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Differential cross sections for rotational excitation of NH₃ by collisions with Ar and He: Close coupling results and comparison with experiment

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By means of the close coupling method we have calculated state-to-state differential and integral cross sections for rotational excitation and inversion of NH₃ by collisions with Ar and He. For NH₃–Ar we used an empirical and a scaled ab initio potential, for NH₃–He an ab initio potential. The differential cross sections for NH₃–Ar obtained from the empirical potential have an angular dependence that is in closer agreement with experiment than those obtained from the scaled ab initio potential. The integral cross sections are reproduced equally well by the two potentials. Also for NH₃–He the differential cross sections are in accordance with experiment. For the integral cross sections the agreement is good too, except for the very small cross sections to some of the higher rotationally excited states. For both complexes the differential cross sections show a strong dependence on energy, both in their angular dependence and in their relative magnitudes. © 1996 American Institute of Physics. [S0021-9606(96)00630-7]

I. INTRODUCTION

Differential cross sections (DCSs) and integral cross sections (ICSs) for rotationally inelastic scattering are both very sensitive probes of the anisotropy of the intermolecular potential surface. The quality of potential surfaces can therefore be tested by comparing the calculated state-to-state cross sections with experimentally determined ICSs or DCSs. The information that can be derived from DCSs, however, is more detailed than that contained in ICSs, because DCSs yield information about the dependence of the collision dynamics on the impact parameter. Small angle scattering results from collisions with a large impact parameter, and is dominated by the attractive long range part of the potential, whereas large angle scattering originates from collisions with a small impact parameter, that mainly probe the repulsive short range part of the potential. For ICSs it is not possible to relate their magnitude to specific regions of the potential.

In three previous papers we calculated ICSs using the close coupling (CC) method for NH₃–Ar¹² and NH₃–He³ at a relative kinetic energy of 485 and 436 cm⁻¹, respectively, and compared them with experiment. For NH₃–Ar we have used three different potentials, an ab initio potential,⁴ a potential in which one parameter in the ab initio potential was scaled to improve the agreement between bound states calculations and spectroscopic data,⁵ and an empirical potential.⁶ We found that the scaling of the ab initio potential improved the agreement with the experimental scattering data as well as with the ab initio potential, just as it did for the bound state spectrum. Best agreement, however, was obtained when we used the empirical potential. For NH₃–He we made use of only one ab initio potential, which gave reasonable agreement with the experimental data.³

As a further test of our potentials, we present in this paper DCSs obtained from CC calculations and we compare them with experimental DCSs that have recently become available for NH₃–Ar⁷ and for NH₃–He⁸ at a relative kinetic energy of 1274 and 1130 cm⁻¹, respectively. The calculations were performed at the energies attained in these experiments for both collision complexes, and for the cross sections that are experimentally observed. In the ortho NH₃–Ar calculations we used the two best potentials, i.e., the scaled ab initio potential and the empirical potential. For NH₃–Ar we only used the empirical potential.

To get an idea of the energy dependence of the DCSs, we have also calculated the DCSs for both complexes at the lower energies at which the calculations in our previous papers were performed. The calculations automatically yield ICSs, too. Since the ICSs have also been experimentally determined for the high energies,⁸,⁹ we also compare the theoretical and experimental ICSs at the high energies for both complexes.

The paper is organized as follows: In Sec. II we describe the intermolecular potentials. In Sec. III we outline the settings of the parameters used in the scattering calculations. The results are presented and discussed in Sec. IV and summarized in Sec. V.

II. INTERMOLECULAR POTENTIALS

The empirical and scaled ab initio potential surfaces that we used in our NH₃–Ar calculations are described in Refs. 6 and 4, respectively. The ab initio potential for NH₃–He is described in Ref. 3. Here, we summarize the most important features of the three potentials.

The empirical NH₃–Ar potential surface has been determined from a least squares fit to 61 far infrared and microwave vibration–rotation–tunneling measurements and to temperature dependent second virial coefficients. A surface with 13 variable parameters has been optimized to accurately reproduce the spectroscopic observables, using the collocation method for the bound state calculations.¹⁰ This model surface contains long-range attractive forces of induction and dispersion and short range coulombic and electron exchange
repulsive forces. The long-range dispersion coefficients were taken from \textit{ab initio} calculations,\textsuperscript{3,11} since they cannot be properly derived from experiment. The three intermolecular coordinates ($R, \theta, \phi$) were treated without invoking any approximation regarding their separability.

