreement is still better.

Unter Benutzung der L.C.A.O.-Funktionen aus der vorstehenden Arbeit von P. Ros und
G. C. A. SCHUIT wurden die Übergangswahrscheinlichkeiten im CuCl\textsuperscript{2−}-Komplex berechnet.
Die absolute Oszillatorstärke, die hauptsächlich vom Gewicht der Ligandenfunktionen ab­
hängt, wird für alle Übergänge um den Faktor 2 zu groß. Verringert man den Beitrag dieser
Funktionen, wird die Übereinstimmung mit dem Experiment besser.

A l'aide des fonctions L.C.A.O. du travail précédent de P. ROS et G. C. A. SCHUIT, nous
calculons les moments de transition pour l'ion complexe CuCl\textsuperscript{2−}. Pour toutes les bandes les
forces oscillatrices calculées sont deux fois trop grandes devant l'expérience. Elles sont déter­
mìnées surtout par le poids des orbitales des ligandes dans les O.M.; l'accord s'améliore par
une diminution de ce poids.

Introduction

In the last twenty years many attempts have been made to calculate the
stability and the charge distribution of inorganic complexes. In most of these
calculations, more or less empirical, the results were checked by comparing them
with data obtained from the optical spectra [2, 3, 4, 6, 7, 9, 10, 14, 15, 20, 21].
In the present report a calculation of the intensities of the bands in the CuCl\textsuperscript{2−}-
spectrum is described. It is carried out making use of the one electron wave
functions and charge distribution arising from a quantitative molecular orbital
calculation in which as few as possible empirical parameters are introduced (see
preceding paper by P. Ros and G. C. A. SCHUIT and [19]). By comparing the
calculated intensities with the experimental values the accuracy of the M.O. cal­
culation is checked. By evaluating some terms which up to now have always been
approximated or neglected we hope to be able to test and maybe to simplify the
usual method of calculating the intensities.

Transition Probability

The probabilities of electric dipole transitions and therefore the intensities of
the corresponding absorption bands can be expressed in the oscillator strengths $f$.  

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bridge, Massachusetts.
It is possible to determine \( f \) from the optical spectrum, as well as to relate it to the wave functions of the states between which the transition occurs [3, 5, 17].

1. **Theoretical Calculation of \( f \):** The formula for the theoretical calculation of \( f \) is given by:

\[
 f = 1.085 \times 10^{11} \tilde{v} P,
\]

\[
P = A_{\Pi} \sum_{\Pi} | \langle \psi_1 | r | \psi_{\Pi} \rangle |^2
\]

with \( \tilde{v} \) = mean wave number of the absorption band (cm\(^{-1}\)),

\( P \) = dipole strength (cm\(^2\)),

\( \psi_1 \) = wave function of the ground state, \( n \)-fold degenerate,

\( \psi_{\Pi} \) = wave function of the excited state, \( m \)-fold degenerate,

\( r \) = position vector, pointing from the origin to the point of integration.

It makes no difference for the calculated value of \( f \) which point is chosen as origin. We shall take the origin of the main coordinate system, located on the central ion.

To evaluate the transition probabilities in CuCl\(_2^–\) we have to calculate \( n \times m \) integrals \( \langle \psi_1 | r | \psi_{\Pi} \rangle \) for each allowed transition. \( \psi_1 \) and \( \psi_{\Pi} \) are wave functions of the whole complex and they may be written as Slater determinantal wave functions, composed from the one electron wave functions of the filled molecular orbitals.

We shall now introduce the following approximations:

a) We assume the CuCl\(_2^–\)-complex to be perfectly tetrahedral. (The real form of the complex determined by X-ray analysis [13] is a tetragonally distorted tetrahedron; the Cu-Cl distance is equal to 2.22 Å). It can be shown that ignoring the distortion does not affect the \( f \)-values much.

b) We suppose that the one electron molecular orbitals are the same before and after the transition. These molecular orbitals are known from the M.O. calculation described in the preceding paper by P. Ros and G. C. A. Schuit. It appears from this calculation that this second assumption is justified.

In the ground state \( \psi_1 \) of CuCl\(_2^–\) we have an electron-hole in, say, the molecular orbital \( \chi_{\Pi} \). In the excited state \( \psi_{\Pi} \) we have a hole in \( \chi_{\Pi} \). We can now prove, making use of the orthonormality relations of the molecular orbitals, that

\[
\langle \psi_1 | r | \psi_{\Pi} \rangle = \langle \psi_{\Pi} | r | \psi_{\Pi} \rangle = \langle \psi_1 | r | \psi_{\Pi} \rangle
\]

The determinantal wave functions can simply be replaced by the one electron molecular orbitals containing the hole. Another consequence of the assumptions is that we can use the molecular orbitals calculated for the ground state of tetrahedral CuCl\(_2^–\). The transitions that are allowed and the component integrals of each "degenerate" transition that are not equal to zero are found by applying the multiplication rules of the group \( T_d \) [8, 12]. They are listed in Tab. 1. In the last column of this table the relation between the different components of the same transition is indicated. The integrals \( \langle \psi_1 | r | \psi_{\Pi} \rangle \) that are not equal to zero because of symmetry have to be calculated. Since a molecular orbital \( \psi \) is of the form

\[
\psi = C_1 \varphi_M + C_2 \chi_L \text{ with } \varphi_M = \text{atomic orbital of the central ion } M
\]

and \( \chi_L = \sum_s C_s \varphi_s \); \( \varphi_s \) = atomic orbital of ligand \( L_s \)

\[\text{and } \chi_L = \sum_s C_s \varphi_s \]
Transition probabilities in the CuCl$_2$-complex

Table 1. The dipole vector $\mathbf{r}$ corresponds to the representation $T_2$

<table>
<thead>
<tr>
<th>Allowed transitions, ground state $T_2$</th>
<th>Allowed components of these transitions with polarisation</th>
<th>Factor of $a_{1}$ in the direct product of the basis vectors. This factor determines the relation between the components of a transition</th>
</tr>
</thead>
<tbody>
<tr>
<td>$T_2 \rightarrow A_1$</td>
<td>$\xi \rightarrow a_{1}$</td>
<td>$\xi x a_{1} = \frac{1}{\sqrt{3}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\eta \rightarrow a_{1}$</td>
<td>$\eta y a_{1} = \frac{1}{\sqrt{3}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\zeta \rightarrow a_{1}$</td>
<td>$\zeta z a_{1} = \frac{1}{\sqrt{3}} a_{1} + \ldots$</td>
</tr>
<tr>
<td>$T_2 \rightarrow E$</td>
<td>$\xi \rightarrow \theta$</td>
<td>$\xi x \theta = -\frac{1}{2\sqrt{3}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\eta \rightarrow \theta$</td>
<td>$\eta y \theta = -\frac{1}{2\sqrt{3}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\zeta \rightarrow \theta$</td>
<td>$\zeta z \theta = -\frac{1}{\sqrt{3}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\xi \rightarrow e$</td>
<td>$\xi x e = -\frac{1}{2} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\eta \rightarrow e$</td>
<td>$\eta y e = -\frac{1}{2} a_{1} + \ldots$</td>
</tr>
<tr>
<td>$T_2 \rightarrow T_1$</td>
<td>$\xi \rightarrow \gamma$</td>
<td>$\xi y \gamma = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\xi \rightarrow \beta$</td>
<td>$\xi z \beta = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\eta \rightarrow \gamma$</td>
<td>$\eta x \gamma = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\eta \rightarrow \beta$</td>
<td>$\eta z \beta = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\zeta \rightarrow \alpha$</td>
<td>$\eta z \alpha = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\zeta \rightarrow \beta$</td>
<td>$\xi x \beta = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\zeta \rightarrow \alpha$</td>
<td>$\zeta y \alpha = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td>$aT_2 \rightarrow bT_2$</td>
<td>$\xi \rightarrow \zeta$</td>
<td>$\xi y \zeta = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
<td></td>
<td>$\xi \rightarrow \eta$</td>
<td>$\xi z \eta = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
<tr>
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</tr>
<tr>
<td></td>
<td>$\zeta \rightarrow \xi$</td>
<td>$\zeta y \xi = -\frac{1}{\sqrt{6}} a_{1} + \ldots$</td>
</tr>
</tbody>
</table>
we obtain three types of integrals:

\[ a: \langle q_M | r | q_M' \rangle, \]
\[ b: \langle q_M | r | \chi_L \rangle, \]
\[ c: \langle \chi_L | r | \chi_L' \rangle. \]

\[ a: \langle q_M | r | q_M' \rangle \] is an integral of functions that belong to the same nucleus. With the assumed atomic orbitals, the radial part of which is given by a linear combination of Slater functions, an analytical form for these integrals is easily derived. Then a computer-program can be written.