The NH$_3$–Ar \textit{ab initio} potential has been obtained from Heitler–London short range calculations and from multipole-expanded dispersion and induction long-range contributions. The \textit{ab initio} potential for NH$_3$–He has been obtained from supermolecule self-consistent field calculations, supplemented with multipole-expanded dispersion contributions at the time-dependent coupled Hartree–Fock plus second order true correlation level.\textsuperscript{11} For both potentials a Tang–Toennies–like damping was applied to the long-range energy. The \textit{ab initio} potentials are expressed with respect to a body fixed coordinate frame in which the $z$ axis coincides with the $C_3$ axis of ammonia, with the nitrogen atom being on the positive $z$ axis. The frame origin is in the center of mass of NH$_3$ and one hydrogen nucleus is put into the $x$–$z$ plane, with positive $x$. The position vector $\mathbf{R}$ of the Ar or He atom is given by its usual spherical polar coordinates ($R, \theta, \phi$). Both potentials are expanded in spherical harmonics $Y_{\lambda \mu}$ through $\lambda = 7$

\begin{equation}
V(R, \theta, \phi) = \sum_{\lambda \mu} v_{\lambda \mu}(R) Y_{\lambda \mu}(\theta, \phi).
\end{equation}

Due to the threefold symmetry of the ammonia only terms with $\mu = 0, 3, 6, \ldots$ are present. The scaling of the NH$_3$–Ar \textit{ab initio} potential consists in multiplying a short range parameter in one of the expansion coefficients, $v_{33}(R)$, by a factor of 1.43. This appeared to improve the agreement with the spectrum considerably.\textsuperscript{5}

The empirical potential is expressed in a frame that is slightly different from the one used for the \textit{ab initio} potentials, i.e., the nitrogen atom is put on the negative $z$ axis. This means that we have to apply the transformation $\theta = \pi - \theta$ and $\phi = \phi$, in order to express both potentials in the same coordinate system.

In the present calculations the potential surfaces are 3-dimensional, i.e., they do not depend explicitly on the ammonia inversion coordinate. This implies that we have to apply the delta function model\textsuperscript{12,13} to treat the NH$_3$ inversion motion. It has been shown,\textsuperscript{1} however, that this has only a small effect—in the order of 3\%—on the cross sections as compared to calculations that take the inversion coordinate explicitly into account.

### III. COMPUTATIONAL ASPECTS

The close coupling calculations were carried out with the HIBRIDON inelastic scattering code.\textsuperscript{14} An explanation of the parameters used in the calculations can be found in Ref. 1.

The values of the total energies $E_{\text{tot}}$ in the present calculations are given in Table I. The values of the higher energies, 1274 cm$^{-1}$ for NH$_3$–Ar and 1130 cm$^{-1}$ for NH$_3$–He, are determined by the total energies in the experiments\textsuperscript{7,8} with which we compare our differential cross sections. The lower energies are set to 485 and 436 cm$^{-1}$ for NH$_3$–Ar and NH$_3$–He, respectively. These values are determined by the settings in earlier calculations, in which we compared theoretical ICSs with experimental ones.

The difference in the ortho and para total energies is caused by the difference in ground state energy of the two species. The ortho NH$_3$ with initial state $j = k = 0$, where $j$ and $k$ are the symmetric top quantum numbers of the NH$_3$ monomer, has zero internal energy, so the total energy is equal to the relative kinetic energy. The initial $j = k = 1$ state of para NH$_3$, which is the ground state of this species, has an internal energy of 16.245 cm$^{-1}$.\textsuperscript{15} The total energy for para NH$_3$ is consequently set 16.245 cm$^{-1}$ higher than the total energy for ortho NH$_3$.

For the reduced masses of NH$_3$–Ar and NH$_3$–He we have used the values 11.9396 and 3.2408 amu, respectively. The values for the rotational constants of NH$_3$ are taken from Ref. 15: $B = 9.9402$ cm$^{-1}$ and $C = 6.3044$ cm$^{-1}$.

Table I also lists the maximum values of $j$ and $k$ in the rotational basis set, the total number of levels and the number of accessible levels included in the calculations, the lowest energy ($E_{\text{omitted}}$) of the states omitted from the rotational basis, and the maximum overall rotational quantum number of the complex $J_{\text{max}}$ and the IBM RS/6000–370 CPU times for HIBRIDON versions 3.8 and 4.0, in the calculations for the NH$_3$–Ar and NH$_3$–He collision complexes. Given in parentheses in the column "open" is the number of levels with $j > J_{\text{max}}$ and an energy less than $E_{\text{tot}}$, that is omitted from the rotational basis set.

\begin{table}[h]
\centering
\begin{tabular}{|c|c|c|c|c|c|c|}
\hline
Atom & $E_{\text{tot}}$ & $(j_{\text{max}}, k_{\text{max}})$ & Total & Open & $E_{\text{omitted}}$ & CPU \\
\hline
\text{ortho NH$_3$} & & & & & & \\
\hline
Ar & 485.0 & (9, 9) & 34 & 19(0) & 798.9 & 219 150 24 \\
Ar & 1274.0 & (12,12) & 57 & 49(0) & 1285.6 & 484 250 160 \\
He & 436.0 & (9, 9) & 34 & 19(0) & 798.9 & 219 160 2 \\
He & 1130.0 & (11, 9) & 48 & 43(2) & 1027.1 & 372 100 12 \\
\hline
\text{para NH$_3$} & & & & & & \\
\hline
Ar & 501.2 & (9, 9) & 66 & 40(0) & 729.8 & 441 150 286 \\
Ar & 1290.2 & (12,11) & 112 & 96(0) & 1369.2 & 972 250 1035 \\
He & 452.2 & (9, 9) & 66 & 34(0) & 729.8 & 441 160 18 \\
He & 1146.2 & (11,11) & 96 & 86(2) & 1110.7 & 772 100 180 \\
\hline
\end{tabular}
\caption{The total collision energies $E_{\text{tot}}$, the maximum symmetric top quantum numbers $(j_{\text{max}}, k_{\text{max}})$ of the levels in the rotational basis set, the total number of (open and closed) levels and the number of accessible (open) levels included in the rotational basis, the lowest energy ($E_{\text{omitted}}$) of the states omitted from this basis, the maximum number of channels $n_{\text{max}}$, the maximum overall rotational quantum number of the complex $J_{\text{max}}$ and the IBM RS/6000–370 CPU times for HIBRIDON versions 3.8 and 4.0, in the calculations for the NH$_3$–Ar and NH$_3$–He collision complexes. Given in parentheses in the column "open" is the number of levels with $j > J_{\text{max}}$ and an energy less than $E_{\text{tot}}$, that is omitted from the rotational basis set.}
\end{table}
1308.5 cm\(^{-1}\) for para NH\(_3\) \((j=11, k=1)\), so that we do in-
clude some states with an energy larger than \(E_{\text{max}}\).

For He–NH\(_3\) at 1130 cm\(^{-1}\), it proved to be sufficient to
retain all levels in the rotational basis set up to \(j=11\); going
from \(j_{\text{max}}=10\) to \(j_{\text{max}}=11\) changed the cross sections by less
than 1%. For Ar–NH\(_3\) at 1274 cm\(^{-1}\), we had to include all
levels up to \(j=12\), since going from \(j_{\text{max}}=11\) to \(j_{\text{max}}=12\)
still introduced changes in the cross sections of about 10%.
We could not perform calculations at \(j_{\text{max}}=13\) since the
maximum value of \(j\) in the program is 12. However, we esti-
mate the difference between the results obtained with
\(j_{\text{max}}=12\) and \(j_{\text{max}}=13\) to be smaller than 5%. In the lower
energy case, both for Ar–NH\(_3\) and for He–NH\(_3\), the value of
\(j_{\text{max}}\) was 9. Retaining all levels up to \(j=11\) changed the cross
sections by less than 3%.

In all calculations \(J\) took the values 0, 1, 2, \ldots, \(J_{\text{max}}\),
with \(J_{\text{max}}\) given in Table I, and taken large enough to obtain
convergence. It is remarkable to see that in the case of
He–NH\(_3\) the value of \(J_{\text{max}}\) at which convergence is reached,
is higher for the low energy. The opposite is true for Ar–
NH\(_3\). The quantum number \(J\) can be regarded as the quan-
tum mechanical analog of the classical impact parameter.\(^7\)
If the potential would still be nonzero at infinity, all impact
parameters would contribute, and the value of \(J_{\text{max}}\) would be
determined by the value of \(J\) at which the overall rotational
energy starts to take up all the available energy. So one
would expect \(J_{\text{max}}\) to increase with increasing energy. How-
ever, if the range of the potential is short, only a limited
amount of impact parameters will contribute. When the total
energy of the collision increases, the distance at which the
potential becomes negligible compared to the total energy
gets smaller, and the number of significant impact parameters
decreases. Apparently, the potential of Ar–NH\(_3\) has
such a long range that the second effect does not dominate,
whereas it does for He–NH\(_3\).

The error in the cross sections due to the finite step size
in the propagator was about 1% in all calculations. The step
size for the He calculations at both energies was 10 times
larger than that of the Ar calculation at the low energy. At
the high energy for Ar, the latter step size was multiplied by
2 to keep the calculations feasible. Although the step size
should decrease with increasing energy, the convergence was
still within a few percent. We have used the possibility of the
program to include only open levels in the rotational basis
set when the overall rotation of the complexes starts to take
up a large part of the available energy. The values of \(J\) at
which we start to reduce the size of the rotational basis set is
75, 35, 40, and 20 for the successive energies and complexes
that are listed in Table I, for both NH\(_3\) species. The effect of
this reduction on the cross sections is found to be about 1%
in all cases. The fact that the values of \(J\) at which the reduc-
tion starts are smaller for the higher energies is caused not
only by the smaller difference between \(E_{\text{tot}}\) and \(E_{\text{max}}\), but
also by the decrease of the level density with increasing en-
ergy.