\[ b: \langle q_M | r | \chi_L \rangle \] with \( \chi_L = \sum \xi C_\alpha q_\alpha \) summation over the ligands \( L_\alpha \). The substitution of \( \chi_L \) yields: \( \sum \xi C_\alpha \langle q_M | r | q_\alpha \rangle \).

Now we have integrals of functions that belong to two different nuclei, namely \( M \) and \( L_\alpha \). These integrals can be evaluated after a rotation of the main coordinate system and a corresponding transformation operating on \( q_M \) and \( r \). By a method given by Ballhausen \([5]\) using elliptical coordinates \([18]\) we can expand them in \( A_n \) and \( B_n \) integrals that can be computed.

\[ c: \langle \chi_L | r | \chi_L' \rangle \] may be written as \( \sum \xi \sum \gamma C_\alpha C_\beta \langle q_\alpha | r | q_\beta \rangle \). Most of these integrals contain functions that belong to three different nuclei: \( M, L_\alpha \) and \( L_\beta \). We see in Fig. 1 that \( r \) can be written as \( r = r_0 + r' \), \( r' \) belonging to \( L_\alpha \) or \( L_\beta \). This gives:

\[ \langle q_\alpha | r | q_\beta' \rangle = r_0 \langle q_\alpha | q_\beta' \rangle + \langle q_\alpha | r' | q_\beta' \rangle. \]

After rotations of the coordinate systems on the ligands \( L_\alpha \) and \( L_\beta \) we are able to calculate \( \langle q_\alpha | q_\beta' \rangle \) and \( \langle q_\alpha | r' | q_\beta' \rangle \) by a similar method as used in \( b \).

The computer used is the IBM 1620.

All integrals have been computed for different charge distributions and have been obtained as functions of the charge distribution \( (A, B, C \) and \( D, \) see M.O.

Table 2

<table>
<thead>
<tr>
<th>Electron transition</th>
<th>( \nu_{max} ) ( \tilde{\nu} ) (cm(^{-1}))</th>
<th>( \xi_{max} ) ( \xi ) (l mol(^{-1}) cm(^{-1}))</th>
<th>Exp. ( f )</th>
<th>Theor. ( f )</th>
</tr>
</thead>
<tbody>
<tr>
<td>crystal field</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2( e ) ( \rightarrow ) 4( t_2 )</td>
<td>6,000</td>
<td>shoulder</td>
<td>0.0037</td>
<td>0.0075</td>
</tr>
<tr>
<td>first charge transfer</td>
<td>8,500</td>
<td>122</td>
<td></td>
<td></td>
</tr>
<tr>
<td>1( t_1 ) ( \rightarrow ) 4( t_2 )</td>
<td>24,500</td>
<td>2400</td>
<td>0.040</td>
<td>0.080</td>
</tr>
<tr>
<td>second charge transfer</td>
<td>34,000</td>
<td>5700</td>
<td>0.120</td>
<td>0.184</td>
</tr>
<tr>
<td>third charge transfer</td>
<td>41,000</td>
<td>( \approx 1800 )</td>
<td>0.032</td>
<td>0.052</td>
</tr>
<tr>
<td>( 1 e ) ( \rightarrow ) 4( t_2 )</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>( 2 t_2 ) ( \rightarrow ) 4( t_2 )</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>( 2 a_1 ) ( \rightarrow ) 4( t_2 )</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>
calculation) by means of interpolation formulas. Substitution of the numerical values of $A$, $B$, $C$, $D$ and of the coefficients of the atomic orbitals as found for the ground state of tetrahedral CuCl$_4^{2-}$ produces the integrals $\langle \psi_I | \mathbf{r} | \psi_{II} \rangle$ and the oscillator strengths $f$. The results are given in Tab. 2. The assumed sequence of the bands is discussed later.