It is not possible to give precise timings for the calcula-
tions at the higher energies, since all of them had to be re-
started in the course of their runs. Moreover, the calculations
were done with different versions of HIBRIDON and on dif-
f erent machines. We therefore give rather crude timings in
Table I. The times given are for an IBM RS/6000 model 370
workstation and for the ESSL version of HIBRIDON, version
3.8 or 4.0.

IV. RESULTS AND DISCUSSION

As explained in Sec. III, the calculations for NH\(_3\)–Ar
were performed at a relative kinetic energy of 1274 and 485
cm\(^{-1}\). The relative kinetic energies in the NH\(_3\)–He calcula-
tions were set to 1130 to 436 cm\(^{-1}\). The high energies are
equal to the relative kinetic energy attained in the counter-
propagating pulsed molecular beam experiments\(^7\) to which we
compare our results. The lower energies were chosen to
 correspond to the energies attained in the crossed molecular
beam experiments to which we previously compared com-
puted ICSs for NH\(_3\)–Ar\(^{16}\) and NH\(_3\)–He.\(^{5,9}\) The experiment
measures the differences in population of a specific rotation-
 inversion state \(|i\rangle = \gamma_k^j\) before and after the collision. The
signal is proportional to

\[
\Delta n(i) = \sum_{j \neq k} [n(j) \sigma(j \rightarrow i) - n(i) \sigma(i \rightarrow j)],
\]

where \(n(i)\) stands for the initial population of state \(i\),
\(\Delta n(i)\) is a measure for the collision induced change in that
population, and \(\sigma(i \rightarrow j)\) is a state-to-state cross section.
The population changes \(\Delta n(i)\) may become negative. We will
nevertheless refer to them as "cross sections" (corrected for
the imperfect initial state preparation) and we will use the
abbreviations ICS and DCS for these population changes as
well. In the experiments of Meyer the ortho NH\(_3\) initial state
consists of 81% \(0^+_1\) and 19% \(1^+_0\) and the para NH\(_3\) initial
state of 57% \(1^+_1\) and 43% \(1^+_1\), both for NH\(_3\)–Ar and for
NH\(_3\)–He.

In Ref. 3 we found that it is important to take the initial
state preparation into account if one wants to make a com-
parison with experiment, especially for NH\(_3\)–He. In order to
make the correction for the imperfect initial state prepara-
tion, the state-to-state ICSs and DCSs obtained from the cal-
culations must be put into Eq. (2). The contributions from
the \(1^+_1\) state of the ortho species and the \(1^+_1\) state of the para
species are taken from the same computations (with the same
total energy) as the cross sections from \(0^+_0\) and \(1^+_1\). This is
justified, because the cross sections are only weakly depend-
don energy.

The results of the calculations of the DCSs are displayed
in Figs. 1–5. The theoretical cross sections are given in
Å\(^2\)sr. The angular resolution in the experiment is \(\Delta(\cos \Theta)\)
= 0.2 and 0.3 for NH\(_3\)–Ar and NH\(_3\)–He, respectively.
The diffraction oscillations at small angles can therefore not be
resolved experimentally. The error in the DCSs is estimated
to be ±15%. To avoid confusion we remark that if the DCS
for a certain final state is dashed in the left panel, it does not
necessarily have to be dashed in the right panel; the dashes
are only used to get a better distinction between the different
DCSs. Since the experiment yields only relative values for
the DCSs it was necessary to scale them to enable a com-

FIG. 1. Theoretical and experimental DCSs for ortho NH₃−Ar at a relative kinetic energy of 1274 cm⁻¹. The theoretical DCSs are calculated from the scaled ab initio potential (left) and from the empirical potential (middle). The experimental DCSs (right) are taken from Ref. 7.

Comparison with the theoretical DCSs. Since it was not possible to find a general recipe for this scaling, we have scaled each case separately, in a way that is described in Secs. IV A and IV B.

In Tables II–VI we have listed both the pure state-to-state ICSs and those that are corrected for the imperfect initial state preparation, together with the experimental ICSs. The theoretical ICSs are given in Å². As for the DCSs, only relative values for the ICSs can be derived. To facilitate comparison, the sum of the experimental ICSs over all states is set equal to the sum of the corrected ICSs from the present calculations. The sum contains only ICSs for excitations to levels j_k that are experimentally observed. The experimental error in the individual ICSs is 10% to 25%. In order to get a single parameter as a measure for the overall agreement between the calculated and experimental ICSs, we have defined an error Q in Ref. 2 in the following way:

$$Q = \left[ \frac{\sum_i (\Delta n(i)_{\text{calc}} - \Delta n(i)_{\text{exp}})^2}{\sum_i \Delta n(i)_{\text{exp}}^2} \right]^{1/2} \times 100\%,$$

where Δn(i)_{calc} are the corrected calculated ICSs, Δn(i)_{exp} are the experimental ICSs, and the summation runs over all measured final states j_k. The values of Q are included in the tables.

A. Differential cross sections for NH₃−Ar

The theoretical and experimental DCSs for NH₃−Ar are shown in Figs. 1–3. In the figures we have indicated the NH₃ species, the relative kinetic energy, and the origin of the DCSs. The experimental DCSs in Fig. 1 are scaled by setting the value of the maximum in the experimental DCS to 3ₚ equal to the value for that maximum calculated from the empirical potential. In Fig. 2 the experimental DCSs for para NH₃−Ar are scaled by setting the experimental DCS to 4₂ equal to the corresponding theoretical value at cos Θ = −1.

Figure 1 enables us to compare the DCSs from the scaled ab initio potential as well as those from the empirical potential with the experimental ones. The maximum in the experimental DCS to 3ₚ is reproduced by both potentials. The corresponding angles are Θ = 80°, 94°, and 63° for the experiment, the scaled ab initio potential, and the empirical potential, respectively.

A somewhat more approximate scattering method is the coupled states (CS) approach. In this approach more symmetry is present than in the CC method, due to the neglect of off-diagonal Coriolis interaction in CS. Thus, CS has certain selection rules that are absent in CC. In the experiment the CS allowed DCSs (with $\epsilon = (-1)^k$) to the states 3ₚ and 4ₚ show an angular dependence that is similar to the angular dependence of their CS forbidden counterparts (with $\epsilon = (-1)^{k+1}$) to 3ₛ and 4ₛ. This is also the case for the corresponding DCSs obtained from the empirical potential. For the scaled ab initio potential, however, the angular dependence of the CS allowed DCSs shows strong deviations from those that are CS forbidden, even at intermediate angles. A characteristic of the experimental CS forbidden DCSs is that they show a stronger decrease towards cos Θ = −1 than the CS allowed DCSs. This characteristic is exhibited by both potentials (for the 4ₚ states of the empirical potential, see also Fig. 3). As explained by Meyer, this is related to the fact that the CS approximation becomes better for small impact parameters. As in the experiment, maxima occur in the DCSs to 3ₛ and 7ₛ for the empirical potential, whereas the scaled ab initio potential only gives a maximum.
FIG. 2. Theoretical and experimental DCSs for para NH$_3$–Ar at a relative kinetic energy of 1274 cm$^{-1}$. The theoretical DCSs (left) are calculated from the empirical potential. The experimental DCSs (right) are taken from Ref. 7.

The empirical potential also gives a maximum in the DCS to 4$^+$, contrary to the experiment and the scaled ab initio potential.

A remarkable difference between the theory and the experiment is that nearly all computed DCSs continue to decrease exponentially in the backward direction, whereas the experimental DCSs are much more constant in that region. This difference might be caused by the method that is used to derive the angular dependence from the experiment. This dependence is not measured directly, but is obtained via a deconvolution procedure that introduces uncertainties in the experimental dependence.

In view of the findings above and the cost of the calculations we have chosen to restrict our attention to the empirical potential. In the para calculation at 1274 cm$^{-1}$, and in the calculations at a relative kinetic energy of 485 cm$^{-1}$ only the empirical potential was used.

The DCSs for para NH$_3$–Ar are displayed in Fig. 2. The largest discrepancy between the theoretical and the experimental DCSs is in the relative magnitude of the DCS to 3$^-$, contrary to the experiment and the scaled ab initio potential. A remarkable difference between the theory and the experiment is that nearly all computed DCSs continue to decrease exponentially in the backward direction, whereas the experimental DCSs are much more constant in that region. This difference might be caused by the method that is used to derive the angular dependence from the experiment. This dependence is not measured directly, but is obtained via a deconvolution procedure that introduces uncertainties in the experimental dependence.

In view of the findings above and the cost of the calculations we have chosen to restrict our attention to the empirical potential. In the para calculation at 1274 cm$^{-1}$, and in the calculations at a relative kinetic energy of 485 cm$^{-1}$ only the empirical potential was used.