2. Experimental determination of $f$: In agreement with the results obtained by several investigators [6, 10, 11, 16] we found that it makes no difference to the spectrum of CuCl$_4^{2-}$ whether it is measured in solid state or in solution. The complex seems to have the same structure independent of the surroundings and
of the type of cation. Measuring in solution it is necessary to maintain a large
Cl⁻-concentration to prevent the formation of complexes as CuCl⁻^[11].
A Zeiss spectrophotometer PM Q II with monochromator M 4 Q II was used.
The obtained spectra are reproduced in Fig. 2, 3, and 4. From the spectrum of
Fig. 4 the f-values of the bands have been determined with the formula

\[ f = 4.32 \times 10^{-12} \int \epsilon_v \, dv \]

\[ \epsilon_v = \text{molar extinction at wave number } v \text{ (cm}^2\text{mol}^{-1}) \]

The results are given in Tab. 2.

**Discussion**

1. The largest contributions to the oscillator strengths of all transitions are
given by certain integrals \( \langle \chi_L \mid r \mid \chi'_L \rangle \) (Tab. 3). The integrals that do not occur
in this table are at least a factor 10 smaller than the tabulated integrals. A fairly

<table>
<thead>
<tr>
<th>Transition</th>
<th>Component</th>
<th>Calculated value</th>
<th>Most important term</th>
<th>Contribution of this term</th>
<th>Approximation of ( \langle \chi \mid r \mid \chi' \rangle )</th>
<th>Approximation ( c_2 c_2 G r_0 )</th>
</tr>
</thead>
<tbody>
<tr>
<td>2e (-\rightarrow) 4t₂ ( \langle \xi \mid x \mid \theta \rangle )</td>
<td>0.28 a.u.</td>
<td>( \langle \chi (x, y) \mid r \mid \chi' (x, y) \rangle )</td>
<td>0.30 a.u.</td>
<td>( x_0 ) = 1.21 a.u.</td>
<td>0.29 a.u.</td>
<td></td>
</tr>
<tr>
<td>1t₁ (-\rightarrow) 4t₂ ( \langle \xi \mid x \mid \beta \rangle )</td>
<td>0.73</td>
<td>( \langle \chi (x, y) \mid r \mid \chi' (x, y) \rangle )</td>
<td>0.80</td>
<td>( x_0 \sqrt{3} ) = 2.10</td>
<td>0.84</td>
<td></td>
</tr>
<tr>
<td>3t₂ (-\rightarrow) 4t₂ ( \langle \xi \mid x \mid \eta \rangle )</td>
<td>0.95</td>
<td>( \langle \chi (z) \mid r \mid \chi' (z) \rangle )</td>
<td>1.07</td>
<td>( x_0 ) = 2.42</td>
<td>1.04</td>
<td></td>
</tr>
<tr>
<td>1e (-\rightarrow) 4t₂ ( \langle \xi \mid x \mid \theta \rangle )</td>
<td>0.32</td>
<td>( \langle \chi (x, y) \mid r \mid \chi' (x, y) \rangle )</td>
<td>0.40</td>
<td>( x_0 ) = 2.42</td>
<td>0.39</td>
<td></td>
</tr>
<tr>
<td>2t₂ (-\rightarrow) 4t₂ ( \langle \xi \mid x \mid \eta \rangle )</td>
<td>0.97</td>
<td>( \langle \chi (x, y) \mid r \mid \chi' (x, y) \rangle )</td>
<td>0.47</td>
<td>( x_0 ) = 2.42</td>
<td>0.44</td>
<td></td>
</tr>
<tr>
<td>2a₁ (-\rightarrow) 4t₂ ( \langle \xi \mid x \mid a₁ \rangle )</td>
<td>1.60</td>
<td>( \langle \chi (z) \mid r \mid \chi' (z) \rangle )</td>
<td>1.60</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

good approximation to the \( f \)-values in the CuCl₁⁻⁻-complex can be given by taking
only these \( \langle \chi (z) \mid r \mid \chi' (z) \rangle \) and \( \langle \chi (x, y) \mid r \mid \chi' (x, y) \rangle \) integrals into account. This
is so because in the molecular orbitals used the coefficients of the Cu 4s- and 4p-
functions and of the Cl \chi (3s)-functions are small (see Tab. 4). The \( \langle \chi (z) \mid r \mid \chi' (z) \rangle \)
and \( \langle \chi (x, y) \mid r \mid \chi' (x, y) \rangle \) integrals can in their turn be approximated by \( G r_0 \).