The DCSs for para NH$_3$–Ar are displayed in Fig. 2. The largest discrepancy between the theoretical and the experimental DCSs is in the relative magnitude of the DCS to 3$^-$. However, many of the features of the experimental DCSs are exhibited by the theoretical ones. The experimental DCSs to 2$^+$, 3$^+$, and 3$^-$ have maxima in the forward direction, in the backward direction they are nearly constant. The corresponding theoretical DCSs have the same characteristics, except that the computed DCS to 3$^+$ continues to decrease at large angles. Both in the theory and in the experiment we can distinguish a shoulder in the DCS to 3$^-$. The maxima in the DCSs to 4$^+$ and 5$^+$ are also reproduced by the theory, even though the angles at which they occur are somewhat larger. In the backward direction the DCS to 4$^+$ continues to decrease exponentially, the DCS to 5$^+$ is practically constant, in the computations as well as in the experiment.

Figure 3 demonstrates that both the angular dependence and the relative magnitudes of the DCSs depend strongly on the energy. The variation in the DCSs as a function of $\Theta$ is much smaller if the energy is lower, especially at intermediate angles. The maxima and the shoulders in the DCSs have nearly all disappeared at 485 cm$^{-1}$.

B. Differential cross sections for NH$_3$–He

The results of the calculations of the DCSs for NH$_3$–He are displayed in Figs. 4 and 5. The NH$_3$ nuclear spin species and the relative kinetic energy are also indicated in the figures. The experimental DCSs for ortho NH$_3$–He in Fig. 4 are scaled by setting the value of the experimental DCS to 3$^-$ at cos $\theta = 1$ equal to its theoretical value, for para NH$_3$–He we set the experimental DCS to 5$^+$ at cos $\theta = -1$ equal to the corresponding theoretical value.

Figure 4 shows the results at a relative kinetic energy of 1130 cm$^{-1}$, together with the experimentally determined DCSs from Ref. 8. If we look at the DCSs for ortho NH$_3$–He we see that there is a good qualitative agreement between theory and experiment. The DCS to 3$^-$ is dominant at small angles, whereas the DCS to 4$^+$ is dominant at large angles. The 7$^+$ DCS for backward scattering is larger than those for 2$^+$ and 3$^-$.$^*$ Both in theory and in experiment the DCSs to 4$^+$ and 5$^+$ have a maximum at intermediate angles. The threshold behavior of the DCSs to 7$^+$ and 7$^+$ is also reproduced fairly well. In general, the relative magnitudes of the DCSs are in reasonable agreement. The largest deviation occurs for the cross section to 4$^+$.$^*$ The shoulders in the DCSs to 3$^+$ and 3$^-$ are, however, not reproduced by the...
calculations, whereas the shoulder in the DCS to $6^-$ is greatly exaggerated.

We have also compared the angular dependence of the CS allowed and the CS forbidden DCSs. As in the experiment we found that they were very similar over the whole range, and that the CS forbidden DCSs were considerably smaller, to the same extent as in the experiment.

In his discussion of the experimental para NH$_3$-He DCSs Meyer distinguishes three categories of DCSs. First, DCSs for transitions involving small energy transfer ($2^*, 3^*$) that are characterized by a maximum at $\cos \theta = 1$, and an exponential decrease towards larger scattering angles. Second, DCSs to states representing intermediate energy transfer ($3^*, 4^*, 5^*$) that show a rotational rainbow behavior, i.e., they have a maximum which shifts to larger angles with increasing energy transfer. And finally, DCSs to states corresponding to large energy transfer that show a threshold behavior and a maximum excitation probability for backward scattering.

Looking at the results of our calculations, we see that this division into three categories can be made as well for the theoretical DCSs. However, the relative magnitudes of the DCSs are quite different, especially for the DCSs to $2^*$, $6^*$, $4^*$, and $7^*$. Also the shoulders in the cross sections to $2^*$ and $3^*$ do not appear in the theoretical DCSs. We must therefore conclude that the overall agreement between theory and experiment for para NH$_3$-He is less good than for ortho NH$_3$-He.

The DCSs computed at an energy of 436 cm$^{-1}$ are shown in Fig. 5. As for NH$_3$-Ar the relative magnitudes of the DCSs change, and the variation in the DCSs as a function of $\Theta$ is smaller. Obviously, fewer states are accessible at the lower energy. For ortho NH$_3$-He the DCSs to $4^*$ and $5^*$ have lost their maximum and the inflection points in the DCSs to $3^*$ and $3^*$ have disappeared. The maxima in the para DCSs have disappeared as well, and the DCSs to $2^*$ and $3^*$ are practically independent of the scattering angle.