\( r_0 = x_0, y_0 \) or \( z_0 = \frac{\text{R}_{\text{Cu-Cl}}}{\sqrt{3}} \), the projection of the Cu-Cl distance on an
axis of the main coordinate system.

\( G \) is a factor 1, \( \frac{3}{2} \) or \( \frac{1}{2} \) \( \sqrt{3} \) depending on the coefficients \( C_\alpha \) in

\[ \chi_L = \sum \alpha C_\alpha q_\alpha \].

The last approximation implies in fact neglecting the overlap between the
different ligands.
In this way it is possible to calculate the \( f \)-values in a fairly good approximation, using only the M.O. coefficients of the ligand orbitals and the distance between the central ion and the ligands (see Tab. 3). This approximation was already used by Wolfsberg and Helmholtz [20]. For the CuCl\(^2^-\)-complex it appears to be a good one. For other complexes it may also be good when certain conditions are fulfilled.

The approximation of Ballhausen and Liehr [2], who neglected all terms involving \( \langle \chi_L | \mathbf{r} | \chi_L \rangle \) appears to be less satisfactory, for from the exact calculation it follows that these terms give the most important contributions to the transition probabilities.

Table 4

| Type of integral \( \langle \varphi | \mathbf{r} | \varphi' \rangle \) | Order of magnitude of \( \langle \varphi | \mathbf{r} | \varphi' \rangle \) | Order of magnitude of \( c' \langle \varphi | \mathbf{r} | \varphi' \rangle \) |
|---|---|---|
| copper – copper | \( \langle s | \mathbf{r} | p \rangle \) | 2 a.u. | \( \leq 0.005 \) a.u. |
| | \( \langle d | \mathbf{r} | p \rangle \) | 0.1 | \( \leq 0.005 \) |
| copper – ligands | \( \langle s | \mathbf{r} | \chi \rangle \) | 0.5 | \( \leq 0.05 \) |
| | \( \langle p | \mathbf{r} | \chi \rangle \) | 1 | \( \leq 0.05 \) |
| | \( \langle d | \mathbf{r} | \chi \rangle \) | 0.1 | \( \leq 0.05 \) |
| ligands – ligands | \( \langle \chi (s) | \mathbf{r} | \chi ' (s) \rangle \) | 2.5 | \( \leq 0.05 \) |
| | \( \langle \chi (z) | \mathbf{r} | \chi ' (z) \rangle \) | 2.5 | \( \leq 1.5 \) |
| | \( \langle \chi (x, y) | \mathbf{r} | \chi ' (x, y) \rangle \) | 1.5 | \( \leq 1.0 \) |
| | \( \langle \chi (s) | \mathbf{r} | \chi ' (s) \rangle \) | 0.5 | \( \leq 0.05 \) |
| | \( \langle \chi (z) | \mathbf{r} | \chi ' (z) \rangle \) | 0.5 | \( \leq 0.05 \) |
| | \( \langle \chi (x, y) | \mathbf{r} | \chi ' (x, y) \rangle \) | 0.1 | \( \leq 0.05 \) |

Several investigators e.g. [7, 9, 15] calculated the terms \( \langle \varphi_M | \mathbf{r} | \varphi_M \rangle \) and \( \langle \varphi_M | \mathbf{r} | \chi_L \rangle \) exactly but approximated the integrals \( \langle \chi_L | \mathbf{r} | \chi_L \rangle \) by the method of Wolfsberg and Helmholtz; however since in this approximation terms that have the same order of magnitude as the terms involving \( \langle \varphi_M | \mathbf{r} | \varphi_M \rangle \) and \( \langle \varphi_M | \mathbf{r} | \chi_L \rangle \) are neglected it does not make much sense to calculate these integrals exactly.