Making an overall comparison of the DCSs of NH$_3$-Ar and NH$_3$-He, we see that the NH$_3$-Ar DCSs are much more peaked in the forward direction than those of NH$_3$-He. Many of the latter DCSs even show a threshold behavior at small scattering angles, and a maximum in the backward direction. These differences in the DCSs are a direct reflection of the differences in the potentials of the two complexes. For NH$_3$-He the long range attraction is much more important than for NH$_3$-He. As a consequence, the influence of the NH$_3$-Ar potential is much stronger at large impact parameters than that of the NH$_3$-He potential. Hence, the NH$_3$-He DCSs are dominated by short range collisions.

C. Integral cross sections for NH$_3$-Ar

In Tables II and III we have listed the ICSs for ortho NH$_3$-Ar for the scaled ab initio and the empirical potential, respectively. The ICSs for para NH$_3$-Ar, for which we only used the empirical potential, are given in Table IV. In Refs. 1 and 2 we also calculated ICSs both for ortho and para NH$_3$-Ar, at a relative kinetic energy of 280 and 485 cm$^{-1}$, and we compared them with experiment. The average error $Q$ was found to be 40% for the scaled ab initio potential and 26% for the empirical potential. Here, the errors for ortho NH$_3$-Ar are practically equal for the two potentials, the scaled ab initio potential performs even slightly better. An explanation for this finding may be that at an energy as high
as 1274 cm\(^{-1}\) the collisions mainly probe the repulsive part of the potential. At lower collision energies the well region of the potential becomes more important. The empirical potential is likely to give a better description of the well region than the scaled \textit{ab initio} potential, since it is determined from the bound state spectrum. The well of the scaled \textit{ab initio} potential, on the other hand, is determined by a delicate balance between the repulsive and attractive contributions to the potential. A small deviation in either of the two contributions already has a large effect on the shape of the well.

If we look at the individual cross sections we see that CS allowed ICSs are clearly larger than the CS forbidden ICSs, in the theory as well as in the experiment, which indicates that the off-diagonal Coriolis interaction is small. The overall agreement with experiment is fairly good for both potentials. The ratio of the corrected calculated cross sections and the experimental ones varies at most by a factor of 2. For ICSs to states with \(j > 3\) the experimental value is always in between the two theoretical ones. The negative value for the corrected ICS to \(1_{0}^{+}\) indicates that the theory predicts a depletion of this state. Such a depletion has indeed been found in the experiment,\(^9\) although the actual value of this negative cross section has not been experimentally determined.

The results for para NH\(_3\)-Ar are given in Table IV. Here the error \(Q\) was determined by averaging the calculated corrected cross sections for the two parity states of each rotational level \(j_{k}\). We note that the experimental values given in Table IV are also the average values of the ICSs to the two inversion states, not their sum.

FIG. 4. Theoretical and experimental DCSs for ortho NH\(_3\)-He (top) and para NH\(_3\)-He (bottom) at a relative kinetic energy of 1130 cm\(^{-1}\). The experimental DCSs (right) are taken from Ref. 8.
D. Integral cross sections for NH$_3$-He

The ICSs for ortho and para NH$_3$-He are listed in Tables V and VI. As indicated in the tables, the values of the ratio of the computed and the experimental ICSs varies by a factor of 2 at most, and that the overall agreement between theory and experiment is reasonable.

For para NH$_3$-Ar we find an error of 40%, which is substantially larger than the error for ortho NH$_3$-Ar. However, if we look at the individual ICSs, again we see that the error Q are 23% and 45%, respectively. It is remarkable to see that both for NH$_3$-Ar and for NH$_3$-He the error for the ortho species is about 25%, whereas the error for the para species is about 43%. It is difficult to find an explanation for this difference. It is unlikely that the inaccuracies in the potential will have different consequences for the ortho and para species, and our CC calculations are equally well converged for the two species. In our previous study of NH$_3$-Ar, in which we compared the theoretical ICSs with the experimental results of ter Meulen, the errors for the

error Q are 23% and 45%, respectively. It is remarkable to see that both for NH$_3$-Ar and for NH$_3$-He the error for the ortho species is about 25%, whereas the error for the para species is about 43%. It is difficult to find an explanation for this difference. It is unlikely that the inaccuracies in the potential will have different consequences for the ortho and para species, and our CC calculations are equally well converged for the two species. In our previous study of NH$_3$-Ar, in which we compared the theoretical ICSs with the experimental results of ter Meulen, the errors for the
orth and para species were about the same for the empirical
potential. It may be, therefore, that the ICSSs from the present
experiment are slightly less accurate for the para than for the
ortho species.

<table>
<thead>
<tr>
<th>$j_f$</th>
<th>$\Delta n(j_f)$</th>
<th>$\Delta n(j_f)$</th>
</tr>
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<td>3.69 2.85</td>
<td>6$^+$ 0.08 0.23</td>
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<tr>
<td>2$^-$</td>
<td>2.22 3.06</td>
<td>6$^+$ 0.33 0.19</td>
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<td>1.96 1.38</td>
<td>4$^+$ 0.94 1.92</td>
</tr>
<tr>
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<td>6$^+$ 0.64 0.31</td>
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<td>0.14 0.35</td>
<td>7$^+$ 0.04 0.07</td>
</tr>
<tr>
<td>6$^+$</td>
<td>0.50 0.30</td>
<td>7$^+$ 0.09 0.06</td>
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<tr>
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<td>5$^+$</td>
<td>0.27 0.29</td>
<td>7$^+$ 0.23 0.16</td>
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</table>

$Q=40\%$

*The experimental values from Ref. 9 are multiplied by 0.73.

If we look at the individual cross sections, we see that the
overall agreement is reasonable both for ortho and para
NH$_3$–He. As in the case of NH$_3$–Ar, the majority of the
ICSSs differ from the experimental values by less than a factor
of 2. There are, however, a few ICSSs that show a large dis-
crepancy between theory and experiment. The ICSSs to 5$^+$, 6$^+$,
6$^+$, 4$^+$, and 7$^+$ are all practically zero in the theory,
whereas they differ significantly from zero in the experiment.
A likely cause for the discrepancies is that the quality of the
potential is insufficient to accurately predict these small
cross sections. We have already seen that our potential is
incapable of predicting the rather subtle population changes
of states that have both in and out scattering.

We already noted that in scattering calculations for
NH$_3$–He it is very important to take the initial state prepa-
ratio into account if one wants to make a comparison with
experiment. Up to now we only included the first excited
state in our correction. In this paper, we have investigated
the effect of taking into account the lowest six states in the
initial state preparation both for ortho and para NH$_3$–He,
instead of the lowest two. The initial population of these
states was taken from Table 2 in Ref. 9. Even though we
introduce an error by taking all six initial states from the
same computation, i.e., we give them different relative ki-
netic energies (see also the beginning of this section) we can

<table>
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<tr>
<th>$j_f$</th>
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<th>$\Delta n(j_f)$</th>
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</tr>
<tr>
<td>4$^+$</td>
<td>0.01</td>
<td>0.44</td>
<td>58</td>
</tr>
</tbody>
</table>

$Q=45\%$

*The experimental values from Ref. 8 are multiplied by 0.93.

get an idea of how important the effect is. The effect is most important on the cross sections to the four states that are now included, in the order of about 10%. The other states are hardly affected. We conclude, therefore, that for the given initial state populations it is not necessary to include more than one excited state.

V. CONCLUSIONS

We have calculated state-to-state DCSs and ICSs for rotational excitation and inversion of NH\textsubscript{3} by collisions with Ar and He using the CC method. In the calculations for ortho NH\textsubscript{3}-Ar we have used an empirical and a scaled \textit{ab initio} potential, for para NH\textsubscript{3}-Ar we have only used the empirical potential. The NH\textsubscript{3}-He calculations were based on an \textit{ab initio} potential. We compare our results to experimentally determined cross sections.

For ortho NH\textsubscript{3}-Ar we find that the DCSs obtained from the empirical potential are in closer agreement with the experimental DCSs than those obtained from the scaled \textit{ab initio} potential. For the ICSs, however, the agreement with experiment is about the same for the two potentials. Both for ortho and para NH\textsubscript{3}-Ar the empirical potential reproduces the features of the experimentally determined DCSs fairly well, and for both species the ICSs differ by less than a factor of 2 from the experimental ones.

The computed DCSs for ortho NH\textsubscript{3}-He are in accordance with the experiment, for the para species some of the relative magnitudes of the calculated DCSs deviate from those in the experiment. Both for ortho and para NH\textsubscript{3}-He most of the ICSs differ by less then a factor of 2 from the corresponding experimental ICSs. However, for both species there are a few small ICSs that deviate significantly from their experimental value. It is likely that the discrepancies for NH\textsubscript{3}-He are caused by the fact that the potential is not sufficiently accurate.

Both for NH\textsubscript{3}-Ar and NH\textsubscript{3}-He we find that the angular dependence and the relative magnitudes of the DCSs depend strongly on the energy. A decrease in energy reduces the variation of the DCSs with the scattering angle, especially at intermediate angles. Scattering calculations should therefore be performed at the same energies as the experiment to enable a meaningful comparison.

Finally, we have assessed that it is sufficient to take only the first excited state, which is the dominant contamination, into account when making the correction for the initial state preparation in the experiments to which we compare our results.

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