2. To obtain the positions of the optical bands we must take the difference between the total energies of the excited and ground states. In the M.O. description we saw, however, that it is too difficult to calculate the total energies from the one electron energies and the electron repulsions. Therefore the best we can do is to estimate the positions of the absorption bands from the one electron energies of the ground state. A qualitative scheme for this is given in Fig. 5 for the tetrahedral \( (T_d) \) and for the real, distorted tetrahedral, \( (D_{2d}) \) form of the CuCl\(^2^-\)-complex. The distortion causes a splitting of the degenerate energy levels:

\[
\begin{align*}
T_d & \quad D_{2d} \\
A_1 & \rightarrow A_1 \\
E & \rightarrow A_1 + B_2 \\
T_1 & \rightarrow E + A_2 \\
T_2 & \rightarrow E + B_2.
\end{align*}
\]
In $T_d$-symmetry the ground state has a symmetry $T_2$, in $D_{2d}$ a symmetry $B_2$. The allowed transitions are (see Tab. 1):

$T_d$

<table>
<thead>
<tr>
<th>Transition</th>
<th>Corresponding to</th>
</tr>
</thead>
<tbody>
<tr>
<td>$T_2 \rightarrow A_1$</td>
<td>$B_2 \rightarrow aA_1$</td>
</tr>
<tr>
<td>$T_2 \rightarrow E$</td>
<td>$B_2 \rightarrow bA_1$</td>
</tr>
<tr>
<td>$T_2 \rightarrow T_1$</td>
<td>$B_2 \rightarrow aE$</td>
</tr>
<tr>
<td>$aT_2 \rightarrow bT_2$</td>
<td>$B_2 \rightarrow bE$</td>
</tr>
</tbody>
</table>

(The prefixes $a$ and $b$ are added to distinguish states with the same symmetry.)

The observed bands in the spectrum are:

- a) The crystal field band that has a maximum extinction at $8,500 \text{ cm}^{-1}$ and a small shoulder at $6,000 \text{ cm}^{-1}$.

  In $T_d$ symmetry the only crystal field transition is $T_2 \rightarrow E$; in terms of one electron transitions: $2e \rightarrow 4t_2$.

  In $D_{2d}$ three crystal field transitions are possible:

  - $B_2 \rightarrow E$ or $5e \rightarrow 4b_2$
  - $B_2 \rightarrow A_1$ or $4a_1 \rightarrow 4b_2$
  - $B_2 \rightarrow B_1$ or $2b_1 \rightarrow 4b_2$

  of which the last one is symmetry-forbidden.

  The transition $4a_1 \rightarrow 4b_2$ corresponds to the large band. The shoulder is caused by the transition $5e \rightarrow 4b_2$ ($5e$ and $4b_2$ are the states resulting from the split $4t_2$ state) or by the forbidden transition $2b_1 \rightarrow 4b_2$.

  b) Three charge transfer bands with maxima at 24,500, 34,000 and 41,000 cm$^{-1}$. We cannot predict the positions of these bands quantitatively from the one electron energies, we can only say something about the sequence in which they appear.

  So we expect the first charge transfer band to be caused by the transition:

  $4e \rightarrow 4b_2$

  the second by:

  $3t_2 \rightarrow 4t_2$

  and the third by:

  $1e \rightarrow 4t_2$

  or by:

  $2t_2 \rightarrow 4t_2$

  or less likely by:

  $2a_1 \rightarrow 4t_2$
This sequence of the absorption bands is affirmed by the comparison of the calculated oscillator strengths with the observed values (see Tab. 2). The third charge transfer band can now be ascribed to the transition $1e \rightarrow 4t_2$ since the other two possibilities require much higher $f$-values.

Conclusions

1. For all transitions the calculated $f$-values are about a factor 2 too high with respect to the experimental values. This may partly be ascribed to the fact that eq. (3) is only valid for ideal gases, whereas our absorption spectrum has been measured in solution.

2. The agreement of the calculated and observed relative $f$-values is still better.

3. The above agreement is good enough to distinguish different transitions from their $f$-values.

4. The transition probabilities, also that of the crystal field transition, are mainly determined by the amount of mixing of the ligand functions in the molecular orbitals. The mixing of the copper $4p$-function affects the intensities of the bands hardly at all.

5. The agreement of the calculated $f$-values with the observed values is still better if one assumes that the mixing of the ligand functions in the molecular orbitals is somewhat less.

References


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